

Original paper

Metamorphic history of skarns, origin of their protolith and implications for genetic interpretation; an example from three units of the Bohemian Massif

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Skarns in the Svratka Unit, in the neighbouring part of the Moldanubian Zone and in the Kutná Hora Complex were studied with respect to their metamorphic evolution, major- and trace-element geochemistry, oxygen isotopic composition and zircon ages. Skarns form competent lenses and layers in metamorphosed siliciclastic rocks and preserve some early deformation structures and several equilibrium assemblages representing the products of successive metamorphic reactions. The main rock-forming minerals, garnet and clinopyroxene, are accompanied by less abundant magnetite, amphibole, plagioclase, epidote ± quartz. In the **Svratka Unit** the early prograde M₁, prograde/peak M₂, and retrograde M₃ metamorphic stages have been distinguished. Metamorphic conditions in skarns of the **Moldanubian Zone** are limited to a relatively narrow interval of amphibolite facies. The prograde and retrograde events in the **Kutná Hora Complex** skarns probably took place under amphibolite-facies conditions. The presence of magnetite and the increasing proportion of the andradite component in the garnet indicate locally increased oxygen fugacity.

Skarn geochemistry does not show systematic differences in the skarn composition among the three units. The regional variations are exceeded by differences among samples from individual localities. The Al₂O₃/TiO₂, Al₂O₃/Zr, TiO₂/Nb ratios point to the variable proportion of the detrital material, combined in skarn protoliths with CaO and FeO, the major non-detrital components. The skarns exhibit elevated abundances of Cu, Zn, Sn and As. The Eu/Eu* ratio varies in the range of 0.5–8.6, the total REE contents vary from 8 to 345 ppm. The lowest ΣREE values (< 100 ppm) occur in skarns with magnetite mineralization. The wide intervals of ΣREE and Eu/Eu* values are interpreted to indicate variations in the temperature and redox conditions among layers of the same locality and at various localities.

The oxygen isotope compositions of garnets, pyroxenes and amphiboles from skarns of the Svratka Unit exhibit a range of δ¹⁸O = 0.1 to 4.1 ‰. *In situ* (laser-ablation ICP-MS) U-Pb dating of zircon from one of the Svratka Unit skarn bodies yielded a wide range of ages (0.5–2.6 Ga), supporting the detrital origin of this zircon population.

The skarn protoliths were probably rocks of mixed detrital-exhalative origin deposited on the sea floor.

The geological position of skarns, with their structural and metamorphic record, probably reflect tectono-metamorphic evolution shared with that of their host rocks. The geochemical characteristics, including oxygen isotopic compositions and the presence of detrital zircons with a wide range of ages exclude metasomatic, and point to a sedimentary-exhalative mode of origin for the studied skarns.

Keywords: skarn, Bohemian Massif, petrology, geochemistry, oxygen isotopes, detrital zircon age

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1. Introduction

Detailed modern contributions dealing with the nature and genesis of skarns were published by many geologists during the past few decades from different geological environments all over the world, for instance Jamtveit et al. (1993), Nicolescu et al. (1998), Meinert et al. (2003) or Gaspar et al. (2008).

The review of world skarn deposits has been presented by Einaudi et al. (1981), Burt (1982), Meinert (1998) and

Meinert et al. (2005). These publications and reviews are devoted to skarns with mostly contact metamorphic and metasomatic history accompanied by igneous-related hydrothermal input. Several authors have studied the mineralogy/petrology and genetic aspects of various skarn bodies in the Bohemian Massif, partly with emphasis on the pre-metamorphic history and open- vs. closed-system conditions of formation.

Formation under the open-system conditions implies a metasomatic mode of formation (i.e., metasomatism

of carbonate rocks by granite-related fluids). This category includes several skarns in the Moldanubian Zone (Koutek 1950; Němec 1991; Houzar and Šrein 1995; Litochleb et al. 1997; Žáček et al. 1997, 2003), in the Krušné hory/Erzgebirge area (Němec 1991; Šrein and Šreinová 2000). Rötztler and Mingram (1998) reported a skarn formation via Ca-metasomatism of basic magmatic rocks (rodingites).

Skarn formation in a closed system, due to regional metamorphism of protoliths with suitable composition, was assumed in the Moldanubian Zone by Zoubek (1946), Vrána (1987), Potužák (1996), Pertold et al. (1997, 2000), Drahotka et al. (2005) and in the Krušné hory/Erzgebirge area by Klomínský and Sattran (1963) with Kotková (1991), in the Lugaicum by Pertold and Pouba (1982), and in the Svratka Unit by Pertoldová (1986) and Pertoldová et al. (1998).

The aim of this paper is to address the widely discussed problem of the origin of the Bohemian skarns and their pre-metamorphic precursors. The study includes skarns from the Varied Group, Monotonous Group and the Gföhl Unit of the Moldanubian Zone, together with the Kutná Hora Complex and Svratka Unit. Altogether, 34 localities of skarn rocks were studied, with their basic characteristics and locations presented in Tab. 1. The paper deals with the geological position, metamorphic

history and whole-rock geochemistry. The oxygen isotope composition of silicate minerals was studied in selected samples from the Svratka Unit. Zircon populations in skarn from the Svratouch locality were dated. The geological positions of the sampled skarn localities are depicted in Fig. 1. Using this integrated data set, it will be demonstrated that the studied skarns have formed by a closed-system metamorphism of sedimentary-exhalative rocks deposited on the sea floor.

2. Geological setting

The Variscan Orogeny (subduction of the "Tethys-type" oceanic crust followed by continental collision of Gondwana and Laurussia; ~380–290 Ma ago) was of major importance for amalgamation and evolution of basement units in the Bohemian Massif (for a review, see Franke 1989, 2000; Tait et al. 2000). In the north-eastern part of the Bohemian Massif, this orogeny involved tectono-metamorphic processes including juxtaposition and stacking of contrasting segments of the continental crust. At a somewhat later stage, the eastern parts of these amalgamated units were overthrust on Neoproterozoic crustal segment of the Brunovistulicum. On a regional scale, the studied area includes two super-units (Fig. 1) –

