Cool molecular highly charged ions for precision tests of fundamental physics

Carsten Zülch, ¹ Konstantin Gaul, ^{1,*} Steffen M. Giesen, ¹ Ronald F. Garcia Ruiz, ^{2,3} and Robert Berger ^{1,3,†}

¹ Fachbereich Chemie, Philipps-Universität Marburg,

Hans-Meerwein-Straße 4, 35032 Marburg, Germany

² Massachusetts Institute of Technology, Cambridge, MA 02139, USA

³ CERN, CH-1211 Geneva 23, Switzerland

(Dated: March 22, 2022)

Molecules and atomic highly charged ions provide powerful low-energy probes of the fundamental laws of physics: Polar molecules possess internal fields suitable to enhance fundamental symmetry violation by several orders of magnitudes, whereas atoms in high charge states can feature large relativistic effects and compressed level structures, ideally posed for high sensitivity to variations of fundamental constants. Polar, highly charged molecules could benefit from both: large internal fields and large relativistic effects. However, a high charge dramatically weakens chemical bonding and drives systems to the edge of Coulomb explosion. Herein, we propose multiply-charged polar molecules, that contain actinides, as promising candidates for precision tests of physics beyond the standard model. Explicitly, we predict PaF³⁺ to be thermodynamically stable, coolable and well-suited for precision spectroscopy. The proposed class of compounds, especially with short-lived actinide isotopes from the territory of pear-shaped nuclei, has potential to advance our understanding of molecules under extreme conditions, to provide a window into unknown properties of atomic nuclei, and to boost developments in molecular precision spectroscopy in various areas, such as optical clocks and searches for new physics.

As of today, important aspects of our universe are hardly understood, such as the nature of dark matter [1] and the origin of the imbalance between matter and anti-matter [2]. Thus, theories that go beyond the current standard model of particle physics are invoked, usually referred to as new physics. Such new physics introduces additional sources of symmetry violations, like the simultaneous violations of the symmetries with respect to spatial inversion (known as parity P) and of the relative direction of time (known as time-reversal T) [3]. Polar, heavy-elemental molecules like ThO allow currently some of the most precise low-energy tests of P, T-violation [4] as they are easy to polarize and possess large internal fields that enhance effects of new physics by several orders of magnitude compared to atoms [5]. Complementary opportunities to probe new physics are offered by atomic highly charged ions (HCIs) [6]. In these systems the electronic spectra are often compressed due to the deshielded nuclear charge which results in energetically close-lying levels as well as large relativistic effects. This special electronic structure can for instance provide favourable enhancement of hypothetical spatio-temporal variations of the fine structure constant [7] by several orders of magnitude [8]. In contrast to neutral systems, HCIs can be trapped comparatively easily in deep potential wells and cooled by different mechanisms such as sympathetic cooling [9, 10] with well understood atomic ions such as Be⁺, paving the way to high-precision experiments and additionally to direct laser-cooling [6].

Small polar molecular HCIs (PMHCI) can combine advantages of polar molecules and atomic HCIs. In addition to the unification of large relativistic effects of HCIs with large internal fields of polar molecules, as HCIs the PMHCIs could induce a transition from Madelung to Coulomb ordering of electronic levels. Coulomb ordered levels could pave the way for first direct laser-cooling of a molecular ion, as until now laser-cooling of molecular ions was limited due to unfavorable arrangement of electronic states compared to neutral molecules [11, 12]. However, several hurdles have to be overcome: i) Few stable and meta-stable small molecules with charge number larger than two are known [13, 14]. Most long-lived meta-stable triply and quadruply charged polar diatomic molecules, which were proposed theoretically or observed experimentally, are fluorides, oxides or nobel gas compounds of metals [13, 14]. But only one stable triply charged polar diatomic molecule, UF³⁺, is experimentally confirmed as of yet [15]. The difficulty lies in the requirement of very stable bonds that are able to counter the large electrostatic repulsion of two or more positive charges that usually lead to spontaneous Coulomb explosion. ii) Heavy-elemental molecules are preferred for the search for new physics, which severely limits the choice of possible systems, as P, T-violating effects are relativistic in nature and scale steeply with increasing nuclear charge Z [16], Moreover, actinide nuclei such as Pa (Z=91) are predicted to enhance the sensitivity to P, T-violating nuclear properties by up to 5 orders of magnitude when compared to molecules with stable nuclei [17]. iii) For precision spectroscopy, it is is essential to cool and perfectly control the molecule. iv) And finally a simple electronic structure is desired to minimize systematic effects.

In this article we demonstrate that a variety of small PMHCIs suitable for precision tests of fundamental physics exist and propose PaF³⁺ as a promising candidate. We show its stability with respect to Coulomb explosion, characterize its electronic levels, propose how to cool it for precision spectroscopy, and compute its sensitivity to new physics. Moreover, we point out further possibly stable highly charged molecules and anticipate the impact that our results will have on molecular precision experiments in future.

The stability of a triply charged molecule AX³⁺ can be deduced from a simple rule of thumb [13, 18]: The third ionization energy $E_i(A^{2+})$ of atom A should be lower or at least nearly equal to the first ionization energy $E_{i}(X)$ of atom X. Indeed, the third ionization energies of the first five actinides (Ac, Th, Pa, U and Np) are low compared to those of other elements ($< 20\,\mathrm{eV}$) [19–21]. The first ionization energies of neon (21.564 eV), fluorine $(17.4 \,\mathrm{eV})$ [22] oxygen $(13.6 \,\mathrm{eV})$ and nitrogen $(14.5 \,\mathrm{eV})$ are relatively large in comparison [22–24]. Whereas trications with rare gas atoms such as CNe^{3+} have been considered theoretically early on by Koch and Frenking [25], corresponding bond dissociation energies are comparatively low. Instead, we expect actinide fluorides to be most stable followed by nitrides and oxides. UF³⁺ was already shown to be stable [15], but it has two unpaired f-electrons, which may complicate the extraction of fundamental parameters from precision spectroscopy experiments. As uranium has a higher ionization energy than Ac, Th and Pa we can expect that the molecules ${\rm AcF^{3+}}$, ${\rm ThF^{3+}}$ and ${\rm PaF^{3+}}$ are stable as well. In the following we focus on ${\rm PaF^{3+}}$ as this molecule is isoelectronic to RaF, which has a comparatively simple electronic structure and is known to be well suited for the study of fundamental physics [26–29]. Moreover, the isotope ²²⁹Pa attained much attraction as it is supposed to possess an extraordinary large static octupole deformation [30–32], which would render ²²⁹Pa highly powerful for the search for P, T-violations in the quark-sector [33, 34]. Until now, experimental knowledge of Pa isotopes is scarce, and molecules containing Pa isotopes are promising systems to access electroweak properties of Pa nuclei. To our knowledge no molecule containing ²²⁹Pa that is suitable for precision spectroscopy was proposed so far. Here, we show that PaF³⁺ offers a versatile laboratory for precision studies of fundamental physics.

We study the stability of PaF^{3+} with respect to the dissociation into $Pa^{2+} + F^+$, $Pa^{3+} + F$ and $Pa^{4+} + F^-$ with state-of-the-art coupled cluster calculations and quasirelativistic density functional theory. The charge separation dissociation channel $Pa^{2+} + F^+$ is at 4.1 eV, whereas the dissociation channels $Pa^{3+} + F$ and $Pa^{4+} + F^-$ lie above this at 4.9 eV and 32.1 eV, respectively. More details can be found in the methods section. An overview of

the dissociation channels of PaF^{3+} and PaF^{4+} is shown in Fig. 1B and in Tables I to III. We can conclude that PaF^{3+} is very stable. Crude estimates of the repulsive potential for $Pa^{3+} + F^+$ suggest that even PaF^{4+} could be meta-stable, i.e. the charge separation channel lies below the ground state potential but the potentials cross far from equilibrium, with a dissociation barrier $> 1\,\mathrm{eV}$ (see methods section for details).

We computed vertical excitation energies, equilibrium bond lengths and harmonic vibrational wavenumbers of PaF³⁺ on the level of Dirac-Coulomb Fock-Space Coupled Cluster with Singles and Doubles amplitudes (DC-FSCCSD) and within a Zeroth Order Regular Approximation complex Generalized Hartree-Fock (ZORAcGHF) self consistent field maximum overlap approach. With the ZORA-cGHF approach we determine transition electric dipole moments, projection of the orbital angular momentum on the molecular axis and hyperfine coupling constants, here reported for ²³¹PaF³⁺, as nuclear magnetic dipole moments for other Pa isotopes are lacking. ZORA-cGHF calculations can be assumed to be accurate within about 10% in comparison to FSCCSD calculations. And effects of larger basis sets are on the order of 5 % (for details see the methods section and Table IV).

All electronic states explicitly studied herein are below the lowest dissociation channel at 4.1 eV (see Fig. 2 on the left and Table IV; potential curves at other levels of theory are provided in Supplementary Figures S1 – S5). A short bond length of about $1.85\,\mathrm{\AA}$ and a large harmonic vibrational wavenumber $\sim 850\,\mathrm{cm}^{-1}$ indicate a strong Pa-F bond. The level of mixing due to spin-orbit coupling is estimated from the projection of the orbital angular momentum on the molecular axis Λ . We find that the (X)5/2 ground state is of 90 % Φ and 10 % Δ character and the first two excited states are of 80 % Δ and 20% II character [(1)3/2 state] and 60%II and 40% Σ character [(1)1/2 state], respectively. As can be seen in Fig. 2 on the left all electronic states of PaF³⁺ appear to have almost parallel potential curves, with bond lengths and harmonic vibrational wavenumbers varying less than 5%. This indicates that electronic excitations are determined by transitions between non-bonding spinors or spinors with nearly the same bonding or anti-bonding character. To characterize the involved atomic orbitals we compare the complex singly occupied ZORA-cGHF spinor to the large component of the analytic solution of the Dirac equation for the hydrogen atom for principal quantum number n = 5 (see Fig. 2 on the right). This confirms that the excited states are dominated by singleelectron transitions between non-bonding or weakly antibonding orbitals. The seven lowest states are dominated by a singly occupied f-orbital located at Pa, of which the three lowest electronic states are characterized by a 5f_{5/2} orbital, and the next four states are characterized by a $5f_{7/2}$ orbital.

