

Original Research

Seasonal Difference and Risk Assessment of Organophosphate Esters in Source Water in South China

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Abstract

Organophosphate esters (OPEs) in the environment, especially in source water, pose a potential threat to both human health and aquatic organisms. This study investigated the concentration, seasonal difference, and risk of 14 widely used OPEs in river and reservoir source water in South China. The total concentration of OPEs was significantly higher during the dry season than during the wet season ($p < 0.05$; median: 144 ng/L vs. 89.4 ng/L). The most dominant OPEs among the 14 investigated in source water were tris(1-chloro-2-propyl) phosphate (TCIPP) during the dry season. No significant difference was found in the concentration and profile of OPEs between the river and reservoir source water investigated ($p > 0.05$). The health risk of OPEs in the source water to humans was negligible (non-carcinogenic risk < 1 ; carcinogenic risk $< 10^{-6}$) with the dominant contributor being tris (2-chloroethyl) phosphate (TCEP). The environmental ecological risk of OPEs was mostly low (risk quotient < 0.1) in the source water investigated. Overall, the source waters in South China were found to be relatively safe for use as drinking water sources in the context of OPE pollution.

Keywords: organophosphate flame retardant, source water, seasonal difference, ecological risk, health risk

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Introduction

Organophosphate esters (OPEs) are emerging persistent organic pollutants with various types of toxicity [1], and are classified into three categories: halogenated OPEs, alkyl OPEs, and aromatic OPEs. OPEs, also known as organophosphorus flame retardants, have been used as substitutes for brominated flame retardants for decades worldwide [2]. Compared to traditional flame retardants, OPEs can also be used as plasticizers and defoamers, making their use more widespread and their sales higher than traditional flame retardants [3]. In Europe, the demand for OPEs in 2006 was 93,000 tons, accounting for 20% of the annual consumption of flame retardants [4]. It is reported that the annual production volume of OPEs was estimated at 598,422 metric tons in China in 2020 [5]. Due to their characteristics, OPEs can be released into the environment throughout their life cycle, including production, use, transportation, recycling, and disposal [6]. OPEs have been found in various environmental matrices worldwide, such as air [7], water [8], and sediment [9].

OPEs were widely detected in surface water and the concentration varied greatly. The concentration of OPEs was at the ng/L- $\mu\text{g/L}$ level in surface water [10], which was much lower than the lethal concentration of 50% of OPEs (mg/L level) [11, 12]. However, tris(1-chloro-2-propyl) phosphate (TCIPP) and tris(2-chloroethyl) phosphate (TCEP) have potential risks of bioaccumulation [13]. The bioaccumulation of OPEs may magnify their toxicity to biota.

OPEs in source water have negative effects on both human health and the environment. OPEs have the potential to pose a threat to human health through multiple exposure pathways, including dietary intake, skin exposure, ingestion of dust, and inhalation [8, 14, 15]. Conventional drinking water treatment plants have limited capacity to remove OPEs, especially chlorinated OPEs [16, 17]. Also, OPEs were detected in tap water in many studies [18-20]. As source water is the source of drinking water, OPEs in source water may be ingested by humans and pose a threat to human health. In addition, OPEs have potential toxicity to organisms. TCEP and tri-n-butyl phosphate (TNBP) could reduce cell viability, increase cell apoptosis, and change the cell morphology of PC12, which indicated that TCEP and TNBP had possibilities of cytotoxicity and neurotoxicity [21, 22]. Triphenyl phosphate (TPHP) at $\mu\text{g/L}$ level could downregulate regulatory feedback genes in the zebrafish [23]. It is necessary to evaluate the health and environmental risks of OPEs.

This study investigated the occurrence and evaluated the health and ecological impact of 14 OPEs in source water in South China. Specifically, the objectives of this study were to (1) investigate the occurrence of OPEs in source water in South China; (2) assess the health risk of OPEs to humans; and (3) assess the ecological environmental risk of OPEs to aquatic biota.

Experimental

Chemicals

The standards of 14 OPEs were all purchased from AccuStandard Inc. (New Haven, CT, USA). Organic solvents including dichloromethane and acetonitrile were all of chromatography grade. Ultrapure water (Milli-Q, MA, USA) was used for all solution preparation and vessel cleaning in this study.