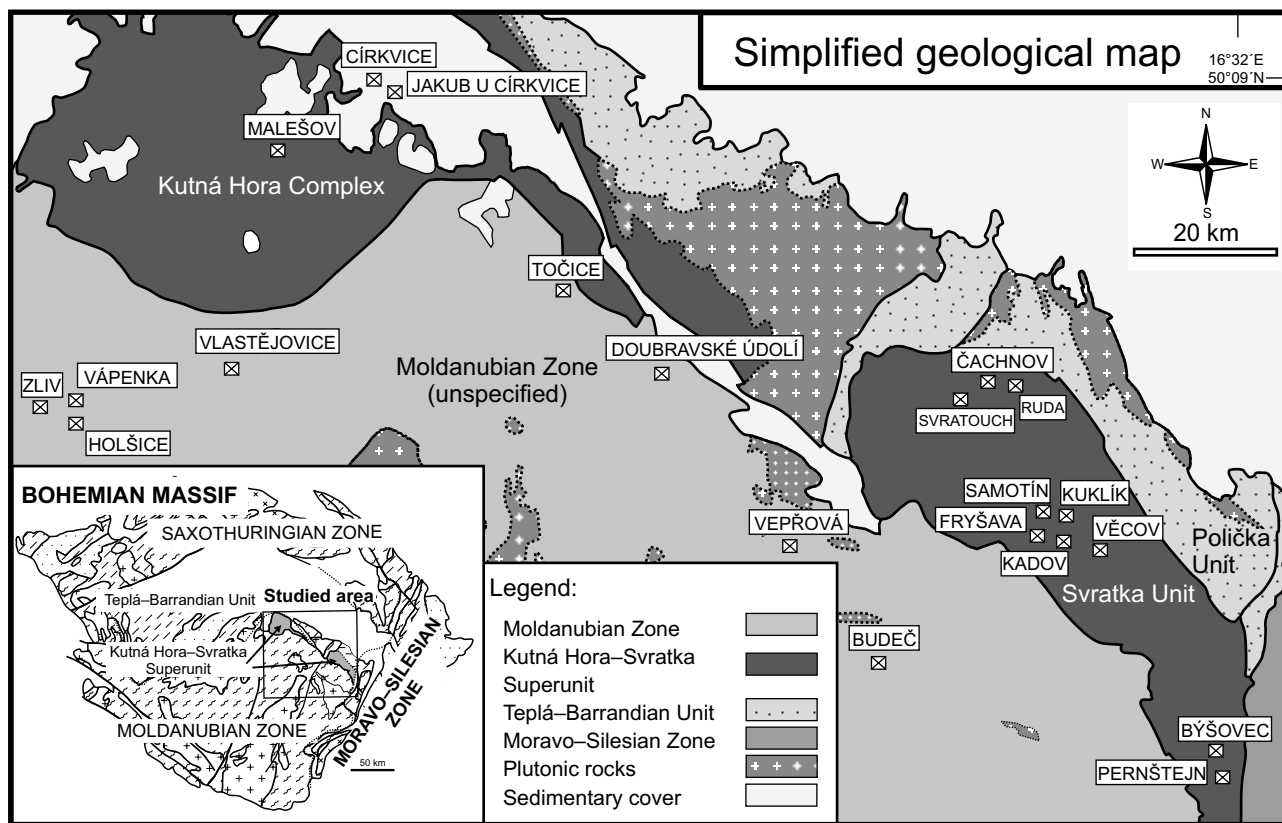


Fig. 1 Simplified geological map of the NE margin of the Moldanubian Zone, showing skarn localities. Modified after Cháb et al. (2007). Inset: Sketch of Bohemian massif modified after Franke et al. (2000).

Tab. 1 List of analyzed skarn samples from the studied units

Geological setting	Sample No.	Sample	Locality	Prevailing mineralogy	Microprobe analysis	WR geochem. analysis	Dating	Oxygen analysis	WGS-84-N	WGS-84-E	
Moldanubian Zone	1	KM8A	Budeč	Grt, Cpx	×	×			49 32.25	15 54.97	
	2	KM8B	Budeč	Mag, Cpx, Grt	×	×			49 32.25	15 54.97	
	3	KM8C	Budeč	Grt, Cpx	×				49 32.25	15 54.97	
	4	KM8FV1	Budeč	Mag, Cpx, Grt		×			49 32.25	15 54.97	
	5	KM8FV2	Budeč	Cpx		×			49 32.25	15 54.97	
	6	KM3F	Vepřová	Grt, Cpx		×			49 37.86	15 50.18	
	7	KM3H	Vepřová	Qtz, Cpx			×		49 37.86	15 50.18	
	8	KM3CH	Vepřová	Grt, Cpx			×		49 37.86	15 50.18	
	9	KM3A	Vepřová	Grt, Cpx	×				49 37.86	15 50.18	
	10	KM3C	Vepřová	Grt, Cpx	×				49 37.86	15 50.18	
	11	KM3E	Vepřová	Grt, Cpx	×				49 37.86	15 50.18	
	12	KM3M	Vepřová	Grt, Cpx	×				49 37.86	15 50.18	
	13	KM3FV	Vepřová	Grt, Cpx			×		49 02.45	16 09.16	
	14	KMRES1	Rešice	Grt, Cpx			×		49 02.45	16 09.16	
	15	KMRES2	Rešice	Cpx, Grt			×		49 02.45	16 09.16	
	16	KMRES3	Rešice	Grt, Cpx			×		49 02.45	16 09.16	
	17	KMRES5	Rešice	Cpx			×		49 02.45	16 09.16	
	18	KMSLA1	Slatina	Grt			×		49 01.25	16 01.31	
	19	KMSLA2	Slatina	Cpx, Grt			×		49 01.25	16 01.31	
	20	KMVLAFV1	Vlastějovice	Grt, Cpx			×		49 44.26	15 11.09	
	21	KMVLAFV2	Vlastějovice	Cpx, Grt, Pl			×		49 44.26	15 11.09	
	22	KMHOLFV1	Holšice	Grt, Cpx			×		49 45.58	15 03.03	
	23	KMZLIFV1	Zliv	Cpx			×		49 46.01	15 02.34	
	24	KMZLIFV2	Zliv	Cpx, Grt			×		49 46.01	15 02.34	
	25	KMVAPFV2	Vápenka	Grt, Hbl			×		49 46.11	15 02.41	
	26	KMZUPFV1	Županovice	Grt, Cpx			×		48 56.59	15 30.06	
	27	KMZUPFV2	Županovice	Cpx			×		48 56.59	15 30.06	
Svratka Unit	28	KM5d	Fryšava	Grt, Hbl		×			49 37.70	16 02.77	
	29	KM5A	Fryšava	Grt, Cpx		×			49 37.70	16 02.77	
	30	KM6A	Kadov	Grt, Cpx		×			49 39.68	16 05.15	
	31	SV1	Svratouch	Grt, Cpx	×				49 44.32	16 01.66	
	32	SV27	Svratouch	Cpx, Grt			×		49 44.32	16 01.66	
	33	SV262	Svratouch	Cpx, Grt, Gru	×		×	×	49 44.32	16 01.66	
	34	SV25	Svratouch	Cpx, Grt	×		×	×	49 44.32	16 01.66	
	35	MSDC114	Teplá	Grt, Cpx			×	×	49 38.66	16 05.09	
	36	MSDC130	Blatiny	Grt, Cpx			×	×	49 38.93	16 05.32	
	37	MSDC128	Samotín	Grt, Cpx			×	×	49 38.83	16 04.91	
	38	KMSB6	Adamova skála	Cpx, Grt			×		49 37.38	16 09.20	
	39	KMSB7	Malá skála	Grt, Cpx			×		49 37.45	16 08.47	
	40	KMSB15	Líšná	Cpx, Grt			×		49 38.44	16 08.41	
	41	KMKUKFV1	Kuklík	Cpx, Grt			×		49 37.80	16 06.81	
	42	KMKUKFV2	Kuklík	Cpx			×		49 37.80	16 06.81	
	43	KMKUKFV3	Kuklík	Cpx			×		49 37.80	16 06.81	
	44	KM7	Kuklík	Grt, Cpx	×		×		49 38.14	16 06.20	
	45	KM7FV	Krátká	Grt, Cpx			×		49 38.14	16 06.20	
	46	KMSB21	Věcov	Grt, Cpx	×		×		49 37.12	16 09.30	
	47	KMLISFV	Líšná	Grt, Cpx			×		49 38.44	16 08.41	
	48	VR1	Čachnov	Cpx, Grt			×		49 45.09	16 02.85	
	49	Cach2	Čachnov	Grt	×		×		49 45.09	16 02.85	
	50	VR10B	Čachnov	Grt, Hbl, Cpx			×		49 45.09	16 02.85	
	51	VR39	Ruda u Čachnova	Grt, Hbl			×		49 44.55	16 04.69	
	52	VR60	Ruda u Čachnova	Grt, Hbl			×		49 44.55	16 04.69	
	53	VR69	Ruda u Čachnova	Grt, Cpx			×		49 44.55	16 04.69	
	54	KM11	Býšovec	Grt, Cpx			×		49 28.43	16 17.67	
	55	KMPERFV1	Pernštejn	Grt			×		49 26.98	16 18.33	
	56	KMPERFV2	Pernštejn	Grt			×		49 26.98	16 18.33	
	57	KMPERFV3	Pernštejn	Grt, Cpx			×		49 26.98	16 18.33	
	Kutná Hora Complex	58	KMMALFV1	Malešov	Cpx, Grt, Mag		×			49 55.52	15 13.56
		59	KMMALFV2	Malešov	Grt, Cpx, Mag		×			49 55.52	15 13.56
60		KMVSM1	Malešov	Cpx, Grt			×		49 55.52	15 13.56	
61		KMVSM2	Malešov	Grt, Cpx	×		×		49 55.52	15 13.56	
62		KMVSM4	Malešov	Grt, Cpx	×		×		49 55.52	15 13.56	
63		KMVSM5	Malešov	Cpx, Grt, Hbl	×		×		49 55.52	15 13.56	
64		KMVSM7	Malešov	Grt, Cpx, Ep	×		×		49 55.52	15 13.56	
65		KMCIRFV1	Čírkvice	Grt, Cpx			×		49 56.83	15 20.32	
66		KMCIRFV2	Čírkvice	Grt, Cpx			×		49 56.83	15 20.32	
67		KMV20	Jakub u Čírkvice	Grt, Cpx	×		×		49 57.16	15 20.29	
68		KMTOCFV	Točice	Grt, Cpx			×		49 48.15	15 33.83	
69		BV1	Točice	Grt, Cpx			×		49 48.15	15 33.83	