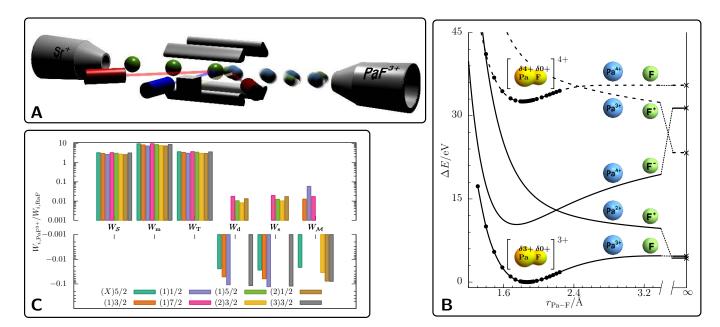


FIG. 1. Schematics of a search for P, T-violation in the quark-sector with the stable highly charged molecule PaF^{3+} . (A) Sketch of an experimental set-up for precision spectroscopy with PaF^{3+} . A decelerated beam of PaF^{3+} is trapped in a Paul trap, sympathetically cooled by Sr^+ ions and probed by the dark red laser. Further lasers for a potential direct cooling of the molecular ion PaF^{3+} are omitted for clarity. (B) Sketch of dissociation channels of PaF^{3+} (solid lines) and PaF^{4+} (dashed lines) as diabatic potentials. Ground state electronic potentials of PaF^{3+} and PaF^{4+} are shown as a spline interpolation of points computed at the level of ZORA-cGKS-PBE0, indicated as circles (molecular calculations) and crosses (separate atomic calculations). Potential curves for the repulsive channels $Pa^{2+} + F^+$ and $Pa^{3+} + F^+$ as well as the attractive channel $Pa^{4+} + F^-$ are represented with the model described in the methods section. Total electronic densities and Mulliken partial charges computed at the level of ZORA-cGKS-PBE0 are shown for the electronic ground states. (C) The parameters of the P, T-odd spin-rotational Hamiltonian of PaF^{3+} at the level of ZORA-cGHF relative to those of RaF computed in Ref. [35] scaled by 6/1.16 (for details see the methods section).

The lowest six transition energies are narrowly spaced and squared transition electric dipole moments $|\vec{\mu}|^2$ indicate low transition rates. This is in accordance with electronic transitions involving primarilly f-type atomic orbitals. Einstein coefficients for spontaneous emission roughly estimated from $|\vec{\mu}|^2$ and T_e at the level of ZORAcGHF considering only electronic degrees of freedom are provided in Table V. From these we infer that the radiative lifetimes of the three lowest electronically excited states [(1)3/2, (1)1/2, (1)7/2] could be on the order of ms, which is on the same order as the lifetime of the $H^3\Delta_1$ state in ThO that was used to provide the so far tightest upper bounds on molecular P, T-violation [4]. PaF³⁺ can be trapped and, due to a mass to charge ratio of about 83 u/e, efficiently cooled sympathetically with Sr⁺ ions as sketched in Fig. 1A. Moreover, other cooling schemes, for instance with buffer gases, can be considered. In addition we find almost diagonal Franck-Condon matrices for all electronic transitions (see Table V) because of the non-bonding or weakly anti-bonding character of the highest occupied spinor and consequently almost parallel potential curves. When combined with an efficient pre-cooling scheme, direct laser-cooling of PaF³⁺ seems feasible. For instance the $(1)7/2 \leftarrow (X)5/2$ transition, which is at about 1800 nm has an estimated cumulated Franck–Condon factor of 0.999997 when taking the 0-0, 1-0 vibrational transitions into account on this level of theory and an estimated lifetime of < 20 ms. Another possibility for direct laser-cooling could be a population of the meta-stable (1)1/2-state and cycling in the (2)1/2 \leftarrow (1)1/2 transition at about 1800 nm, which has a similar estimated Franck–Condon factor (0.999992) but a probably much shorter lifetime (< 70 µs). These properties clearly point to favourable prospects to obtain cold samples of PaF³⁺ for precision experiments.

In order to estimate the enhancement of new physics effects in ${\rm PaF}^{3+}$ we compute the electronic structure parameters W_i of the P,T-violating spin-rotational Hamiltonian [35–37] that reads in good approximation (see methods section for details)

$$H_{\rm sr} = \Omega \left(W_{\rm d} d_{\rm e} + W_{\rm s} k_{\rm s} \right) + \Theta W_{\mathcal{M}} \tilde{\mathcal{M}}$$

$$+ \mathcal{I} \left(W_{\rm T} k_{\rm T} + W_{\mathcal{S}} \mathcal{S} + (W_{\rm m} + W_{\mathcal{S}} R_{\rm vol}) d_{\rm p} \right) ,$$

$$(1)$$

where Ω is the projection of the total electronic angular momentum on the molecular axis, \mathcal{I} is the projection of total spin of Pa on the molecular axis and Θ accounts for the electron and nuclear spin interaction along the molecular axis [35, 37]. We account here for

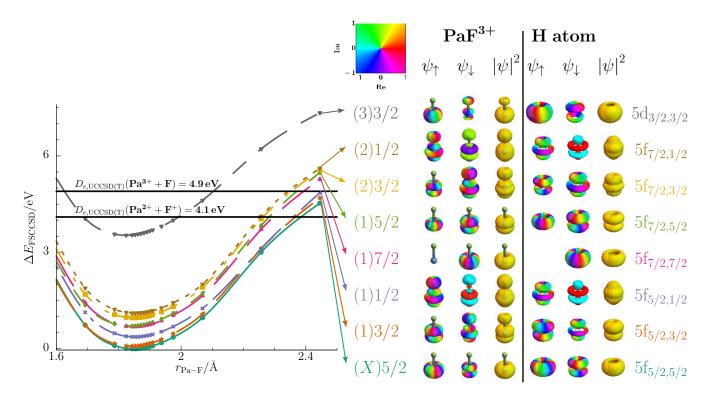


FIG. 2. Potential energy curves for the eight energetically lowest electronic states relative to the ground state of PaF³⁺ computed at the level of DC-FSCCSD/ANO-RCC. Lines between the points are shown to guide the eye. The two lowest dissociation channels Pa²⁺ + F⁺ and Pa³⁺ + F, as computed on a highler level of theory (RECP-UCCSD(T)-SOC), are indicated by black horizontal lines. The various electronic states are additionally characterized by the complex two-component ZORA-cGHF spinor $\psi = (\psi_{\uparrow}, \psi_{\downarrow})$ of the unpaired electron and compared to the upper component of the analytic solution of the Dirac equation for the hydrogen atom. Atomic hydrogen spinors are labelled as ℓ_{j,m_j} , where ℓ is the symbol for the electronic orbital angular momentum quantum number, j is the total electronic angular momentum quantum number and m_j is the magnetic total electronic angular momentum quantum number.

P, T-violation in a molecule via an electric dipole moment of the electron $d_{\rm e}$, an electric dipole moment of the proton d_p , the collective Schiff moment S, the nuclear magnetic quadrupole moment \mathcal{M} , the P, T-odd scalarpseudoscalar nucleon-electron current k_s , and tensorpseudotensor nucleon-electron current $k_{\rm T}$ interactions. $R_{\rm vol}$ is a nuclear structure factor. Ab initio results for the various W_i parameters at the level ZORA-cGHF for the eight lowest electronic states of PaF³⁺ are provided in Table VI and are compared to the isoelectronic RaF molecule in Fig. 1C. The large enhancement of P, T-odd effects that stem from P, T-violation in the quark-sector. In the X(5/2) state the collective Schiff moment is enhanced by $W_{\mathcal{S}} \sim -72\,000\,e/(4\pi\epsilon_0)/a_0^4$, which is more than three times larger in absolute value than in isoelectronic RaF ($\sim -21\,000\,e/(4\pi\epsilon_0)/a_0^4$ computed with the same method [35]). We analyse this large absolute value of $W_{\mathcal{S}}$ in comparison to RaF in Fig. 3. Whereas in RaF the singly occupied molecular orbital (SOMO) has a pronounced s-character, with its contribution to $W_{\mathcal{S}}$ partially cancelling the contribution from the highest doubly occupied orbital (SOMO-1), there is no contribution from the f-type SOMO in PaF³⁺. Furthermore, we see an additive uncompensated contribution from the core orbitals in PaF³⁺ and a much larger contribution from the (SOMO-1), which can be attributed to the pronounced relativistic effects in PaF³⁺. Similar effects are observed for the enhancement factors $W_{\rm T}$ and $W_{\rm m}$. Thereby, the magnetic interaction with a valence proton $W_{\rm m}$ is up to ten times larger than in RaF. We can thus conclude that PaF³⁺ has a pronounced sensitivity to P, T-violation in the quark-sector.