Study Area and Sampling

This study selected the southern region of China as the research site, encompassing three reservoir water sources (Hedi, Shenzhen, and Niuweiling), and two river water sources (Xi River and Dong River) (Fig. 1). A total of 34 samples were collected, including 4 from Shenzhen reservoir (SZ-1 to SZ-2), 6 from Hedi reservoir (HD-1 to HD-3), 4 from Niuweiling reservoir (NW-1 and NW-2), 6 from the north branch of Dong River (DR-1 to DR-3), and 14 from the lower reaches of Xi River (XR-1 to XR-7). The samples were collected in March and August 2023, which represented the dry and wet seasons, respectively. Water samples (50 cm depth) were collected using an organic glass water sampler, stored in brown glass bottles at 4°C, and preprocessed within 72 hours.

Chemical Analyses

The water samples were filtrated with 0.7 μm glass fiber filters (Whatman, GF/F). The filtrated samples (500 mL each) were spiked with 5 ng surrogate (TPP-d21, TNBP-d27, TCIPP-d18, and TPHP-d15) and processed by solid-phase extraction (SPE). The ENVI-18 cartridges (6 mL, 500 mg, Supelclean) were conditioned with sequential elution of dichloromethane (5 mL), acetonitrile (5 mL), and Milli-Q water (10 mL). The samples were passed through the cartridges at a flow rate of 1 mL/min. The cartridges were dried under high-purity nitrogen. The analytes were subsequently eluted with 8 mL of dichloromethane/acetonitrile (25:75, v/v). The extracts were evaporated by nitrogen at 40°C to nearly dry and diluted to 1.0 mL with acetonitrile. Extracts were filtered with 0.22 μm membrane prior to analysis by liquid chromatography.

The concentrations of OPEs in samples were analyzed by an ultra-high performance liquid chromatography tandem mass spectrometry system (Xevo TQ-S micro IVD system, Waters, Manchester, UK) according to the method previously described with slight modification [24]. Briefly, all analytes were separated on a Waters BEH C18 column (2.1 mm \times 50 mm, 1.7 μm particle size, Milford, MA). The flow rate was 0.4 mL/min, and the column temperature was set at 40°C. The source temperature was set at 150°C and the ionization mode was positive. The analytes were

Environmental Risk Assessment

The environmental risk of OPEs was evaluated using the risk quotient (RQ) that has been widely employed in aquatic environmental risk assessment [27, 28]. RQ was a ratio of the measured environmental concentration (MEC) and predicted no-effect concentrations (PNEC). If RQ is lower than 0.1, the risk is considered low. If RQ is higher than 0.1 and lower than 1, the risk is considered medium. If RQ is higher than 1, the risk is considered high [29]. RQ was calculated using Eq. (4) and Eq. (5).

$$RQ = \frac{MEC}{PNEC} \quad (4)$$

$$PNEC = \frac{EC_{50} \text{ or } LC_{50}}{f} \quad (5)$$

where MEC is the measured environmental concentration; PNEC is the predicted no-effect concentration, generally estimated as a quotient of the toxicological relevant concentration (EC_{50} or LC_{50}) and a safety factor (f) obtained from available literature [28].

Results and Discussion

Occurrence and Seasonal Difference of Organophosphate Esters in Source Water

All investigated OPEs were detected in source water in South China during the wet season while 10 of 14 investigated OPEs were detected during the dry season. During the wet season, the detection frequencies of triethyl phosphate (TEP), TCIPP, tricresyl phosphate (TMPP), and tributoxyethyl phosphate (TBOEP) were 100% while the detection frequencies of trimethyl phosphate (TMP), tripropyl phosphate (TPP), TNBP, TCEP, TPHP, 2-Ethylhexyl diphenyl phosphate (EDHPP), tris(1,3-dichloro-2-propyl) phosphate (TDCIPP), and tris(2-ethylhexyl) phosphate (TEHP) ranged from 82-94%. The detection frequencies of OPEs during the dry season followed the order: TNBP, TCEP, and TCIPP (100%) > TMP and TBOEP (88%) > TEP (82%) > tri-iso-butyl phosphate (TIBP) and TDCIPP (35%) > TPP (18%) > TPHP (12%), while cresyldiphenyl phosphate (CDP), EDHPP, TMPP, and TEHP were not detected during the dry season. The higher detection frequencies of OPEs during the wet season, compared to the dry season, might be attributed to the increased rainfall during the wet season, which leads to a greater influx of OPEs from soils, roads, and wet deposition into surface water [8, 30]. Besides, the increase in human activities during the wet season (summer), especially in transportation, could result in the use and production of a wider range of OPEs [30]. This could contribute to the higher detection frequency of OPEs during the

wet season. The detection frequencies of TPHP, CDP, EDHPP, TMPP, and TEHP were much higher during the wet season than during the dry season. This may be due to their high $\log K_{ow}$ values (4.70-9.49), which cause most of these substances to attach to solid particles. Rainfall can transport them to source water along with solid particles, especially during the wet season.