WGS-84 = World Geodetic System 1984

the Moldanubian Zone (MZ) and the Kutná Hora-Svratka Superunit (KHS). A detailed review of the geology and structure of the basement units along the north-eastern margin of the Moldanubian Zone is presented by Verner et al. (this volume).

2.1. Svratka Unit

The Svratka Unit (SU) is composed of migmatites, metagranites, orthogneisses and metasedimentary rocks with layers of calc-silicate gneisses and amphibolites. The Variscan tectonometamorphic conditions were estimated at ~0.4–0.8 GPa and ~650–670 °C (Pitra and Guiraud 1996; Buriánek 2008; Buriánek and Čopjaková 2008). The rocks of the SU have retained some relics of Cambro–Ordovician structures; the SU belongs to the mid-crustal level of the Variscan orogen (Schulmann et al. 2005; Verner et al. this volume).

The dominant planar fabric in this unit is NW–SE trending, except for the western termination, which is a large-scale antiformal closure. Foliation planes with predominating moderate dips contain subhorizontal lineation defined by elongated mineral aggregates (Fig. 2a). Relics of older deformation structures are locally preserved in more competent rocks, particularly in calc-silicate gneisses and skarns. The skarn bodies are moderately to strongly elongated, with dimensions of ~0.1–0.3 km × 0.5–1.5 km, often oriented parallel with the structures in the host rocks.

The dominating structure in skarns is compositional layering, usually consisting in alternating garnet-rich and pyroxene-rich bands. This planar fabric is typically oriented parallel with the foliation in the enclosing rocks. Relics of older structures (isoclinal and rootless folds) show steeply dipping foliation trending WNW–ESE, especially in limbs of asymmetric folds (Fig. 2a). Minor structural elements include superimposed anastomosing cleavage domains and shear zones. The geological positions of skarn bodies in SU are documented in 1: 25 000 maps (Hanžl ed. 2008; Melichar ed. 2008).

2.2. North-eastern part of the Moldanubian Zone

The Moldanubian Zone (MZ) represents exhumed lower- to mid-crustal rocks that recorded a polyphase Variscan tectonometamorphic history (for a general review see Urban and Synek 1995; Schulmann et al. 2005, 2009). The structural development was related to stacking of individual lithotectonic units at ~350–340 Ma, followed by HT-LP metamorphism and anatexis at ~340–330 Ma and subsequent (Late Carboniferous to Permian) wrench

tectonics (Edel et al. 2003). The northern and north-eastern parts of the MZ, locally termed the Strážek Unit, include migmatized paragneisses and migmatites with abundant layers of amphibolite, calc-silicate gneiss, quartzite and relatively small bodies of HP/HT granulite and peridotite/lherzolite. The metamorphic conditions of granulites were estimated by Tajčmanová et al. (2006) at ~1.8 GPa and ~850 °C for the high-pressure event, followed by low-pressure re-equilibration at 0.4–0.6 GPa and 680–720 °C.

The structural patterns in the northern and north-eastern parts of the MZ and metamorphic conditions show notable variation due to complicated superposition of structures. Along the northern margin of the MZ, foliations are parallel with the border of the unit and exhibit a shallow to moderate dip to the north or northwest. Skarn bodies in the northern part of the MZ are hosted by migmatites and migmatized paragneisses. They form asymmetric bodies with longer axis of ~50–500 m and internal structures that are usually concordant with the regional structures of the enclosing paragneisses (Drahota et al. 2005). The geological situation of the skarn bodies in the northern part of the MZ is presented in the Czech Geological Survey maps 1:25 000 (Hrdličková ed. 2008; Hanžl and Hrdličková eds 2009; Rejchrt ed. 2009).