Atomic f-type orbitals, which determine essentially the SOMO of ${\rm PaF}^{3+}$, have vanishing probability density within the nucleus, so that P,T-odd effects dependent on the electron spin $(W_{\rm d},W_{\rm s})$ are suppressed. However, in the possibly meta-stable (1)1/2-state, these effects are still on the same order as in BaF, for which an experiment to search for $d_{\rm e}$ is planned [38]. The relative suppression of electron-spin dependent effects can be an advantage for the disentanglement of the fundamental sources of P,T-violation, for which the ratio of different enhancement factors W_i plays an important role [17, 39]. The ratio of $W_{\rm d}/W_{\rm s}$ is 12% lower than predicted by the model presented in Ref. [39] for Z=91 and, thus, would correspond to Z=97 in the model. This can be ex-

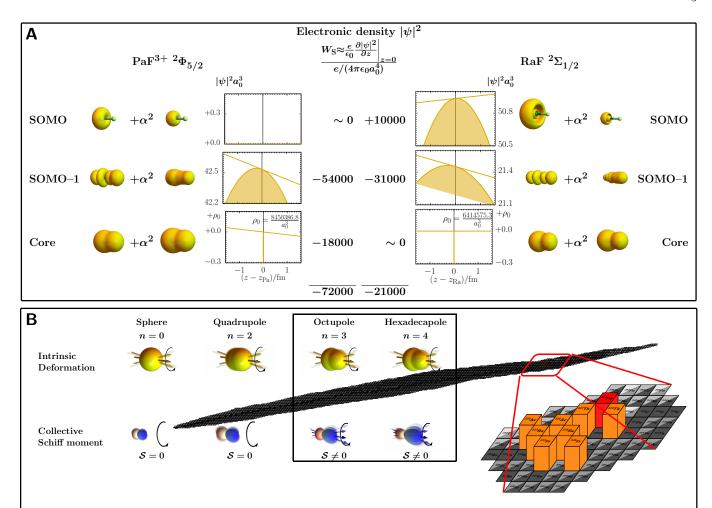


FIG. 3. Visualization of enhancement of P, T-violation in the quark-sector within the PMHCI PaF³⁺ compared to isoelectronic RaF, illustrated by example of the nuclear Schiff moment. This moment would induce an energy shift propotional to $SW_S\mathcal{I}$ (see text). (A) Enhancement as per the electronic structure parameter W_S is shown. Non-vanishing W_S stems from a shift of the maximum electron density away from the nuclear center by polarization, which in good approximation is characterized by the slope of the four-component (4c) electron density (indicated by a line in the plots in the middle) at the position of the Pa or Ra nucleus. Contributions from the SOMO, the highest doubly occupied orbital (SOMO-1) and uncompensated contributions from core electrons are shown. Approximate 4c electron densities computed at the ZORA-cGHF level are plotted with contour value of 0.035^2 a_0^{-3} , with lower component density contributions (right isodensities) being magnified by α^{-2} to enhance visibility as compared to the upper component (left isodensities). In each isodensity plot, the heavy nucleus (Pa, Ra) is located on the left, the fluorine nucleus on the right as indicated by the ball-and-stick structures. (B) Visualization of the strong enhancement of a collective Schiff moment S in octupole deformed nuclei. Large octupole deformations accumulate in the highlighted region of the nuclide chart. Nuclear charge densities with deformation of order n and corresponding angular densities of the Schiff operator are modeled in spherical plots (for details on how this is realized see the methods section). The resulting collective Schiff moments S are indicated by a vector of length S along the deformation axis in the intrinsic frame. The precession of the intrinsic moment around the z-axis is also indicated.

plained by the high charge of the molecule that leads to a higher effective nuclear charge. The ratios between other parameters are considerably different compared to RaF and some even have opposite sign. All this renders experiments with ${\rm PaF}^{3+}$ complementary in the search for P,T-violation to other experiments with polar open-shell molecules

Moreover, PaF³⁺ can also be advantageous for other tests of fundamental physics. The electronic potentials of

PaF³⁺ are largely overlapping, leading to possibly very close lying nearly degenerate vibrational states. In combination with the large relativistic effects this could result in a high sensitivity to a variation of fundamental constants [40].

In summary, highly charged polar actinide molecules can be stable and have favourable properties for precision tests of fundamental physics. Such PMHCIs can be cooled sympathetically and can show a pronounced enhancement of new physics effects in the quark-sector. In particular, we demonstrated that the molecule PaF³⁺ is favorable for precision tests of fundamental physics. PaF³⁺ has a very simple electronic structure with one valence f-electron in a non-bonding or weakly anti-bonding orbital located at Pa. This leads to parallel potential curves for essentially all energetically low-lying electronic states, which may allow to establish an efficient direct laser-cooling scheme after sympathetic pre-cooling with Sr⁺ ions or pre-cooling in a buffer-gas cell. Furthermore, due to the high charge many close lying states can be found that may be advantageous in the search for a variation of fundamental constants.

Beyond PaF^{3+} , we suggest to study also the PMH-CIs UO^{3+} and NpN^{3+} as possibly stable candidates for molecular precision spectroscopy. From the discussion above we infer, moreover, that ThF^{3+} can be a very stable closed-shell highly charged molecule. Furthermore, doubly charged molecular ions, such as ThF^{2+} and PaO^{2+} that are isoelectronic to RaF, might also be considered for high-precision spectroscopy. Some selected molecular properties computed for PaO^{2+} are reported in Table S6-S8 in the Supplementary Material. Previously, PaO^{2+} was studied theoretically [41] and observed in the gas-phase [42]. From this it can be expected that PaO^{2+} has similar properties as PaF^{3+} . Sympathetic cooling or pre-cooling of these doubly charged ions may be achievable with PaC^{2+} ions.

Recent progress [43–45] allows precision searches for new physics with polyatomic molecules and polyatomic molecular ions which provide several experimental benefits [46, 47]. We indicate here for instance PaNC³⁺, PaNCS³⁺ as candidates for possibly stable polyatomic PMHCIs. Following Ref. [48] it may be worthwhile to study [PaNCCH₃]⁴⁺ as potentially long-lived meta-stable symmetric-top PMHCI for a precision search for new physics.

Our study opens up a new route to molecular precision spectroscopy and is a starting point to search for further candidates of molecular HCIs. We anticipate that this class of systems has possible applications as optical clocks, for quantum logic spectroscopy and for precision test of new physics. Molecular HCIs such as PaF³⁺ advance our understanding of chemical bonding, can provide a powerful probe of nuclear electroweak properties and will boost precision searches for new physics beyond the standard model. These studies provide further motivation for the emergent field of short-lived radioactive molecules. This is timely with the development of future radioactive beam facilities such as the Facility for Rare Isotopes (FRIB) in the U.S., which starts operation in 2022, and is expected to produce unprecedented amounts of Pa isotopes and other actinide nuclei.

Acknowledgments.—We thank Nicholas R. Hutzler, Andrei Zaitsevskii and Timur Isaev for discussions and Gernot Frenking for his comments on the manuscript. R.B. is indebted to Helmut Schwarz, Jana Roithova and the late Detlef Schröder for early discussions on highly charged molecular ions. This work was funded by the Deutsche Forschungsgemeinschaft (DFG, German Research Foundation) – Projektnummer 445296313. Computer time provided by the center for scientific computing (CSC) Frankfurt is gratefully acknowledged.

Author contributions.—R.F.G.R. had the initial idea to search for a Pa containing molecule for tests of fundamental physics. K.G and R.B. proposed PaF³⁺ as possible candidate molecule and coordinated the research project. K.G. did preliminary calculations on the stability of PaF³⁺ and its symmetry violating properties and implemented the MOM scheme for quasi-relativistic calculations of excited states. S.M.G. provided an implementation for matrix elements of non-orthogonal wave function and for visualization of complex two-component orbitals. C.Z. did all calculations presented in the final version of the manuscript. C.Z. and K.G. wrote the initial draft of the manuscript. All authors contributed to the discussion and the final version of the manuscript.

Extended Data.—Explicit numerical values for dissociation energies of PaF^{3+} and PaF^{4+} and ionization energies of Pa, F are given at different levels of theory. Spectroscopic constant of PaF^{3+} are provided at the two-component ZORA and FSCCSD levels of theory. Franck-Condon factors, Einstein coefficients and estimated lifetimes for the eight lowest states of PaF^{3+} are given at the level of ZORA-cGHF. Explicit values of P, T-odd properties for the eight lowest electronic states are given at the ZORA two-component level.

Supplementary information.—We provide the potential energy curves of selected electronic states of PaF³⁺ at different levels of theory and the potential energy curve of PaF⁴⁺ at the level of FSCCSD. Selected properties of PaO²⁺ for the lowest three electronic states are given at the level of ZORA-cGKS-B3LYP. We provide all data to reproduce the figures shown in the main text and the supplement.

 $^{^*}$ konstantin.gaul@staff.uni-marburg.de; kongaul@web.de † robert.berger@uni-marburg.de

Spergel, D. N. The dark side of cosmology: Dark matter and dark energy. Science 347, 1100-1102 (2015).
 URL https://science.sciencemag.org/content/347/6226/1100.

 ^[2] Canetti, L., Drewes, M. & Shaposhnikov, M. Matter and antimatter in the universe. New J. Phys. 14, 095012 (2012). URL https://doi.org/10.1088%2F1367-2630% 2F14%2F9%2F095012.

^[3] Gross, D. J. The role of symmetry in fundamental physics. Proc. Natl. Acad. Sci. USA 93, 14256-14259 (1996). URL http://www.pnas.org/content/93/25/14256.full.

