

## **Electronic supplementary information**

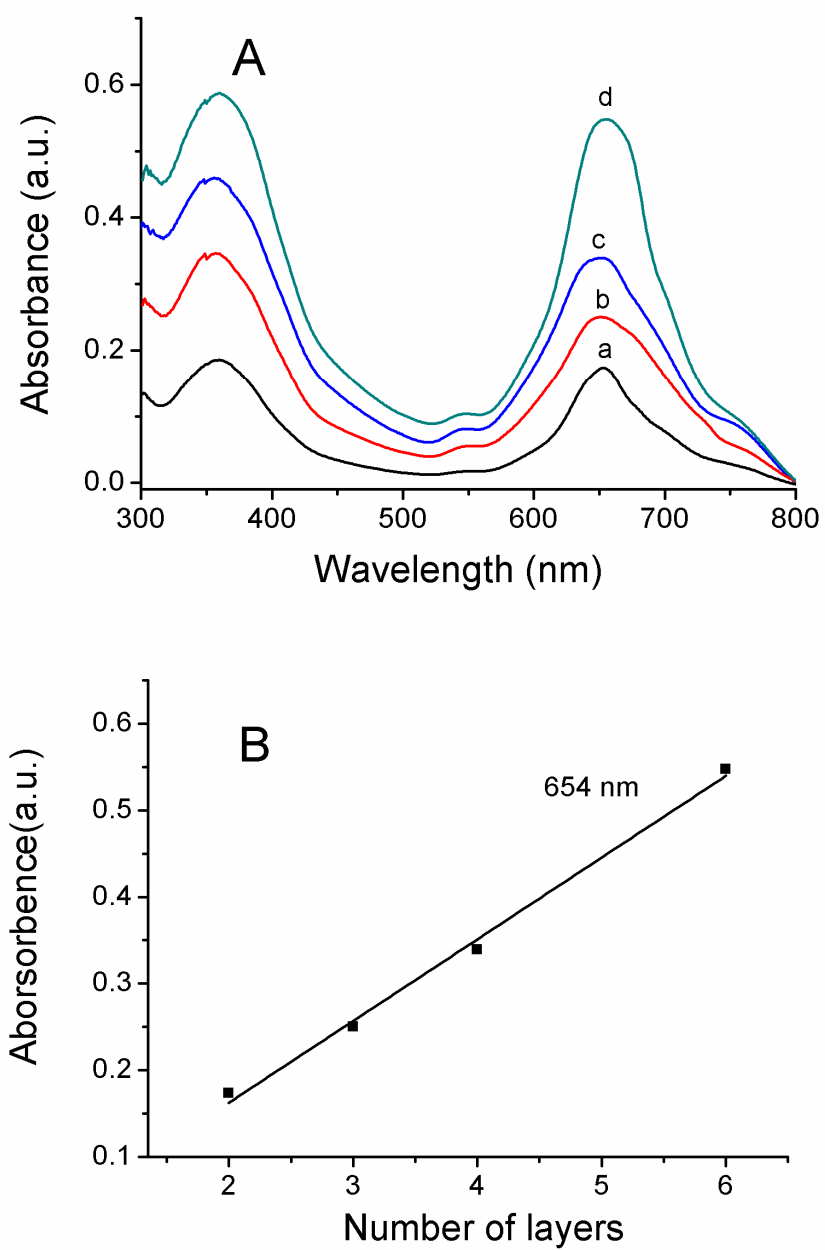
**Facile Approaches to built Ordered Amphiphilic Tris(phthalocyaninato)**