2.3. Kutná Hora Complex

The Kutná Hora Complex (KHC) is a heterogeneous assemblage of lower- to mid-crustal segments at the northern border of the MZ, comprising orthogneisses, migmatites, metasediments and ultramafic upper mantle rocks (Synek and Oliveriová 1993). The metamorphic evolution of the upper part of the KHC (Běstvína, Malín and Plaňany Unit) corresponds to the HP/HT–HP/MT events (Vrána et al. 2005) followed by HP/MT tectonometamorphic overprint. The lower part of the KHC is composed of the Kouřim Nappe (fine-grained leucocratic migmatites and orthogneisses) and further, in the marginal part, of the Micaschist Unit (metapelites). Rocks of these two units recorded a Variscan MP-MT metamorphic event (Nahodilová et al. 2006).

The tectonic structure of the KHC is interpreted as resulting from polyphase deformation and the probable effects of shallow-dipping thrusting, similar to nappe structures. The early structures include N–S trending foliation locally preserved in orthogneisses and migmatites. They are usually strongly overprinted by younger foliation with shallow to moderate dip to the NNW to ENE. Review of structural information is presented by Synek and Oliveriová (1993) and Nahodilová et al. (2005). The Malešov skarn body (Figs 1, 2b) is located in the structurally highest part of the KHC, the Malín

Unit, composed of polymetamorphic migmatites and two-mica kyanite migmatites with boudin-like bodies of upper mantle garnet lherzolites. Metamorphic foliation in the rocks of the Malín Unit is folded to open, moderately asymmetric folds but it is generally dipping at moderate to low angles to ~NNW. The Malešov skarn body (~200 × 1 000 m) forms an asymmetric antiformal structure with limbs dipping at moderate to low angles to WNW and NE (Fig. 2b). The internal structure of skarn is defined by alternating irregular bands rich in garnet or pyroxene. These planar structures dip steeply to the NNW to W. They are thus oriented discordantly in relation to the metamorphic planar fabrics predominating in the enclosing rocks.

3. Analytical methods

Mineral analyses were carried out with a Cameca Camebax SX-100 in the Joint Laboratory of Electron Microscopy and Microanalysis of Masaryk University and the Czech Geological Survey in Brno. Operating conditions were 15 kV accelerating voltage and 30 nA beam current.

The beam was focused to 3 μm ; the peak counting times were 20 seconds. The analyses were recalculated to chemical formulae or end member proportions using the MinPet software v. 2.02 of Richard (1995).

Whole-rock major- and trace-element concentrations were determined at the Central Laboratories of the Czech Geological Survey (CGS), Prague, or Acme Analytical Laboratories, Ltd., Vancouver. Major oxides were determined by FAAS, ICP-OES or titration methods (ČGS) and the ICP-MS method (Acme). The REE and other trace elements were analyzed by the ICP-MS, ICP-OES and XRF methods. Geochemical data were processed using the GCDkit software package (Janoušek et al. 2006).

The calculations of P-T conditions for mica schists were performed using THERMOCALC 3.25 (Powell et al. 1998; 2005 upgrade) and the internally consistent thermodynamic dataset 5.5 (Holland and Powell 1998; November 2003 upgrade). Mixing models for most solid solutions were taken from White et al. (2001) and the THERMOCALC documentation (Powell and Holland 2004).

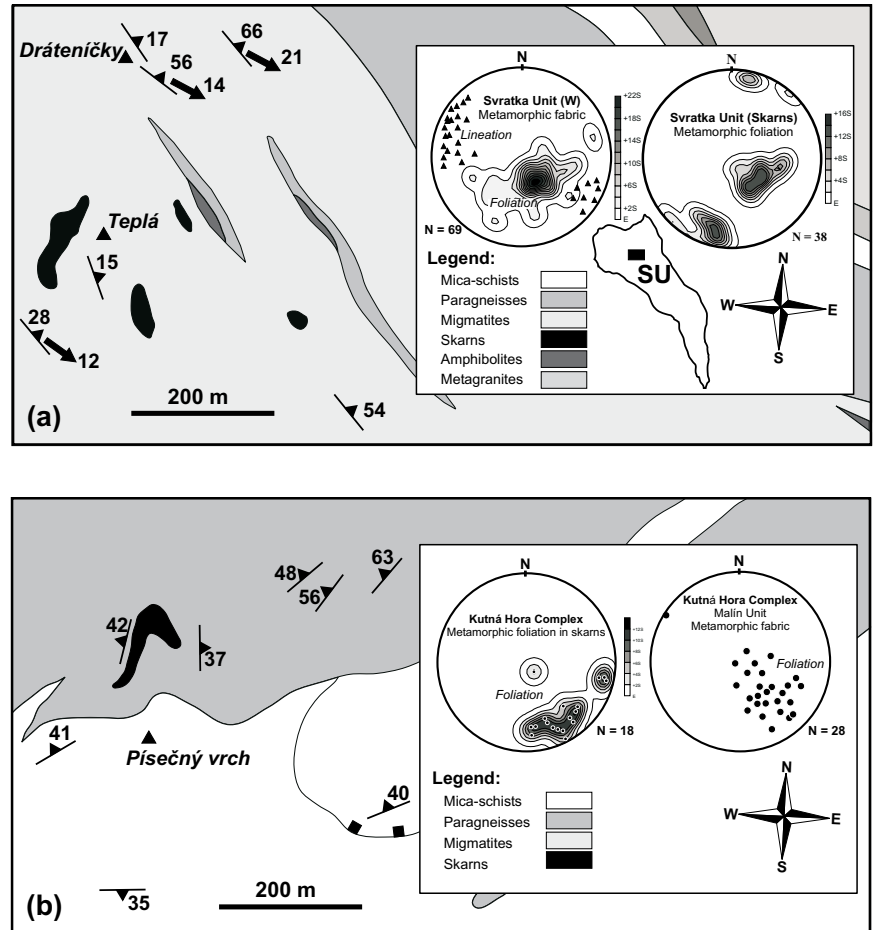


Fig. 2 Schematic structural maps of the two selected skarn bodies: a – Samotín (western part of the Svatka Unit); b – Malešov (SE part of the Kutná Hora Complex).

Plagioclase was formulated using the model of Holland and Powell (2003), the paragonite–muscovite solution is after Coggon and Holland (2002) and the paragonite–margarite solution as in Štípská et al. (2006).

Oxygen for isotopic analyses was liberated from mineral separates by a fluorination technique using BrF_5 , similar to that of Clayton and Mayeda (1963), at the Institute of Geochemistry, Mineralogy and Mineral Resources, Charles University in Prague. During this procedure, oxygen is cryogenically cleaned to remove the reaction products and excess fluorination agent; finally it is converted to CO_2 on a heated graphite rod. Isotopic measurements were performed using the Finnigan MAT 251 mass spectrometer at the Czech Geological Survey, Prague. The overall analytical uncertainty of the $\delta^{18}\text{O}$ values tested by repeated analyses of NBS 28 standard was 0.2 ‰ (SMOW).