^[4] Andreev, V. et al. Improved limit on the electric dipole

- moment of the electron. *Nature* **562**, 355–360 (2018). URL https://doi.org/10.1038/s41586-018-0599-8.
- [5] DeMille, D. Diatomic molecules, a window onto fundamental physics. *Physics Today* 68, 34–40 (2015). URL https://doi.org/10.1063/PT.3.3020. https://doi.org/10.1063/PT.3.3020.
- [6] Kozlov, M. G., Safronova, M. S., Crespo López-Urrutia, J. R. & Schmidt, P. O. Highly charged ions: Optical clocks and applications in fundamental physics. Rev. Mod. Phys. 90, 045005 (2018). URL https://link.aps. org/doi/10.1103/RevModPhys.90.045005.
- [7] Uzan, J.-P. The fundamental constants and their variation: observational and theoretical status. Rev. Mod. Phys. 75, 403-455 (2003). URL https://link.aps.org/doi/10.1103/RevModPhys.75.403.
- [8] Berengut, J. C., Dzuba, V. A. & Flambaum, V. V. Enhanced laboratory sensitivity to variation of the fine-structure constant using highly charged ions. *Phys. Rev. Lett.* 105, 120801 (2010). URL https://link.aps.org/doi/10.1103/PhysRevLett.105.120801.
- [9] Drewsen, M., Brodersen, C., Hornekær, L., Hangst, J. S. & Schifffer, J. P. Large ion crystals in a linear paul trap. Phys. Rev. Lett. 81, 2878-2881 (1998). URL https://link.aps.org/doi/10.1103/PhysRevLett.81.2878.
- [10] Bowe, P. et al. Sympathetic crystallization of trapped ions. Phys. Rev. Lett. 82, 2071-2074 (1999). URL https: //link.aps.org/doi/10.1103/PhysRevLett.82.2071.
- [11] Nguyen, J. H. V. et al. Challenges of laser-cooling molecular ions. New. J. Phys. 13, 063023 (2011). URL https://doi.org/10.1088/1367-2630/13/6/063023.
- [12] Ivanov, M. V., Jagau, T.-C., Zhu, G.-Z., Hudson, E. R. & Krylov, A. I. In search of molecular ions for optical cycling: a difficult road. *Phys. Chem. Chem. Phys.* 22, 17075–17090 (2020). URL http://dx.doi.org/10.1039/D0CP02921A.
- [13] Schröder, D. & Schwarz, H. Generation, stability, and reactivity of small, multiply charged ions in the gas phase. J. Phys. Chem. A 103, 7385-7394 (1999). URL https://doi.org/10.1021/jp991332x.
- [14] Franzreb, K. et al. Gas-phase diatomic trications of Se_2^{3+} , te_2^{3+} , and LaF³⁺. J. Chem. Phys. 121, 12293-12302 (2004). URL https://aip.scitation.org/doi/abs/10. 1063/1.1821496.
- [15] Schröder, D., Diefenbach, M., Klapötke, T. M. & Schwarz, H. UF³⁺ a thermochemically stable diatomic trication with a covalent bond. Angew. Chem. Int. Ed. 38, 137-140 (1999). URL https://onlinelibrary.wiley.com/doi/abs/10.1002/%28SICI%291521-3773%2819990115%2938%3A1/2%3C137%3A%3AAID-ANIE137%3E3.0.C0%3B2-M.
- [16] Khriplovich, I. B. & Lamoreaux, S. K. CP Violation without Strangeness (Springer, Berlin, 1997).
- [17] Chupp, T. E., Fierlinger, P., Ramsey-Musolf, M. J. & Singh, J. T. Electric dipole moments of atoms, molecules, nuclei, and particles. Rev. Mod. Phys. 91, 015001 (2019). URL https://link.aps.org/doi/10.1103/RevModPhys.91.015001.
- [18] Bates, D. R. & Carson, T. R. Doubly charged diatomic molecular ions. Proceedings of the Physical Society. Section A 68, 1199-1202 (1955). URL https://doi.org/10.1088/0370-1298/68/12/417.
- [19] Migdalek, J. & Glowacz-Proszkiewicz, A. Dirac-Fock + core-polarization calculations of E1 transitions in the francium isoelectronic sequence. J. Phys. B 40,

- 4143-4154 (2007). URL https://doi.org/10.1088/0953-4075/40/21/002.
- [20] Wyart, J.-F. & Kaufman, V. Extended analysis of doubly ionized thorium (Th III). Phys. Scr. 24, 941-952 (1981).
 URL https://doi.org/10.1088/0031-8949/24/6/006.
- [21] Cao, X. & Dolg, M. Theoretical prediction of the second to fourth actinide ionization potentials. Mol. Phys. 101, 961–969 (2003). URL https://doi.org/10.1080/0026897021000046807. https://doi.org/10.1080/0026897021000046807.
- [22] Edlén, B. On the identification of Ar x and Ar xiv in the solar corona and the origin of the unidentified coronal lines. Sol. Phys. 9, 439-445 (1969). URL https://doi. org/10.1007/BF02391668.
- [23] Eriksson, K. B. S. & Isberg, H. B. S. New measurements in the spectrum of atomic oxygen, O I. Arkiv. Fysik 24, 549 (1963).
- [24] Eriksson, K. B. S. & Pettersson, J. E. New measurements in the spectrum of the neutral nitrogen atom. *Phys. Scr.* 3, 211–217 (1971). URL https://doi.org/ 10.1088/0031-8949/3/5/003.
- [25] Koch, W. & Frenking, G. Theoretical investigations of small multiply charged cations. ii. CNe^{n+} ($1 \leq n \leq 4$). J. Chem. Phys. 86, 5617–5624 (1987). URL https://doi.org/10.1063/1.452538.
- [26] Isaev, T. A., Hoekstra, S. & Berger, R. Laser-cooled RaF as a promising candidate to measure molecular parity violation. *Phys. Rev. A* 82, 052521 (2010).
- [27] Isaev, T. A. & Berger, R. Lasercooled radium monofluoride: A molecular all-in-one probe for new physics. ArXiv e-prints 1302.5682, physics.chem-ph (2013). URL http://arxiv.org/abs/1302.5682. 1302.5682.
- [28] Garcia Ruiz, R. F. et al. Spectroscopy of short-lived radioactive molecules. Nature 581, 396–400 (2020). URL https://doi.org/10.1038/s41586-020-2299-4.
- [29] Udrescu, S. M. et al. Isotope shifts of radium monofluoride molecules. Phys. Rev. Lett. 127, 033001 (2021). 2105.10549.
- [30] Ahmad, I., Gindler, J. E., Betts, R. R., Chasman, R. R. & Friedman, A. M. Possible ground-state octupole deformation in ²²⁹Pa. Phys. Rev. Lett. 49, 1758-1761 (1982). URL https://link.aps.org/doi/ 10.1103/PhysRevLett.49.1758.
- [31] Ahmad, I., Chasman, R. R., Greene, J. P., Kondev, F. G. & Zhu, S. Electron capture decay of 58-min ²²⁹₉₂U and levels in ²¹⁹₉₁Pa. Phys. Rev. C 92, 024313 (2015). URL https://link.aps.org/doi/10.1103/PhysRevC.92.024313.
- [32] Singh, J. T. A new concept for searching for time-reversal symmetry violation using Pa-229 ions trapped in optical crystals. *Hyperfine Interact.* **240**, 29 (2019). URL https://doi.org/10.1007/s10751-019-1573-z.
- [33] Auerbach, N., Flambaum, V. V. & Spevak, V. Collective T- and P-odd electromagnetic moments in nuclei with octupole deformations. *Phys. Rev. Lett.* 76, 4316–4319 (1996). URL https://link.aps.org/doi/10.1103/PhysRevLett.76.4316.
- [34] Flambaum, V. V. Enhanced nuclear Schiff moment and time-reversal violation in ²²⁹Th-containing molecules. Phys. Rev. C 99, 035501 (2019). URL https://link. aps.org/doi/10.1103/PhysRevC.99.035501.
- [35] Gaul, K. & Berger, R. Toolbox approach for quasirelativistic calculation of molecular properties for precision tests of fundamental physics. J. Chem. Phys.

- **152**, 044101 (2020). URL https://doi.org/10.1063/1.5121483. 1907.10432.
- [36] Hinds, E. A. & Sandars, P. G. H. Electric dipole hyperfine structure of TlF. Phys. Rev. A 21, 471– 479 (1980). URL https://link.aps.org/doi/10.1103/ PhysRevA.21.471.
- [37] Kozlov, M. G. & Labzowsky, L. N. Parity violation effects in diatomics. J. Phys. B 28, 1933–1961 (1995).
- [38] Aggarwal, P. et al. Measuring the electric dipole moment of the electron in BaF. Eur. Phys. J. D 72, 197 (2018). URL https://doi.org/10.1140/epjd/e2018-90192-9.
- [39] Gaul, K., Marquardt, S., Isaev, T. & Berger, R. Systematic study of relativistic and chemical enhancements of P, T-odd effects in polar diatomic radicals. Phys. Rev. A 99, 032509 (2019). URL https://link.aps.org/doi/10.1103/PhysRevA.99.032509. 1805.05494.
- [40] Chin, C., Flambaum, V. V. & Kozlov, M. G. Ultracold molecules: new probes on the variation of fundamental constants. New J. Phys. 11, 055048 (2009). URL https: //doi.org/10.1088/1367-2630/11/5/055048.
- [41] Kovács, A., Infante, I. & Gagliardi, L. Theoretic study of the electronic spectra of neutral and cationic PaO and PaO₂. Struct. Chem. **24**, 917–925 (2013). URL https://doi.org/10.1007/s11224-013-0251-z.
- [42] Santos, M. et al. Oxidation of gas-phase protactinium ions, Pa⁺ and Pa²⁺: Formation and properties of PaO₂²⁺(g), Protactinyl. J. Phys. Chem. A 110, 5751–5759 (2006). URL https://doi.org/10.1021/jp057297d. PMID: 16640369, https://doi.org/10.1021/jp057297d.
- [43] Isaev, T. A. & Berger, R. Polyatomic candidates for cooling of molecules with lasers from simple theoretical concepts. *Phys. Rev. Lett.* 116, 063006 (2016). URL https://link.aps.org/doi/10.1103/ PhysRevLett.116.063006.
- [44] Kozyryev, I., Baum, L., Matsuda, K. & Doyle, J. M. Proposal for laser cooling of complex polyatomic molecules. *ChemPhysChem* 17, 3641-3648 (2016). URL https://onlinelibrary.wiley.com/doi/abs/10. 1002/cphc.201601051.
- [45] Fan, M. et al. Optical mass spectrometry of cold RaOH⁺ and RaOCH₃⁺. Phys. Rev. Lett. 126, 023002 (2021). URL https://link.aps.org/doi/10.1103/ PhysRevLett.126.023002.
- [46] Isaev, T. A., Zaitsevskii, A. V. & Eliav, E. Laser-coolable polyatomic molecules with heavy nuclei. J. Phys. B 50, 225101 (2017). URL http://stacks.iop.org/0953-4075/50/i=22/a=225101.
- [47] Kozyryev, I. & Hutzler, N. R. Precision measurement of time-reversal symmetry violation with laser-cooled polyatomic molecules. *Phys. Rev. Lett.* 119, 133002 (2017). URL https://link.aps.org/doi/10.1103/PhysRevLett.119.133002.
- [48] Harvey, J. N. & Kaczorowska, M. Microsolvation of metal ions: on the stability of [Zr(CH₃CN)]⁴⁺ and other multiply charged ions. *Int. J. Mass Spec.* 228, 517– 526 (2003). URL https://www.sciencedirect.com/ science/article/pii/S138738060300160X. Special Issue: In honour of Helmut Schwarz.
- [49] DIRAC, a relativistic ab initio electronic structure program, Release DIRAC19 (2019), written by A. S. P. Gomes, T. Saue, L. Visscher, H. J. Aa. Jensen, and R. Bast, with contributions from I. A. Aucar, V. Bakken, K. G. Dyall, S. Dubillard, U. Ekström,