**Europium Triple-Decker Complex Thin Films and their comparative**

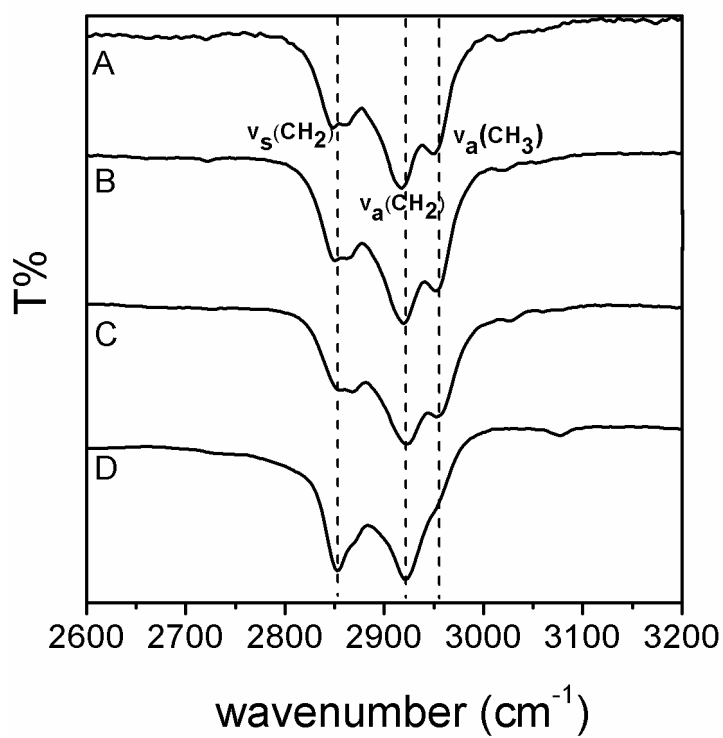
**performances as the ozone sensor**

Yanli Chen, Marcel Bouvet\*, Thibaut Sizum, Yingning Gao, Cedric Plassard, Eric

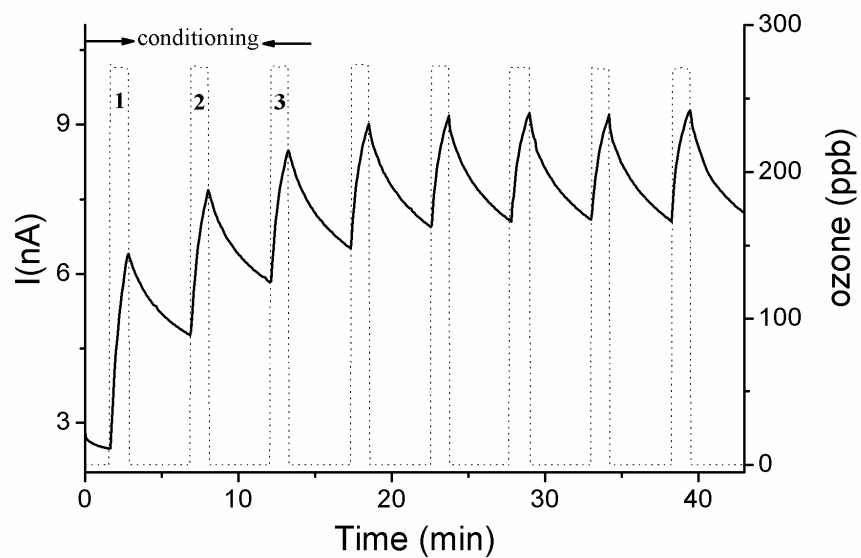
Lesniewska and Jianzhuang Jiang\*



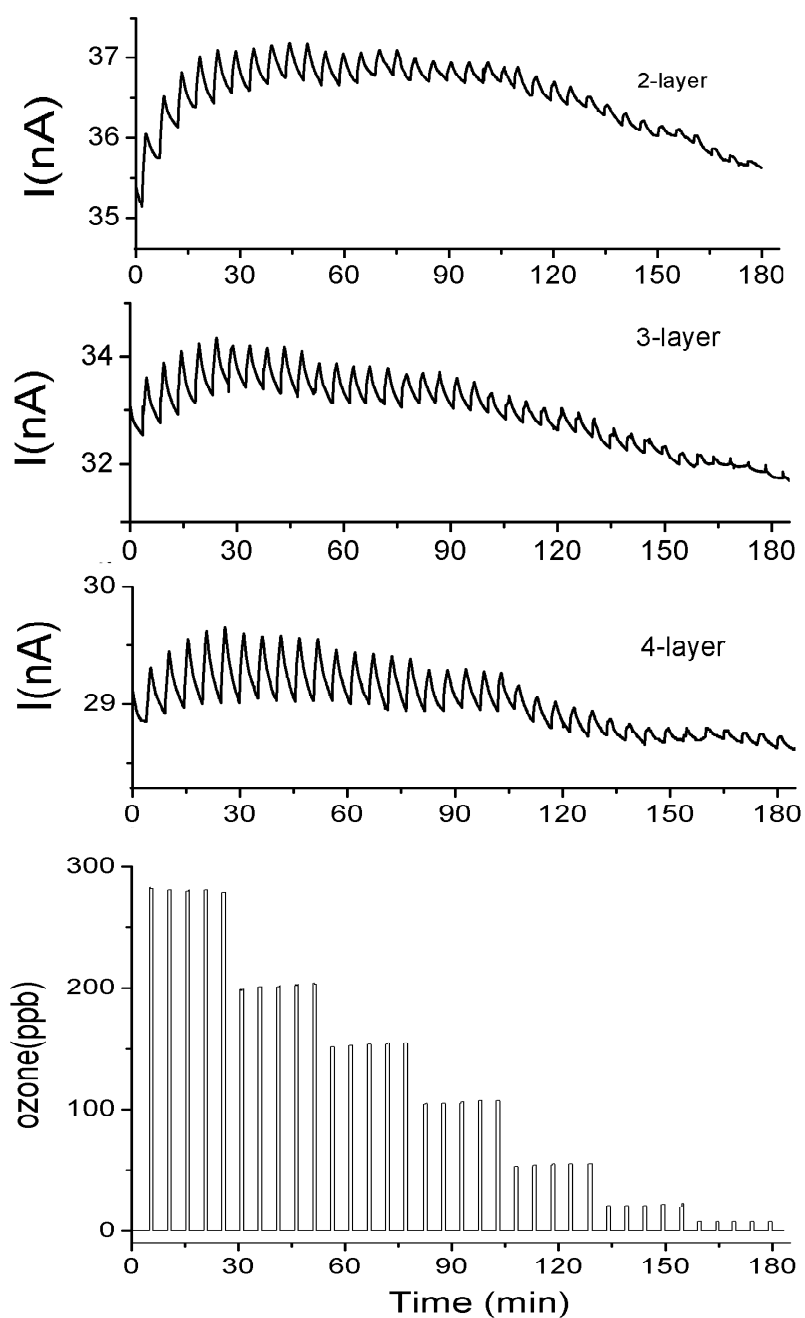
**Fig. S1** (A) UV-vis spectra of QLS Eu<sub>2</sub>[Pc(15C5)<sub>4</sub>]<sub>2</sub>[Pc(OC<sub>10</sub>H<sub>21</sub>)<sub>8</sub>] films (a-d), (a: 2 layers; b: 3 layers; c: 4 layers; d: 