Zircon grains were extracted from rock samples using conventional crushing, heavy liquids and magnetic separation. The grains were mounted in 1 inch epoxy-filled blocks and polished to obtain even surfaces suitable for

laser ablation ICP-MS analyses. Prior to analysis, the carbon coating was removed and the sample surfaces were cleaned in deionised water and ethanol. Isotopic analysis of zircons by laser ablation ICP-MS followed the technique described in Košler et al. (2002) with Košler and Sylvester (2003). A Thermo-Finnigan Element 2 sector field ICP-MS coupled to a 213 NdYAG laser (New Wave UP-213) at Bergen University was used to measure the Pb/U and Pb isotopic ratios in zircons. The raw data were corrected for the dead time of the electron multiplier, processed off line in the Lamdate spreadsheet-based program (Košler et al. 2002) and plotted on concordia diagrams using Isoplot (Ludwig 1999). Data reduction included correction for the gas blank, laser-induced elemental fractionation of Pb and U and instrument mass bias. No common Pb correction was applied to the data. Zircon reference materials 91500 (1065 Ma – Wiedenbeck et al. 1995) and Plešovice (338 Ma – Sláma et al. 2008) were periodically analysed during this study.

4. Petrography, mineral chemistry and metamorphic evolution

Petrological description was done for the following samples: the Svatka Unit – SV1, SV262, SV25, KM7, KMSB21, Cach2; the Moldanubian Zone – KM8A, KM8B, KM8C, KM3A, KM3C, KM3E, KM3M, the Kutná Hora Complex – KMVSM1–7, KMV20.

The summary of petrological results is presented in Tab. 2.

4.1. Svatka Unit

Skarns exhibit banded or massive structure with nematoblastic or nemato-granoblastic texture.

The following skarn types were recognised under the microscope: garnet-pyroxene skarn, garnet-amphibole skarn, pyroxene skarn with garnet, grunerite-pyroxene skarn with garnet, and pyroxene-garnet skarn grading to

Tab. 2 The summary of petrological results for skarns from the Svatka Unit, the Moldanubian Zone and the Kutná Hora Complex

unit	skarn type	texture	structure	minerals				note
				rock-forming	minor	acces.	secondary	
Svatka Unit	Grt-Cpx, Cpx-Grt, garnierite, Cpx skarn, Grt-Hbl, Gru-Cpx	nematoblastic, nematogranoblastic	banded, massive	Grt (Grs ₂₁₋₄₄ Alm ₂₅₋₇₈ Sps ₀₋₂₃ Adr ₀₋₁₂ Prp _{0.5-7}), Cpx	Qtz, Pl, Ep, Mag	Zrn, Spn, Aln, Ap, Ilm, Ccp, Au, Py, native Bi	Hbl, Pl	Garnets show prograde compositional zoning. Decompression textures: Spn→Spn-Pl symplectite, Grt→Hbl-Pl symplectite, Cpx→Cpx-Pl symplectite. The content of jadeite component in Cpx is 0.5–24 mol.%.
Moldanubian Zone	Grt-Cpx, Cpx-Grt, garnierite, Cpx, Mag with Hbl	nematoblastic, nematogranoblastic	banded, massive	older Grt (Grs ₇₄₋₆₇ Alm ₉₋₁₂ Adr ₁₅₋₁₈), Cpx, younger Grt (Grs ₃₄₋₃₅ Alm ₆₋₈ Sps ₄ Adr ₅₃₋₅₄)	Qtz, Pl, Ep, Mag, Czo	Spn, Ap, native Bi, Py, Po	Hbl, Pl	Garnets show weak retrograde zoning. Younger Grt rich in Adr fills fractures in older Grt.
Kutná Hora Complex	Grt-Cpx, Cpx-Grt, garnierite, Cpx with Grt	nematoblastic, nematogranoblastic	banded, massive, schlieren-like	Grt (Grs ₅₀₋₇₅ Alm ₂₄₋₁₅ Sps ₂₋₀ Adr ₃₁₋₉ Prp ₁₋₃), Grt (Grs ₂₄₋₆₅ Alm ₉₋₀ Sps ₄₋₀ Adr ₆₅₋₃₃), Cpx, Ep	Mag, Pl	Spn, Aln, Ap, cerite, hydro grossular	Grt, Ep, Hbl	Two types of garnet. Both are characterized by inverse variation in the Grs and And components. Grt and Cpx show weak prograde zoning. Four generations of Ep and three generations of Cpx. Some Ep are enriched in REE.

(acces.=accessory)

garnetite \pm quartz. Garnet bands are several mm to several dm wide and often exhibit irregular folding, swelling to elongated lenses, or pass to garnet clusters and schlieren. Subordinate minerals, accompanying garnet and clinopyroxene, are quartz, plagioclase, epidote, amphibole, and accessory zircon, titanite, allanite and apatite. Some specimens exhibit reaction textures typical of decompression of eclogites or mafic granulites, such as amphibole-plagioclase symplectite coronas around garnet (Fig. 3a), clinopyroxene replaced by clinopyroxene-plagioclase symplectite (Fig. 3b) and Al-rich titanite transformed to titanite-plagioclase symplectite (Fig. 3c).

The following opaque minerals were identified in skarns: magnetite and ilmenite, locally cobaltite, arsenopyrite, pyrrhotite, chalkopyrite, pyrite, rare marcasite, löllingite, safflorite, safflorite-löllingite, FeCo-sulfarsenide, bismuthinite, native bismuth, gold, Bi-tellurides, and galena. A detailed petrological study of skarn localities near Pernštejn (western Moravia) was published by Pertoldová et al. (1998).

4.1.1. Garnet

Garnet grains enclose clinopyroxene, epidote, oligoclase, and exceptionally quartz. In other samples, garnet is affected by alteration, eventually it is completely replaced by plagioclase, epidote, amphibole and carbonate. Garnet compositions correspond dominantly to the grossular-almandine series ($\text{Grs}_{21-44}\text{Alm}_{25-78}\text{Sps}_{0-23}\text{Adr}_{0-12}\text{Prp}_{0.5-7}$, $X_{\text{Fe}} = 0.95-1.00$ mol. %; Fig. 2a). A prominent compositional zoning is defined by variation in the spessartine, almandine and grossular components (Fig. 5a-b, Tab. 3).

In one particular sample (Cach2), garnet exhibits two distinct compositional zones seen as optical zoning. The darker part corresponds to the crystal core, while the outer zone has a lighter colour (Fig. 3d, Tab. 3) and contains a higher proportion of andradite. These garnets contain more grossular than the almandine component ($\text{Grs}_{45-73}\text{Alm}_{13-23}$; Fig. 4). Representative chemical analyses of garnets from the Svatka Unit are presented in Tab. 3.

4.1.2. Clinopyroxene

Clinopyroxene compositions correspond to the diopside-hedenbergite series (Fig. 6, Tab. 4). The content of the jadeite component is 0.5–24 mol. %. Clinopyroxenes with elevated Jd component (up to 24 mol. %) occur in the matrix of some samples in pyroxene-plagioclase symplectite, or form inclusions in garnet (Fig. 3b).