- E. Eliav, T. Enevoldsen, E. Faßhauer, T. Fleig, O. Fossgaard, L. Halbert, E. D. Hedegård, B. Heimlich-Paris, T. Helgaker, J. Henriksson, M. Iliaš, Ch. R. Jacob, S. Knecht, S. Komorovský, O. Kullie, J. K. Lærdahl, C. V. Larsen, Y. S. Lee, H. S. Nataraj, M. K. Nayak, P. Norman, G. Olejniczak, J. Olsen, J. M. H. Olsen, Y. C. Park, J. K. Pedersen, M. Pernpointner, R. di Remigio, K. Ruud, P. Sałek, B. Schimmelpfennig, B. Senjean, A. Shee, J. Sikkema, A. J. Thorvaldsen, J. Thyssen, J. van Stralen, M. L. Vidal, S. Villaume, O. Visser, T. Winther, and S. Yamamoto (available at http://dx.doi.org/10.5281/zenodo.3572669, see also http://www.diracprogram.org).
- [50] Dyall, K. G. Relativistic and nonrelativistic finite nucleus optimized triple-zeta basis sets for the 4p, 5p and 6p elements. Theor. Chem. Acc. 108, 335–340 (2002). URL http://dx.doi.org/10.1007/s00214-002-0388-0.
- [51] Dyall, K. G. Relativistic quadruple-zeta and revised triple-zeta and double-zeta basis sets for the 4p, 5p, and 6p elements. Theor. Chem. Acc. 115, 441–447 (2006). URL http://dx.doi.org/10.1007/s00214-006-0126-0.
- [52] Roos, B. O., Lindh, R., Malmqvist, P., Veryazov, V. & Widmark, P.-O. New relativistic ANO basis sets for actinide atoms. *Chem. Phys. Lett.* 409, 295–299 (2005). URL https://www.sciencedirect.com/science/article/pii/S0009261405006810.
- [53] van Wüllen, C. A Quasirelativistic Two-component Density Functional and Hartree-Fock Program. Z. Phys. Chem 224, 413–426 (2010).
- [54] Ahlrichs, R., Bär, M., Häser, M., Horn, H. & Kölmel, C. Electronic structure calculations on workstation computers: The program system turbomole. *Chem. Phys. Lett.* 162, 165–169 (1989).
- [55] van Wüllen, C. Molecular density functional calculations in the regular relativistic approximation: Method, application to coinage metal diatomics, hydrides, fluorides and chlorides, and comparison with first-order relativistic calculations. J. Chem. Phys. 109, 392–399 (1998).
- [56] Liu, W., van Wüllen, C., Wang, F. & Li, L. Spectroscopic constants of MH and M₂ (M = Tl, E113, Bi, E115): direct comparisons of four- and two-component approaches in the framework of relativistic density functional theory. J. Chem. Phys. 116, 3626–3634 (2002).
- [57] Vosko, S. H., Wilk, L. & Nuisar, M. Accurate spindependent electron liquid correlation energies for local spin density calculations: A critical analysis. *Can. J. Phys.* 58, 1200–1211 (1980).
- [58] Becke, A. D. Density-functional exchange-energy approximation with correct asymptotic-behavior. *Phys. Rev. A* 38, 3098–3100 (1988).
- [59] Lee, C., Yang, W. & Parr, R. G. Development of the Colle-Salvetti correlation-energy formula into a functional of the electron-density. *Phys. Rev. B* 37, 785–789 (1988).
- [60] Stephens, P. J., Devlin, F. J., Chabalowski, C. F. & Frisch, M. J. Ab initio calculation of vibrational absorption and circular dichroism spectra using density functional force fields. J. Phys. Chem. 98, 11623–11627 (1994).
- [61] Perdew, J. P., Burke, K. & Ernzerhof, M. Generalized gradient approximation made simple. *Phys. Rev. Lett.* 77, 3865–3868 (1996).
- [62] Adamo, C. & Barone, V. Toward reliable density func-

- tional methods without adjustable parameters: The PBE0 model. J. Chem. Phys. 110, 6158–6170 (1999). URL https://doi.org/10.1063/1.478522.
- [63] Roos, B. O., Lindh, R., Malmqvist, P., Veryazov, V. & Widmark, P. O. Main Group Atoms and Dimers Studied with a New Relativistic ANO Basis Set. J. Phys. Chem. A 108, 2851–2858 (2004).
- [64] Gaul, K. & Berger, R. Zeroth order regular approximation approach to electric dipole moment interactions of the electron. J. Chem. Phys. 147, 014109 (2017).
- [65] Visscher, L. & Dyall, K. G. Dirac-fock atomic electronic structure calculations using different nuclear charge distributions. At. Data Nucl. Data Tables 67, 207–224 (1997).
- [66] Gilbert, A. T. B., Besley, N. A. & Gill, P. M. W. Self-consistent field calculations of excited states using the maximum overlap method (MOM). J. Phys. Chem. A 112, 13164–13171 (2008). URL https://doi.org/10.1021/jp801738f. PMID: 18729344, https://doi.org/10.1021/jp801738f.
- [67] Barca, G. M. J., Gilbert, A. T. B. & Gill, P. M. W. Simple models for difficult electronic excitations. J. Chem. Theo. Comp. 14, 1501–1509 (2018). URL https://doi.org/10.1021/acs.jctc.7b00994. PMID: 29444408, https://doi.org/10.1021/acs.jctc.7b00994.
- [68] Berger, R., Fischer, C. & Klessinger, M. Calculation of the vibronic fine structure in electronic spectra at higher temperatures. 1. benzene and pyrazine. J. Phys. Chem. A 102, 7157–7167 (1998).
- [69] Jankowiak, H.-C., Stuber, J. L. & Berger, R. Vibronic transitions in large molecular systems: Rigorous prescreening conditions for Franck-Condon factors. J. Chem. Phys. 127, 234101 (2007).
- [70] Huh, J. & Berger, R. Coherent state-based generating function approach for Franck-Condon transitions and beyond. In SYMMETRIES IN SCIENCE XV, vol. 380 of J. Phys. Conf. Ser. (2012). International Symposium on Symmetries in Science XV, Bregenz, AUSTRIA, JUL 31-AUG 05, 2011.
- [71] Huh, J., Neff, M., Rauhut, G. & Berger, R. Franck-Condon profiles in photodetachment-photoelectron spectra of HS₂⁻ and DS₂⁻ based on vibrational configuration interaction wavefunctions. *Mol. Phys.* 108, 409 (2010).
- [72] Löwdin, P.-O. Quantum theory of many-particle systems .1. physical interpretations by means of density matrices, natural spin-orbitals, and convergence problems in the method of configurational interaction. *Phys. Rev.* 97, 1474–1489 (1955).
- [73] Flambaum, V. V. & Ginges, J. S. M. Nuclear Schiff moment and time-invariance violation in atoms. *Phys. Rev. A* 65, 032113 (2002). URL https://link.aps.org/doi/10.1103/PhysRevA.65.032113.
- [74] Flambaum, V. V., Dzuba, V. A. & Tran Tan, H. B. Time- and parity-violating effects of the nuclear schiff moment in molecules and solids. *Phys. Rev. A* 101, 042501 (2020). URL https://link.aps.org/doi/10. 1103/PhysRevA.101.042501.
- [75] Werner, H.-J., Knowles, P. J., Knizia, G., Manby, F. R. & Schütz, M. Molpro: a general-purpose quantum chemistry program package. WIREs Computational Molecular Science 2, 242-253 (2012). URL https://wires.onlinelibrary.wiley.com/doi/abs/10.1002/wcms.82.
- [76] Werner, H.-J. et al. Molpro, version 2019.2, a package of ab initio programs (2019). See http://www.molpro.net.

- [77] Werner, H.-J. et al. The molpro quantum chemistry package. J. Chem. Phys. 152, 144107 (2020). URL https://doi.org/10.1063/5.0005081.
- [78] Cao, X., Dolg, M. & Stoll, H. Valence basis sets for relativistic energy-consistent small-core actinide pseudopotentials. J. Chem. Phys. 118, 487–496 (2003). URL https://doi.org/10.1063/1.1521431.
- [79] Dunning, T. H., Jr. Gaussian basis sets for use in correlated molecular calculations. i. the atoms boron through neon and hydrogen. J. Chem. Phys. 90, 1007–1023 (1989).
- [80] Wolfram Research, Inc. Mathematica 11.0 (2016).
- [81] Batista, E. R., Martin, R. L. & Hay, P. J. Density functional investigations of the properties and thermochemistry of UF_n and UCl_n (n = 1,...,6). J. Chem. Phys. 121, 11104–11111 (2004). URL https: //aip.scitation.org/doi/abs/10.1063/1.1811607. https://aip.scitation.org/doi/pdf/10.1063/1.1811607.
- [82] Blondel, C., Delsart, C. & Goldfarb, F. Electron spectrometry at the μev level and the electron affinities of si and f. J. Phys. B: Atom. Mol. Opt. Phys. 34, L281–L288 (2001). URL http://stacks.iop.org/0953-4075/34/i=9/a=101.
- [83] Walker, N. R., Wright, R. R., Barran, P. E., Murrell, J. N. & Stace, A. J. Comparisons in the behavior of stable Copper(II), Silver(II), and Gold(II) complexes in the gas phase: Are there implications for condensed-phase chemistry? J. Am. Chem. Soc. 123, 4223–4227 (2001). URL https://doi.org/10.1021/ja003431q. PMID: 11457187.
- [84] Axe, J. D., Stapleton, H. J. & Jeffries, C. D. Paramagnetic resonance hyperfine structure of tetravalent Pa²³¹ in Cs₂ZrCl₆. Phys. Rev. 121, 1630-1637 (1961). URL https://link.aps.org/doi/10.1103/ PhysRev.121.1630.