The median total concentration of OPEs was 89.4 ng/L (range 77.1-95.7 ng/L) during the wet season and 144 ng/L (range 17.8-276 ng/L) during the dry season (Fig. 2). The total concentration of OPEs was significantly lower during the wet season than during the dry season ($p < 0.05$). The median total concentration of alkyl-OPEs was 47.9 ng/L (range 37.2-58.7 ng/L) during the wet season and 32.9 ng/L (range 0.928-103 ng/L) during the dry season. However, there is no significant difference in the total concentrations of alkyl-OPEs between the wet and dry seasons ($p > 0.05$). The median total concentration of halogenated OPEs was 15.7 ng/L (range 8.91-18.8 ng/L) during the wet season and 95.5 ng/L (range 1.33-253 ng/L) during the dry season. The total concentrations of halogenated OPEs were significantly higher during the dry season than during the wet season ($p < 0.01$). The concentrations of alkyl-OPEs were higher than the concentrations of halogenated OPEs during the wet season ($p < 0.01$), while the concentrations of alkyl-OPEs were lower than the concentrations of halogenated OPEs during the wet season ($p < 0.01$). This may be because halogenated OPEs are more resistant to photodegradation than alkyl-OPEs [31].

The dominant OPEs of the 14 investigated OPEs in source water were TCIPP (median: 46.6 ng/L), TCEP (32.8 ng/L), and TEP (27.1 ng/L) during the dry season. The annual production volumes of TCIPP, TCEP, and TEP were reported as 44,681, 30,957, and 64,694 metric tons, respectively [5], which were relatively higher compared to the other 11 investigated OPEs. This may explain their dominance during the dry season. During the wet season, the median concentrations of OPEs in source water were relatively similar (2.61-10.1 ng/L). Factors such as high water flow, redissolution from sediment, and wet deposition affected the concentration of OPEs in source water during the wet season [8].

TCIPP, primarily used as a flame retardant and plasticizer [32], is frequently employed due to its cost-effectiveness and superior flame retardancy [33]. The poor removal of TCIPP observed in wastewater treatment plants, as reported in previous studies [34, 35], could be a contributing factor to its high concentration. The European Union banned the production of TCEP in 2011 due to its high toxicity [36], leading to its replacement by TCIPP. The significant levels of TCEP and TCIPP detected in this study underscore their extensive usage in China and the pressing need for innovative eco-friendly alternatives.

The maximum concentrations of OPEs detected in the source water of this study were found to be at the medium level when compared to the concentrations

Table 1. Maximum concentrations of investigated OPEs in surface water worldwide (ng/L)

Country	Site	TMP	TEP	TPP	TIBP	TNBP	TCEP	TPHP	TCIPP	TBOEP	TDCIPP	Reference
China	South China	10.6	96.2	17.8	17.0	17.1	89.9	106	196	11.1	7.76	This study
	Taihu Lake	ND	2.81	-	-	ND	27.83	377.1	10.29	ND	-	(Chen et al., 2018)
	Henan	-	-	-	-	-	49.3	-	66.4	-	-	(Wu et al., 2019)
	Hanshui River	3.76	7.27	-	-	-	16.4	0.84	-	-	-	(Huang et al., 2023)
	Yangtze River	19.4	7.27	-	-	-	11.2	0.88	-	-	-	(S. Zhang et al., 2021)
	Xiangjiang River	-	-	-	-	8.89	0.46	8.41	-	-	-	(Shi et al., 2020)
	Pearl River	-	33.4	ND	-	125	102	16.5	215	-	28.9	(Lin et al., 2022)
	Jiaozhou Bay	0.96	341.13	11.66	21.33	-	691.18	2.72	835.27	33.21	212.84	(Xing et al., 2023)
	Bohai Sea	384.2	1189.7	-	105.9	65.5	1721.3	-	93	353.4	4.6	(R. Wang et al., 2015)
	North China	-	-	-	218	-	268	-	921	47	-	(Truong et al., 2023)
Vietnam	Hanoi	-	-	-	-	-	126	272	3471	177	-	(Schmidt et al., 2020)
France	Rhône River	-	-	-	9.3	138.1	25	1.4	173.1	-	8.7	(Lee et al., 2018)
Korea	Shihwa Lake	-	3677	-	-	-	5963	96.2	5102	839	-	(Bollmann et al., 2012)
Germany	Elbe River	-	-	-	-	7.5	20	4	250	-	-	
	Rhine River	-	-	-	-	28	25	2	160	-	-	