4.1.3. Calcic amphibole

Amphiboles were frequently formed by replacement of clinopyroxene or garnet, such as in amphibole-

plagioclase symplectite. Some samples contain quartz-amphibole veinlets or a plagioclase-amphibole mosaic, filling domains among large garnet and pyroxene crystals (Fig. 3a). The chemical composition of these amphiboles corresponds to ferro-pargasite, hastingsite and ferro-actinolite (Tab. 5).

4.1.4. Grunerite

The fine-grained matrix consists of subhedral grains of light green clinopyroxene intergrown with paler, prismatic to acicular crystals of grunerite (Fig. 3e, Tab. 5).

4.1.5. Plagioclase

Plagioclase occurs as a minor component in small veinlets with amphibole among grains of garnet and clinopyroxene or it forms symplectite coronas around garnet (Fig. 3a). The chemical composition varies between albite and andesine (An_{2-42} , Tab. 5).

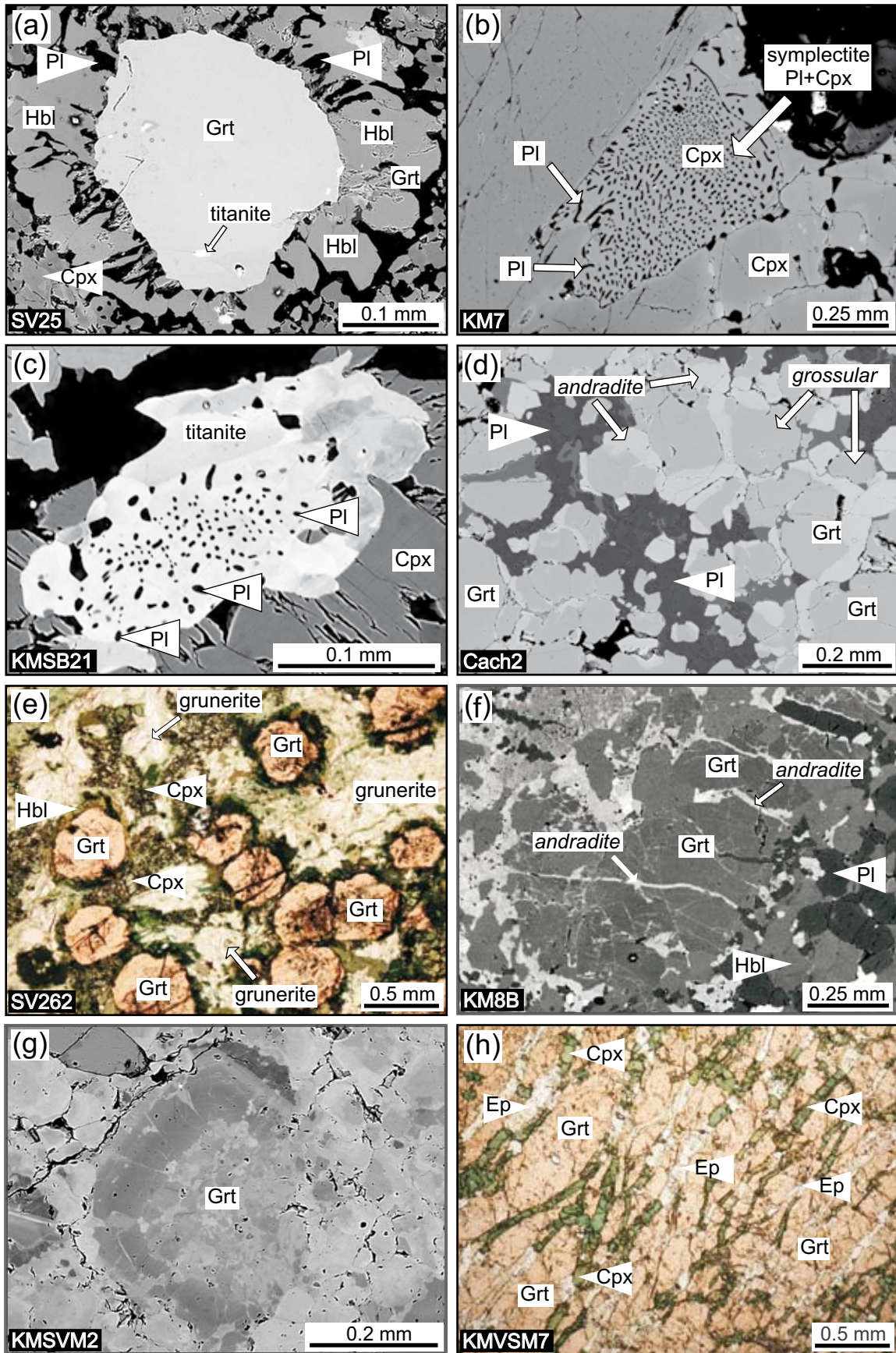
4.1.6. Titanite

Titanite occurs as a common accessory mineral in skarns. Sample KMSB21c contains grains of inhomogeneous titanite, in part unmixed to plagioclase-titanite symplectite (Fig. 3c). The older titanite I shows notable enrichment in Al and F (2 to 6 wt. % Al_2O_3 , 0.7 to 2 wt. % F), while younger titanite II in symplectite contains 1–3 wt. % Al_2O_3 and 0–1 wt. % F (Tab. 7).

4.1.7. Metamorphic development

Garnets exhibit well-defined prograde compositional zoning (Fig. 5a-b). The chemical composition of garnet and variation in the jadeite component in pyroxenes indicate at least three metamorphic episodes, similar to those in the enclosing rocks (see discussion).

The prograde zoning in garnet is defined by a high content of Mn in the crystal cores, corresponding to prograde stage M_1 . The following prograde stage M_2 is one of a high-pressure increase from amphibolite facies to eclogite facies over a temperature interval of *c.* 100 °C. Peak metamorphic conditions are characterized by pressure near 1.4 GPa (Gasparik and Lindsley 1980, see Fig. 13), expressed by jadeite component up to 24 mol. %. The increased Jd content corresponds to equilibration at eclogite-facies conditions. The following stage M_3 is related to exhumation of the Complex and decompression of skarns back to amphibolite-facies conditions. Reaction textures, such as amphibole-plagioclase coronas around garnet (Fig. 3a), disintegration of Jd-rich pyroxene to clinopyroxene-plagioclase symplectite