METHODS

Computational methods

Relativistic four component calculations were performed with the quantum chemistry program package Dirac19 [49] employing the Dirac-Coulomb (DC) Hamiltonian. The four-component Dirac-Coulomb Hartree-Fock (DC-HF) method was used to describe the electronic structure of Fluorine and Protactinium in different electron configurations, with the open-shell situation being treated within the average-of-configuration framework (ORHF). Relativistic Fock-space coupledcluster calculations with singles and doubles cluster amplitudes (DC-FSCCSD) were performed starting from the PaF⁴⁺ closed-shell electronic ground state as reference wavefunction and attaching an additional electron. Seven electrons of F $(2s^22p^5)$ and 19 electrons of Pa $(5d^{10}6s^26p^65f^1)$ were explicitly included in the electron correlation treatment. Virtual spinors were considered up to an energy of $50 E_h$. In DC-ORHF and DC-FSCCSD calculations we employed the Dyall all-electron triple zeta (Dyall-ae3z) basis set for the Pa and F atom [50, 51] or a triple zeta atomic natural orbital type basis set (ANO-RCC-VTZP) [52]. To assess the quality of the basis set, additional calculations were performed with the Dyall all-electron quadruple zeta (Dyall-ae4z) basis set [50, 51]. DC-ORHF spinors were optimized in a selfconsistent manner until a change in the orbital gradient below $10^{-7}/a_0^{3/2}$ was reached.

Quasirelativistic two-component calculations were performed with a modified version [53] of the quantum chemistry program package TURBOMOLE [54] at the level of complex generalized Hartree-Fock (cGHF) and complex generalized Kohn-Sham (cGKS) density functional theory (DFT) within the zeroth order regular approximation (ZORA) employing a model potential to alleviate the gauge dependence of the ZORA Hamiltonian as proposed by van Wüllen [55]. The ZORA-model potential was employed with additional damping [56]. Calculations on the DFT level were performed with the hybrid Becke three parameter exchange functional and the Lee, Yang and Parr correlation functional (B3LYP) [57–60], as well as with the hybrid version of the Perdew, Burke, Ernzerhof functional (PBE0) [61, 62]. On this level of theory, a basis set consisting of 37s, 34p, 14d and 9f uncontracted Gaussian functions with the exponential coefficients α_i composed as an even-tempered series as $\alpha_i = a/b^{i-1}$; i = 1, ..., N, where N is the number of functions, with b=2 for s and p functions and $b=(5/2)^{1/25}\times 10^{2/5}\approx 2.6$ for d and f functions was used for Pa. The largest exponent coefficients a of the subsets are $2\times 10^9~a_0^{-2}$ (s), $5\times 10^8~a_0^{-2}$ (p), $13300.758~a_0^{-2}$ (d) and $751.8368350~a_0^{-2}$ (f). A decontracted atomic natural orbital basis set of double- ζ quality augmented with polarization valence basis functions (ANO-RCC-VDZP) [63] was used for the F atom. These basis sets have performed well in previous studies of P, T-violation in molecules [27, 39, 64].

In all relativistic or quasi-relativistic calculations, a normalized spherical Gaussian nuclear density distribution $\rho_A\left(\vec{r}\right) = \frac{\zeta_A^{3/2}}{r^{3/2}} \mathrm{e}^{-\zeta_A |\vec{r}-\vec{r}_A|^2}$ with $\zeta_A = \frac{3}{2r_{\mathrm{nuc},A}^2}$ and the root-mean-square radius $r_{\mathrm{nuc},A}$ was used as a finite nucleus as suggested by Visscher and Dyall [65]. Isotopes $^{231}\mathrm{Pa}$ and $^{19}\mathrm{F}$ were used to determine the size of the finite nucleus. An exception to this are calculations of the energy gradient within the ZORA-cGKS approach, which were performed assuming a pointlike nucleus instead.

At the level of cGKS and cGHF, excited state orbitals were obtained by SCF calculations choosing occupation numbers regarding to maximum overlap with the determinant of the initial guess (initial guess maximum overlap method, IMOM) [66, 67]. As initial guess we used the cGKS or cGHF determinant, that was found with occupation of energetically lowest spinors. If the change in the differential density with respect to the previous cycle was below $10^{-3}/a_0^{-3}$ the standard MOM was used, where occupation numbers are chosen with respect to maximum overlap with the determinant of the previous cycle [66].

Franck–Condon factors were obtained with the program package HOTFCHT [68–71] by calculating the harmonic force constants with the module NumForce within the modified version of Turbomole mentioned above.

Transition electric dipole moments $\vec{\mu}$ were computed for the independently obtained cGHF and cGKS determinants using Löwdin rules [72] for single-particle operator transition matrix elements between nonorthogonal single-determinantal initial wave function $\Phi_{\rm i}$ and final wave function $\Phi_{\rm f}$

$$\vec{\mu} = \langle \Phi_{\mathrm{f}} | \vec{r} | \Phi_{\mathrm{i}} \rangle = \sum_{ij} \langle \psi_{\mathrm{f},i} | \vec{r} | \psi_{\mathrm{i},j} \rangle \operatorname{adj}(\mathbf{S})_{ij}$$
 (2)

with the molecular spinors ψ_i of the initial (i) and final (f) state, approximately described by a single Slater determinant, and the adjugate of the overlap matrix between the initial and final state determinant with elements $S_{ij} = \langle \psi_{\mathbf{f},i} | \psi_{\mathbf{i},j} \rangle$.

All other molecular properties were computed with the toolbox approach presented in Ref. [35]. We neglected magnetic and hyperfine coupling induced interactions of d_e and k_s as well as higher order P, T-odd nuclear moments and described the nuclei in the non-relativistic limit. For a definition of the various electronic structure factors W_i except for W_S we refer to Ref.35. We used here a different definition of W_S than in Ref.35 which includes a factor of 6. Furthermore, we employed a finite nucleus model for the calculation of the W_S operator [73], whereas in Ref.35 a point-like nulcear model was used:

$$W_{\mathcal{S}} = -4\pi k_{\rm es} e \left\langle \Psi \left| \frac{\partial}{\partial z} \rho_A(\vec{r}) \right| \Psi \right\rangle$$

$$= 8\zeta_A \pi k_{\rm es} e \left\langle \Psi \left| (z - z_A) \rho_A(\vec{r}) \right| \Psi \right\rangle ,$$
(3)

where the electrostatic constant $k_{\rm es}$ is $\frac{1}{4\pi\epsilon_0}$ in SI units and ϵ_0 is the electric constant. The last equality applies only to the spherical Gaussian nuclear model used in this work. In opposite to the operator used in Ref.35 this operator can be evaluated in numerical integration within the toolbox apporach. As suggested in Ref.74, for direct comparison to the present values computed with a finite nuclear model for PaF³⁺, we scaled the corresponding values of W_S of RaF that were obtained with the operator for a pointlike nucleus in Ref.35 by a factor of 1/1.6.

Bond lengths were optimized to an energy change of less than 10^{-6} $E_{\rm h}$ as convergence criterion. The wavefunctions were optimized to a change in energy and spinorbit coupling contribution of 10^{-11} $E_{\rm h}$ or better, with exception of the calculations with the ANO-RCC-VTZP basis set, which was optimized to a change in energy and spin-orbit coupling contribution of 10^{-9} $E_{\rm h}$.

Calculations in a non-relativistic framework using scalar relativistic ECPs (RECP), which are reported in Tables I-III, were performed with the quantum chemistry program package Molpro [75-77] on the level of restricted open-shell Hartree-Fock (ROHF), spinunrestricted-Kohn-Sham (UKS) using the functionals B3LYP and PBE0 and on the level of unrestricted coupled-cluster with iterative singles and doubles amplitudes combined with perturbative triples amplitudes (UCCSD(T)). For Pa a relativistic energy-consistent small-core pseudopotential (ECP60) was used together with atomic natural orbital valence basis set [78]. An augmented correlation-consistent polarized basis with quadruple- ζ qualtiy (aug-cc-pVQZ) [79] was used on the F-atom. The bond length was optimized up to a change in energy of 10^{-6} $E_{\rm h}$. Self consistent field calculations were performed until a change in the gradient in respect to the orbital rotation lower than $10^{-13} E_h/a_0$ was

The complex two-component orbitals are visualized by calculating orbital amplitudes on a three-dimensional grid and plotting them with the help of Mathematica version 11 [80] by mapping the phase in the complex plane via a color code on the contour surface of the absolute value of the spinors.

Disscociation channels

We study the stability of PaF^{3+} with respect to the dissociation into $Pa^{2+} + F^+$, $Pa^{3+} + F$ and $Pa^{4+} + F^-$ by separate energy calculations of the atomic products and the corresponding molecular species at its equilibrium structure with state-of-the-art unrestricted coupled cluster calculations with single and double amplitudes and preturbative triples [UCCSD(T)], in which we account for scalar-relativistic effects by an relativistic effective core potential (RECP). We find that the charge separation dissociation channel $Pa^{2+} + F^+$ is

at $5.1\,\mathrm{eV}$. The dissociation channels $\mathrm{Pa^{3+}}$ + F and $Pa^{4+} + F^{-}$ lie above this at 5.8 eV and 32 eV, respectively. In these calculations spin-orbit coupling is, however, not accounted for. To quantify this effect, quasirelativistic two-component complex generalized Kohn-Sham (ZORA-cGKS) calculations within zeroth order regular approximation are compared to relativistic effective core potential unrestriced Kohn-Sham calculations (RECP-UKS), using the PBE0 functional, which performed well in the computation of bond dissociation energies of uranium halides [81]. The size of the spin-orbit coupling contribution to the three dissociation channels is $-1.0\,\mathrm{eV}$, $-0.9\,\mathrm{eV}$ and $0.1\,\mathrm{eV}$, respectively, yielding the spin-orbit corrected [RECP-UCCSD(T)+SOC] dissociation energies of 4.1 eV, 4.9 eV and 32.1 eV, which are in good agreement with dissociation energies at the level of ZORA-cGKS-PBE0 (3.8 eV, 4.7 eV and 32.5 eV). The RECP-UCCSD(T)+SOC method is in good agreement with ionization energies of Pa computed in Ref.21 (deviations < 5%) and experimental ionization energy and electron affinity [22, 82] of F (deviations < 1%).