6 layers). (B) The relationship between the absorption intensity at 654 nm and the number of layers.



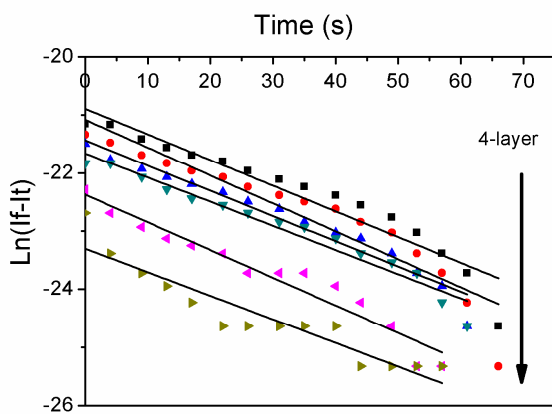
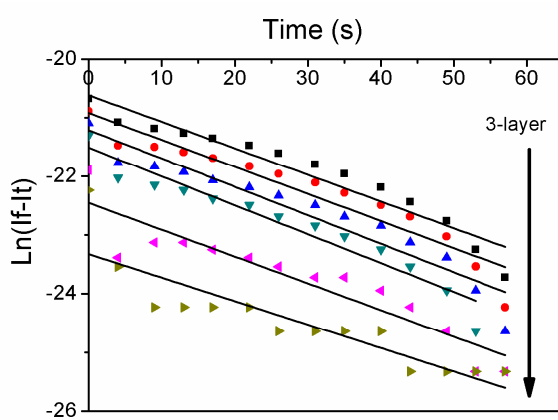
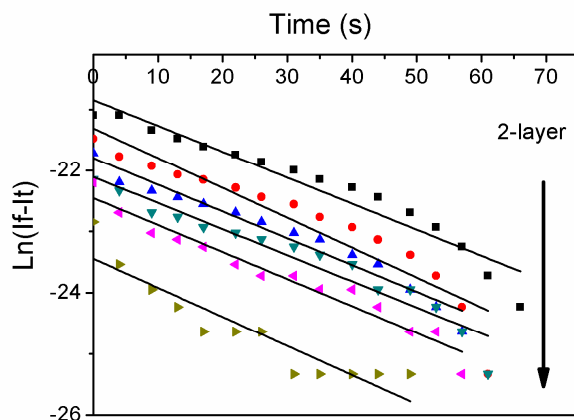
**Fig. S2** FTIR in ATR mode spectra of (A) SA films; (B) QLS films; (C) cast films and (D) bulk  $\text{Eu}_2[\text{Pc}(\text{15C5})_4]_2[\text{Pc}(\text{OC}_{10}\text{H}_{21})_8]$  adsorbed at ITO/glass substrates in the region of 2600-3200  $\text{cm}^{-1}$  with 2  $\text{cm}^{-1}$  resolution.