Tab. 3 Representative chemical analyses of garnets from the Svatka Unit, the Moldanubian Zone and the Kutná Hora Complex

location sample position	Svatka Unit						Moldanubian Zone						Kutná Hora Complex			
	SV262		SV25		Cach2		KM8A		KM8B		KM3M		KMVSM2		KMVSM7	
	core	rim	core	rim	core	rim	core	rim	core	rim	core	rim	core	rim	core	rim
(wt. %)																
SiO ₂	36.26	37.51	36.93	37.12	38.49	37.20	38.02	38.02	38.30	38.29	37.20	38.03	36.26	36.73	37.50	38.43
TiO ₂	0.19	0.05	0.11	0.10	0.25	0.22	0.09	0.18	0.58	0.48	0.28	0.31	0.70	0.00	0.35	0.18
Cr ₂ O ₃	0.02	0.01	0.01	0.02	0.00	0.00	0.01	0.00	0.00	0.00	0.02	0.02	0.00	0.00	0.03	0.01
Al ₂ O ₃	19.64	20.13	20.29	20.88	20.26	14.64	18.79	19.63	17.13	17.08	19.22	19.13	8.83	10.01	15.37	19.53
FeO	22.43	27.32	22.34	30.96	12.06	17.67	16.96	15.22	9.02	9.14	9.13	10.94	21.15	16.40	18.12	13.80
MnO	4.34	2.38	9.79	0.48	0.14	1.97	1.77	0.97	1.36	1.36	0.70	0.94	1.60	0.30	0.87	0.12
MgO	0.11	0.37	0.42	1.47	0.08	0.15	0.22	0.36	0.34	0.39	0.03	0.05	0.04	0.05	0.25	0.60
CaO	15.16	12.10	8.77	8.39	28.80	27.38	24.09	24.65	33.20	33.08	31.78	30.26	30.01	34.62	26.82	27.54
Na ₂ O	0.00	0.02	0.07	0.04	0.02	0.00	0.04	0.00	0.00	0.01	0.00	0.02	0.03	0.00	0.01	0.02
K ₂ O	0.00	0.00	0.00	0.00	0.06	0.00	0.13	0.30	0.12	0.01	1.12	0.93	0.00	0.50	0.00	0.00
Total	98.14	99.88	98.74	99.46	100.15	99.24	100.13	99.34	100.06	99.84	99.47	100.64	98.62	98.61	99.32	100.22
(apfu)*																
Si	2.95	3.01	3.01	2.99	2.97	2.96	2.98	2.99	2.96	2.96	2.91	2.95	2.95	2.95	2.97	2.97
Ti	0.01	0.00	0.01	0.01	0.01	0.01	0.01	0.01	0.03	0.03	0.02	0.02	0.04	0.00	0.02	0.01
Cr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Al	1.88	1.90	1.95	1.97	1.84	1.37	1.73	1.82	1.56	1.59	1.75	1.74	0.85	0.95	1.44	1.77
Fe ²⁺	1.33	1.77	1.51	2.06	0.62	0.49	0.88	0.97	0.18	0.12	0.30	0.35	0.25	0.20	0.63	0.62
Fe ³⁺	0.19	0.07	0.01	0.03	0.16	0.68	0.24	0.03	0.40	0.47	0.30	0.35	1.18	0.90	0.57	0.27
Mn	0.30	0.16	0.68	0.03	0.01	0.13	0.12	0.07	0.09	0.04	0.05	0.06	0.11	0.02	0.06	0.01
Mg	0.01	0.05	0.05	0.18	0.01	0.02	0.03	0.04	0.04	0.09	0.00	0.01	0.01	0.01	0.03	0.07
Ca	0.00	1.04	0.77	0.72	2.38	2.33	2.02	2.08	2.75	2.74	2.66	2.51	2.61	2.98	2.28	2.28
Na	1.32	0.00	0.01	0.01	0.00	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.01	0.00	0.00	0.00
F	0.00	0.00	0.00	0.00	0.02	0.00	0.00	0.00	0.03	0.00	0.27	0.22	0.00	0.13	0.00	0.00
X _{Fe}	1.00	0.99	0.98	0.95	0.99	0.99	0.99	0.98	0.96	0.96	1.00	1.00	1.00	1.00	0.99	0.96
(mol. %)																
Alm	37.70	57.87	49.33	68.79	14.48	13.87	26.80	21.27	2.24	3.67	8.86	12.08	7.15	0.00	15.48	16.84
Adr	11.03	3.40	0.72	1.45	8.71	35.54	11.95	1.60	20.47	23.55	15.04	18.12	60.39	48.71	30.65	14.25
Grs	39.31	31.74	25.26	22.70	76.17	45.37	56.37	73.28	72.91	68.30	74.37	67.43	28.55	50.33	50.65	66.15
Prp	0.49	1.50	1.72	5.91	0.32	0.62	0.87	1.52	1.34	1.49	0.11	0.19	0.17	0.22	1.04	2.44
Sps	11.41	5.46	22.95	1.08	0.33	4.60	3.98	2.33	3.03	2.99	1.56	2.11	3.75	0.74	2.10	0.29

*recalculated on the basis of Rickwood (1968), Fe²⁺/Fe³⁺ was estimated according to Droop (1987)

⇐

Fig. 3 Photomicrographs of skarns showing crystallization relationships (a–d, f, g – BSE; e, h – plane polarized light). **a** – Svatka Unit, Svatouch: garnet-pyroxene skarn, garnet is surrounded by a decompression plagioclase-amphibole corona and encloses titanite; **b** – Svatka Unit, Kuklík: garnet-pyroxene skarn, disintegration of Jd-rich pyroxene to clinopyroxene-plagioclase symplectite; **c** – Svatka Unit, Věcov: garnet-pyroxene skarn, disintegration of titanite (Al-bearing) to titanite-plagioclase symplectite; **d** – Svatka Unit, Čachnov: garnet-pyroxene skarn, older garnet grains are overgrown by younger garnet rich in andradite component; **e** – Svatka Unit, Svatouch: grunerite-pyroxene skarn with garnet, garnets are surrounded by amphibole coronas, pyroxenes intergrown with grunerite; **f** – Moldanubian Zone, Budeč: garnet-pyroxene skarn, fractures in older Grs-rich garnets are filled by younger andradite; **g** – Kutná Hora Complex, Malešov: inhomogeneous grossular-andradite garnet with patchy zoning; **h** – Kutná Hora Complex, Malešov: garnet-pyroxene skarn with cleavage veinlets, filled with epidote and clinopyroxene. Abbreviations used: Grt = garnet, Pl = plagioclase, Cpx = clinopyroxene, Hbl = hornblende, Ep = epidote

(Fig. 3b) and disintegration of titanite (Al-bearing) to titanite-plagioclase symplectite, are frequent (Fig. 3c). Stage M₃ produced the typical amphibole-plagioclase amphibolite-facies assemblage.

4.2. Moldanubian Zone

The following types of skarn were identified: garnet-clinopyroxene skarn, garnetite, magnetite skarn with amphibole, and clinopyroxene skarn. The most rocks are massive, only clinopyroxene skarn is banded. The microscopic structure is nematoblastic and nematogranoblastic.