From ab initio calculations we find that the dissociation channel $Pa^{3+} + F^+$ is placed $\sim 10\,\mathrm{eV}$ below the equilibrium energy of PaF^{4+} . A crude estimate for a repulsive potential V(r) as a function of the internuclear separation r can be obtained from purely repulsive Coulomb and Pauli potentials [48, 83]. Assuming that the ionic radius r_0 is equal for charge separation and homolytic dissociation, we model the Pauli repulsion by a Lennard-Jones potential $C^{(12)}/(r+r_0)^{12}-C^{(6)}/(r+r_0)^6$ for a fixed dissociation energy (indicated by crosses in Fig. 1 on the right), which is fitted to the ground state potentials of PaF³⁺ and PaF⁴⁺ computed on the level of ZORAcGKS-PBE0 as indicated by circles in Fig. 1 on the right to obtain $V(r) = \Delta E + q_1 q_2 / (4\pi \epsilon_0 (r + r_0)) + C^{(12)} / (r + r_0)$ r_0)¹². Here ΔE is received at the level of ZORA-cGKS-PBE0 as described above and q_1q_2 is determined by the charge of the fragments. The fit of the ground state potential of PaF³⁺ yields $r_0 = 1.55 \,\text{Å}$, $C^{(12)} = 1.2 \times 10^7 \, \text{eV} \,\text{Å}^{12}$ and $C^{(6)} = 1.5 \times 10^4 \, \text{eV} \,\text{Å}^{6}$. The fit of the ground state potential of PaF^{4+} yields $r_0 = 1.00 \,\text{Å}$, $C^{(12)} = 6.9 \times 10^5 \,\text{eV} \,\text{Å}^{12} \text{ and } C^{(6)} = 2.7 \times 10^3 \,\text{eV} \,\text{Å}^{6}$. Relative to the ground state energy of PaF³⁺ we receive for the Pa²⁺ + F⁺ channel $\Delta E = 3.8 \,\text{eV}, \, r_0 = 1.55 \,\text{Å},$ $C^{(12)} = 1.2 \times 10^7 \,\mathrm{eV}\,\text{Å}^{12}, \, q_1 q_2 / (4\pi\epsilon_0) = 28.8 \,\mathrm{eV}\,\text{Å}$ yielding an avoided crossing at $> 30\,\text{Å}$ and for the Pa³⁺ $+ F^{+}$ channel $\Delta E = 35.5 \,\text{eV}, r_0 = 1.00 \,\text{Å}, C^{(12)} =$ $6.9 \times 10^5 \,\mathrm{eV}\,\text{Å}^{12}, \ q_1 q_2 / (4\pi\epsilon_0) = 43.2 \,\mathrm{eV}\,\text{Å}$ yielding an avoided crossing at $> 2.4 \,\text{Å}$ (see Fig. 1 right). This suggests that PaF⁴⁺ could be meta-stable with a dissociation barrier > 1 eV. In this model the potential for the Pa⁴⁺ + F⁻ channel is modeled as V(r) = $32 \,\mathrm{eV} - 57.6 \,\mathrm{eV} \,\text{Å}/(r + 1.55 \,\text{Å}) + 1.2 \times 10^7 \,\mathrm{eV} \,\text{Å}^{12}/(r +$ $1.55 \,\text{Å})^{12} - 1.7 \times 10^4 \,\text{eV} \,\text{Å}^6/(r + 1.55 \,\text{Å})^6.$

In calculations of the dissociation energy (Table I) the

mean-field RECP-ROHF and ZORA-cGHF methods underestimate the dissociation energy dramatically as missing electron correlation destabilizes considerably the Fluoride ions (see also Table II). Moreover, in the ZORA-cGHF method spin-orbit coupling and spin-polarization are incorporated self-consistently. As both effects stabilize the Pa cations, the ZORA-cGHF values are much lower than those at correlated levels of theory due this imbalanced consideration of electron-correlation effects.

Electronic excitation energies

In calculations of the electronic spectra (Table IV) ZORA-cGHF results are in a good agreement with DC-FSCCSD results, with deviations in excitation wavenumbers of maximally $600\,\mathrm{cm^{-1}}$ (for higher excitations $1100\,\mathrm{cm^{-1}}$) and deviations of harmonic vibrational wavenumbers and equilibrium bond lengths $\leq 10\,\%$. We expect other molecular properties at the level of ZORA-cGHF to be accurate within $10\,\%$. Furthermore, the effect of a larger basis set (ANO-RCC vs dyall.3aez vs dyall.4aez) in FSCCSD calculations is found to be $< 5\,\%$.

Visualisation of nuclear densities

Nuclear charge densities with deformation of order n visualised in Fig. 3 on the bottom are realised as spherical plots of a Rayleigh expansion with axial symmetry $R_n(\theta,\phi,r)=\left[1+\sum_{l=2}^n a_l r^l Y_{l,0}(\theta,\phi)\right]$ averaged over the radial part as $\int R_n(\theta,\phi,r) \exp[-2/(3\left\langle r^2\right\rangle)r^2]r^2 dr$, with $a_2=0.231,\ a_3=0.097,\ a_4=0.04$ and $\left\langle r^2\right\rangle=1$. These coefficients are chosen for optimal representation and have no physical meaning. The Schiff moment operator can be written as $\hat{S}=(r^3-5/2\left\langle r^2\right\rangle r)Y_{10}(\theta,\phi)$. The corresponding Schiff moments are modeled in a spherical plot of the angular function $S(\theta,\phi)=\int \hat{S}R_n(\theta,\phi,r)\exp[-2/(3\left\langle r^2\right\rangle)r^2]r^2 dr$, with the resulting moment being calculated as $\mathcal{S}=\int S(\theta,\phi)\sin(\theta)\mathrm{d}\theta\mathrm{d}\phi$.

TABLE I. Dissociation energies of PaF³⁺ given for the three most probable dissociation channels. The dissociation channels correspond to $D_e(A^{n+} + B^{[+,0,-]}) = E(A^{n+}) + E(B^{[+,0,-]}) - E(AB^{[(n+1)+,n+,(n-1)+]})$.

Method	$D_e(\mathrm{Pa}^{2+}+\mathrm{F}^+)/\mathrm{eV}$	$D_e(\mathrm{Pa}^{3+}+\mathrm{F})/\mathrm{eV}$	$D_e(\mathrm{Pa^{4+}+F^-})/\mathrm{eV}$
RECP-ROHF	2.7	4.2	30.8
$\mathrm{ZORA\text{-}cGHF}^*$	0.3	2.7	30.5
RECP-UKS-B3LYP	4.7	5.6	32.7
ZORA-cGKS-B3LYP	4.1	4.7	32.2
RECP-UKS-PBE0	4.8	5.6	32.4
ZORA-cGKS-PBE0	3.8	4.7	32.5
RECP-UCCSD(T)	5.1	5.8	32.0
RECP-UCCSD(T)+SOC	4.1	4.9	32.1

^{*} The ZORA-cGHF method underestimates the dissociation energy dramatically (see discussion in the methods section).

TABLE II. Relevant n-th ionization energies $E_{\mathbf{i}}$ and electron affinities $E_{\mathbf{ea}}$ for Pa and F in comparison with literature (lit) values. Relative deviation is given as dev = (calculation - lit)/lit.

Method	$E_{\rm i}\left({\rm Pa}^{2+}\right)/{\rm eV}$	dev / $\%$	$E_{\rm i}\left({\rm Pa^{3+}}\right)/{\rm eV}$	dev/%	$E_{\rm i}\left({ m F}\right)/{ m eV}$	$\mathrm{dev}/\%$	$E_{\mathrm{ea}}\left(\mathrm{F}\right)/\mathrm{eV}$	$\mathrm{dev}/\%$
RECP-ROHF	17.2	-8	27.9	-10	15.7	-10	1.31	-64
ZORA-cGHF	18.0	-4	29.1	-6	15.6	-10	1.27	-63
RECP-UKS-B3LYP	18.4	-2	30.3	-2	17.5	0	3.24	-5
ZORA-cGKS-B3LYP	18.4	-1	31.1	1	17.8	2	3.67	8
RECP-UKS-PBE0	18.4	-1	30.2	-2	17.6	1	3.43	1
ZORA-cGKS-PBE0	18.4	-1	31.1	1	17.4	0	3.40	0
RECP-UCCSD(T)	18.0	-4	29.6	-4	17.4	-0	3.40	0
RECP-UCCSD(T)+SOC	18.0	-3	30.5	-1	17.4	-0	3.40	0
Literature	18.7^{a}	_	$30.9^{\rm \ a}$	_	17.4 $^{\rm b}$	_	$3.40^{\ b}$	_

^a Scalar relativistic effective core potential calculations at the level of complete active space self-consistent field (RECP-CASSCF) with a correction for spin-orbit coupling by comparison to multi-configuration Dirac-Hartree-Fock (MCDHF) calculations.[21]

^b Experimental data. $E_{\rm i}$ (F) /eV was measured in Ref.22 and $E_{\rm ea}$ (F) /eV was measured in Ref.82.

TABLE III. Dissociation energies for PaF⁴⁺ given for the two most probable dissociation channels as $D_{\rm e}({\rm channel})$.

$D_{\rm e}({\rm Pa^{4+}}{+}{\rm F})/{\rm eV}$	$D_{\rm e}({\rm Pa^{3+}\!+\!F^{+}})/{\rm eV}$
0.6	-11.6
0.1	-13.4
3.5	-9.4
3.2	-10.1
3.4	-9.2
3.3	-10.3
3.7	-8.6
3.6	-9.7
	0.6 0.1 3.5 3.2 3.4 3.3 3.7

TABLE IV. Spectroscopically relevant properties of the eight energetically lowest electronic states of $\mathrm{PaF_3}^+$. Equilibrium bond length r_e , harmonic vibrational wavenumber $\tilde{\omega}_\mathrm{e}$ and excitation wavenumber \tilde{T}_e estimated as vertical excitation energy are shown at the level of Dirac–Coulomb Fock-Space Coupled Cluster (DC-FSCCSD) with two different basis sets and at the level of Zeroth Order Regular Approximation complex Generalized Hartree-Fock (ZORA-cGHF). The projection of the electronic orbital angular momentum quantum number on the molecular axis $\Lambda_{\rm h}$ the squared transition dipole moment $|\vec{\mu}|^2$ and hyperfine coupling constant along the molecular axis A_{\parallel} are given. Hyperfine coupling constants were calculated using μ ($^{231}\mathrm{Pa}$) = 2.01 $\mu_{\rm N}$ and I=3/2 [84]. The DC-FSCCSD/dyall.ae4z results were computed at the equilibrium bond length taken from the ZORA-cGKS-B3LYP calculations.