**Fig. S3.** Alternation of exposure and static recovery periods showing the conditioning stage(cycles 1–3) which is a prerequisite to obtain reproducible measurements of cast  $\text{Eu}_2[\text{Pc}(\text{15C5})_4]_2[\text{Pc}(\text{OC}_{10}\text{H}_{21})_8]$  films.(similar situation took place for another two types of films and have been omitted for clarity). Dotted line: concentration of ozone.



**Fig. S4** The time-dependent current plots for 2-4 layers of QLS  $\text{Eu}_2[\text{Pc}(15\text{C}5)_4]_2[\text{Pc}(\text{OC}_{10}\text{H}_{21})_8]$  films exposed to  $\text{O}_3$  at varied concentration in the range of 0-300 ppb( exposure:1min, recovery: 4 min), while the bottom rectangular pulses represent the  $\text{O}_3$  concentration as a function of time.



**Fig. S5** The kinetics plots of 2-4 layers of QLS  $\text{Eu}_2[\text{Pc}(15\text{C}5)_4]_2[\text{Pc}(\text{OC}_{10}\text{H}_{21})_8]$  films:  $\ln(I_f - I_t)$  versus time is predominantly linear for different concentration of  $\text{O}_3$ , suggesting first-order kinetics. Arrows indicate the direction of change with decreasing concentration of ozone.

**Table S1.** Characteristics of electrical conductivity and gas sensing behavior from three types of  $\text{Eu}_2[\text{Pc}(\text{15C5})_4]_2[\text{Pc}(\text{OC}_{10}\text{H}_{21})_8]$  films at room temperature.

Type of films	film-thickness (nm)	electrical conductivity ( $\text{S}\cdot\text{cm}^{-1}$ ) <sup>a</sup>	Sensor response (% ppb <sup>-1</sup> )	Average sensor response rate constant ( $\text{s}^{-1}$ )
SA films	4.6 <sup>b</sup>	~more than $10^{-4}$ <sup>c</sup>	0.0376	0.0498
QLS films	4.42~13.26 <sup>d</sup>	$\sim 10^{-5}$	0.0267 <sup>e</sup>	0.0574
Cast films	92	$\sim 10^{-6}$ <sup>f</sup>	0.0721	0.0486

<sup>a</sup> Calculated by Eq: (1) in experimental section

<sup>b</sup> Thickness was obtained by AFM measurement.

<sup>c</sup> Electrical conductivity was calculated with an estimated channel coverage less than 50% .

<sup>d</sup> Thickness was deduced from the value of layer spacing obtained by XRD and the number of layers in a 2-6 layers range.

<sup>e</sup> Calculated for 6-layer QLS films

<sup>f</sup> The thickness of electrode of 20nm is preferred since the thickness of cast films calculated is ca. 92nm.