State	Method	$r_{\mathrm{e}}/\mathrm{\mathring{A}}$	$\tilde{\omega}_{\rm e}/{\rm cm}^{-1}$	$\tilde{T}_{\rm e}/{\rm cm}^{-1}$	Λ	$ \vec{\mu} ^2 / (e^2 a_0{}^2)$	$A_{\parallel}/\mathrm{MHz}$
	ZORA-cGHF	1.87	846	_	2.9		-983
	ZORA-cGKS-B3LYP	1.89	828		2.8		-1020
(X)5/2	ZORA-cGKS-PBE0	1.87	828		2.8		-1020
	DC-FSCCSD/dyall.ae3z	1.85	859	_			
	DC-FSCCSD/ANO-RCC	1.85	892	_			
	ZORA-cGHF	1.87	829	1250	1.8	3×10^{-3}	-953
	ZORA-cGKS-B3LYP	1.88	816	152	1.9	4×10^{-3}	-1140
(1)3/2	ZORA-cGKS-PBE0	1.86	815	185	1.9	4×10^{-3}	-1130
	DC-FSCCSD/dyall.ae3z	1.85	844	658			
	DC-FSCCSD/dyall.ae4z			752			
	DC-FSCCSD/ANO-RCC	1.84	876	646			
	ZORA-cGHF	1.87	840	3020	0.6	6×10^{-8}	-1050
	ZORA-cGKS-B3LYP	1.89	831	3470	0.8	2×10^{-9}	-1600
(1)1/2	ZORA-cGKS-PBE0	1.87	830	3540	0.8	1×10^{-11}	-1560
	DC-FSCCSD/dyall.ae3z	1.86	852	3060			
	DC-FSCCSD/dyall.ae4z			3050			
	DC-FSCCSD/ANO-RCC	1.85	885	2990			
	ZORA-cGHF	1.87	849	5520	3.0	2×10^{-4}	-328
(1)7/2	DC-FSCCSD/dyall.ae3z	1.85	862	5540			
	DC-FSCCSD/dyall.ae4z			5550			
	DC-FSCCSD/ANO-RCC	1.85	895	5550			
	ZORA-cGHF	1.87	831	6440	2.1	2×10^{-3}	-329
(1)5/2	DC-FSCCSD/dyall.ae3z	1.84	846	5790			
` ' '	DC-FSCCSD/dyall.ae4z			5680			
	DC-FSCCSD/ANO-RCC	1.84	878	5680			
	ZORA-cGHF	1.87	838	8000	1.2	5×10^{-4}	-329
(2)3/2	DC-FSCCSD/dyall.ae3z	1.85	853	7810			
	DC-FSCCSD/dyall.ae4z			7830			
	DC-FSCCSD/ANO-RCC	1.85	885	7780			
	ZORA-cGHF	1.87	841	8890	0.4	1×10^{-11}	-473
(2)1/2	DC-FSCCSD/dyall.ae3z	1.86	857	8940			
	DC-FSCCSD/dyall.ae4z			9040			
	DC-FSCCSD/ANO-RCC	1.85	890	8960			
	ZORA-cGHF	1.84	888	30 000	2.0	0.5	-2600
(3)3/2	DC-FSCCSD-dyall-ae3z	1.82	882	29200			
	DC-FSCCSD/dyall.ae4z			29900			
	DC-FSCCSD-ANO-RCC	1.82	901	28900			

TABLE V. The transition properties relevant for investigating laser-cooling are shown for the transitions between the eight lowest states of PaF^{3+} . Excitation energies T_e , Einstein coefficients A for spontaneous emission from electronic states (as detailed in the methods section) and Franck-Condon factors for the 0-0 transition $(f^{(0)})$ and cumulated for the 1-0 transition $(f^{(1)})$ are computed on the level of ZORA-cGHF. The life-time of electronic states τ_e is estimated from the electronic Einstein coefficients as $(\sum_i^a A_i^a)^{-1}$ for each state i, where A_i^a is the Einstein coefficient for the electronic spontaneous emission from i to a. Franck-Condon factors in parantheses indicate that the corresponding vibrational level lies above the vibrational ground state of the next higher electronic state.

	(1)3/2	(1)1/2	(1)7/2	(1)5/2	(2)3/2	(2)1/2	(3)3/2
$ au_e/\mathrm{s}$	7×10^{-2}	6×10^{-3}	2×10^{-2}	7×10^{-4}	4×10^{-4}	6×10^{-6}	2×10^{-8}
T_e/cm^{-1}	1300	3000	5500	6400	8000	8800	30000
T_e/nm	7700	3300	1800	1600	1300	1100	300
$(X)5/2 A/s^{-1}$	1×10^{1}	3×10^{-3}	5×10^{1}	1×10^{3}	5×10^{2}	2×10^{-5}	3×10^{7}
$f^{(0)}$	0.9855	0.9856	0.9994	0.9844	0.9989	0.9999	0.7391
$f^{(1)}$	(0.9999)	0.9999	0.9999	0.9999	0.9999	0.9999	0.9570
T_e/cm^{-1}		1800	4300	5200	6700	7500	29000
T_e/nm		5500	2300	1900	1500	1300	300
$(1)3/2 A/s^{-1}$		2×10^2	2×10^{-5}	3×10^{2}	2×10^{3}	1×10^5	1×10^{7}
$f^{(0)}$		0.9857	0.9795	0.9999	0.9922	0.9843	0.8342
$f^{(1)}$		0.9997	0.9994	0.9999	0.9999	0.9997	0.9798
T_e/cm^{-1}			2500	3400	5000	5800	27000
T_e/nm (1)1/2 A/s^{-1}			$4000 \\ 5 \times 10^{-7}$	2900	2000	1700	370 1×10^{6}
$f^{(0)}$				3×10^{-4}	1	1×10^4	
$f^{(1)}$			0.9994	0.9846	0.9989	0.9969	0.7399
$\frac{J}{T_e/\text{cm}^{-1}}$			0.9999	0.9999	0.9999	0.9999	0.9563
T_e/cm T_e/nm				920 10900	2500 4000	3300 3030	24000 400
$(1)7/2 A/s^{-1}$				3	3×10^{-3}	2×10^{-6}	9×10^{-2}
$f^{(0)}$				0.9782	0.9969	0.9997	0.7194
$f^{(1)}$				(0.9999)	0.9999	0.9999	0.9509
$\frac{J}{T_e/\text{cm}^{-1}}$				(0.5555)	1600	2400	24000
T_e/nm					6300	4200	400
$(1)5/2 A/s^{-1}$					1×10^{2}	2×10^{-6}	1×10^{6}
$f^{(0)}$					0.9914	0.9831	0.8374
$f^{(1)}$					(0.9999)	0.9997	0.9806
T_e/cm^{-1}					, , ,	820	22000
T_e/nm						12200	500
$(2)3/2 A/s^{-1}$						1×10^{1}	3×10^{5}
$f^{(0)}$						(0.9986)	0.7768
$f^{(1)}$						(0.9999)	0.9670
T_e/cm^{-1}							21000
T_e/nm							500
$(2)1/2 A/s^{-1}$							1×10^3
$f_{(1)}^{(0)}$							0.7353
$f^{(1)}$							0.9550

TABLE VI. \mathcal{P} , \mathcal{T} -odd electronic structure parameters for PaF³⁺ are shown. The properties were calculated on the level of ZORA-cGHF and ZORA-cGKS-B3LYP and PBE0. For computation of the magnetic interaction with the proton EDM $W_{\rm m}$ μ (231 Pa) $\simeq 2.01\mu_{\rm N}$ [84] and I=3/2 were used.

State	Method	$W_{\mathcal{S}}/\frac{e}{4\pi\epsilon_0 a_0^4}$	$W_{ m m}/rac{10^{18}\ h{ m Hz}}{e{ m cm}}$	$W_{ m T}/(h{ m Hz})$	$W_{ m d}/rac{10^{24}h{ m Hz}}{e{ m cm}}$	$W_{ m s}/(h{ m kHz})$	$W_{\mathcal{M}}/\frac{10^{33}h\mathrm{Hz}}{ce\mathrm{cm}^2}$
	ZORA-cGHF	-72000	6.3	-6700	0.66	4.2	0.038
X5/2	ZORA-cGKS-B3LYP	-58000	4.9	-5300	0.41	2.5	0.023
	ZORA-cGKS-PBE0	-58000	4.9	-5400	0.41	2.9	0.027
	ZORA-cGHF	-66000	5.5	-6200	1.4	9.3	-0.022
(1)3/2	2 ZORA-cGKS-B3LYP	-53000	4.5	-4900	0.92	5.8	-0.0023
	ZORA-cGKS-PBE0	-54000	4.4	-5000	1.1	6.8	-0.0018
	ZORA-cGHF	-59000	4.8	-5600	2.9	20	-0.10
(1)1/2	2 ZORA-cGKS-B3LYP	-49000	4.1	-4800	1.8	12	-0.13
	ZORA-cGKS-PBE0	-49000	3.9	-4600	2.0	13	-0.14
(1)7/2	2 ZORA-cGHF	-73000	6.4	-6800	-0.48	-3.0	-0.030
(1)5/2	2 ZORA-cGHF	-68000	5.7	-6400	-0.28	-1.9	-0.0014
(2)3/2	2 ZORA-cGHF	-61000	5.1	-5500	-0.22	-1.6	0.061
(2)1/2	2 ZORA-cGHF	-58000	4.9	-5500	-0.36	-2.6	0.14
(3)3/2	2 ZORA-cGHF	-70000	5.8	-6600	3.2	19	0.14