The major minerals, garnet and clinopyroxene, are accompanied by less abundant magnetite, quartz, pla-

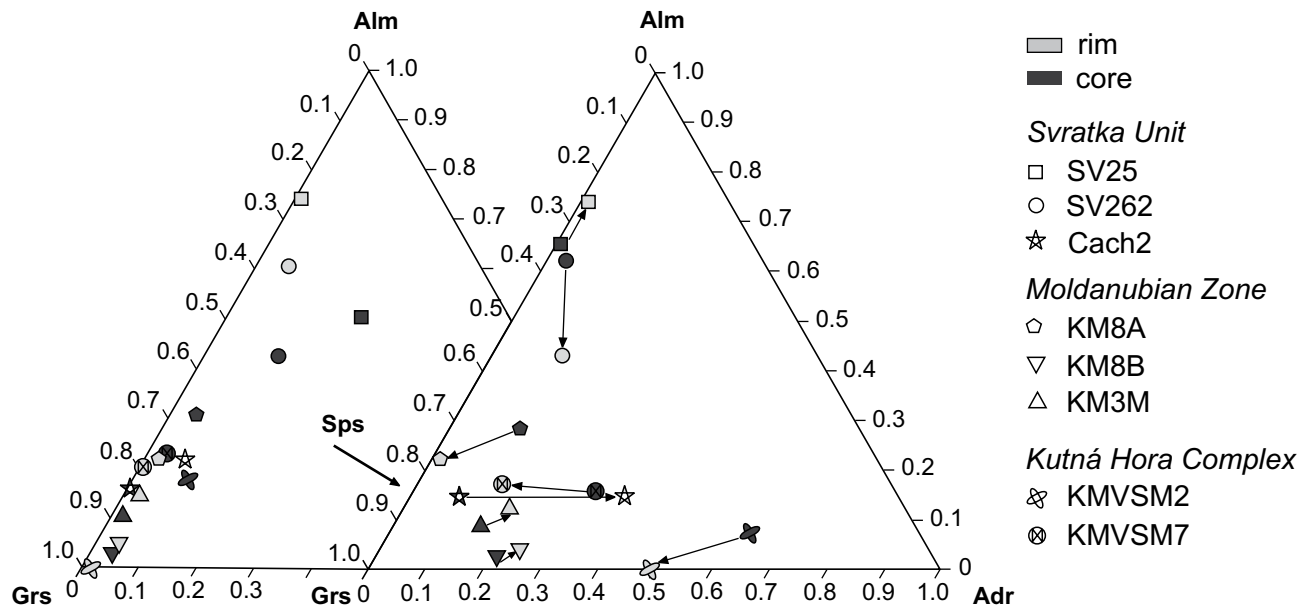


Fig. 4 Representative compositional trends for garnets from the Svatka Unit, the Moldanubian Zone and the Kutná Hora Complex. Arrows in the ternary diagrams indicate growth trends from the core to the rim.

Tab. 4 Representative chemical analyses of pyroxenes from the Svatka Unit, the Moldanubian Zone and the Kutná Hora Complex

location	Svatka Unit					Moldanubian Zone				Kutná Hora Complex			
	SV262	SV25	KM7b	KMSB1	Cach2	KM8A	KM8B	KM3A	KMVSM2	KMVSM7		KMVSM4	
sample	matrix	matrix	matrix	inclusion	matrix	matrix	matrix	matrix	matrix	core	rim	cleavage	incl. in Grt
(wt. %)													
SiO ₂	50.77	53.85	50.04	51.48	49.28	50.65	50.04	52.23	48.64	51.72	48.67	47.79	49.82
TiO ₂	0.00	0.11	0.02	0.11	0.06	0.11	0.12	0.08	0.04	0.00	0.09	0.14	0.03
Al ₂ O ₃	0.38	6.63	0.87	4.88	1.15	0.84	2.64	1.25	0.69	0.67	2.08	2.19	0.47
FeO	22.51	10.70	21.15	16.04	20.03	18.75	12.99	12.31	25.02	13.37	20.03	21.76	22.62
MnO	0.20	0.03	0.27	0.13	1.01	0.32	0.51	1.15	1.12	0.22	0.31	0.36	0.82
MgO	5.43	7.92	4.48	5.70	5.06	6.68	9.32	10.59	1.98	9.23	5.07	3.83	3.70
CaO	20.53	17.06	21.55	16.99	21.84	23.62	24.16	23.00	22.57	23.55	22.75	22.27	22.42
Na ₂ O	0.68	4.48	0.88	4.00	0.64	0.09	0.20	0.22	0.34	0.49	0.59	0.61	0.33
K ₂ O	0.00	0.00	0.00	0.00	0.10	0.00	0.01	0.01	0.00	0.00	0.00	0.00	0.00
Total	100.50	100.78	99.26	99.33	99.16	101.05	99.99	100.83	100.40	99.25	99.59	98.95	100.21
(apfu)*													
Si	2.00	1.97	1.99	1.96	1.96	1.97	1.91	1.97	1.96	1.99	1.92	1.92	1.99
Ti	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Al	0.02	0.28	0.04	0.21	0.18	0.04	0.09	0.05	0.03	0.03	0.10	0.10	0.01
Fe ²⁺	0.70	0.23	0.66	0.35	0.59	0.58	0.42	0.38	0.78	0.41	0.56	0.63	0.73
Fe ³⁺	0.04	0.09	0.04	0.16	0.08	0.03	0.00	0.01	0.07	0.03	0.10	0.10	0.02
Mn	0.01	0.00	0.01	0.00	0.03	0.01	0.02	0.04	0.04	0.01	0.01	0.01	0.03
Mg	0.32	0.43	0.27	0.32	0.30	0.39	0.53	0.60	0.12	0.53	0.30	0.23	0.22
Ca	0.87	0.67	0.92	0.69	0.93	0.98	0.99	0.93	0.98	0.97	0.96	0.96	0.96
Na	0.05	0.32	0.07	0.30	0.05	0.01	0.02	0.02	0.03	0.04	0.05	0.05	0.03
K	0.00	0.00	0.00	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
(mol %)													
Wo	44.81	47.00	48.42	45.00	48.19	49.41	50.68	47.68	49.35	49.81	49.81	49.64	48.88
En	16.49	30.00	14.01	21.00	15.55	19.45	27.20	30.53	6.01	27.15	15.43	11.87	11.21
Fs	38.70	23.00	37.57	34.00	36.27	31.14	22.12	21.80	44.64	23.03	34.76	38.49	39.91
WEF	94.81	68.00	93.14	70.00	94.98	99.34	98.50	98.38	97.31	96.35	95.35	95.10	97.43
Jd	1.56	24.00	3.03	16.00	0.75	0.06	1.50	1.88	0.00	1.29	0.76	0.81	0.85
Aeg	3.64	9.00	3.79	14.00	4.26	0.60	0.00	0.44	2.69	2.37	3.89	4.08	1.72

incl.=inclusion

*recalculated on the basis of Cawthorn and Collerson (1974), Fe²⁺/Fe³⁺ was estimated by charge balance proposed by the IMA guidelines

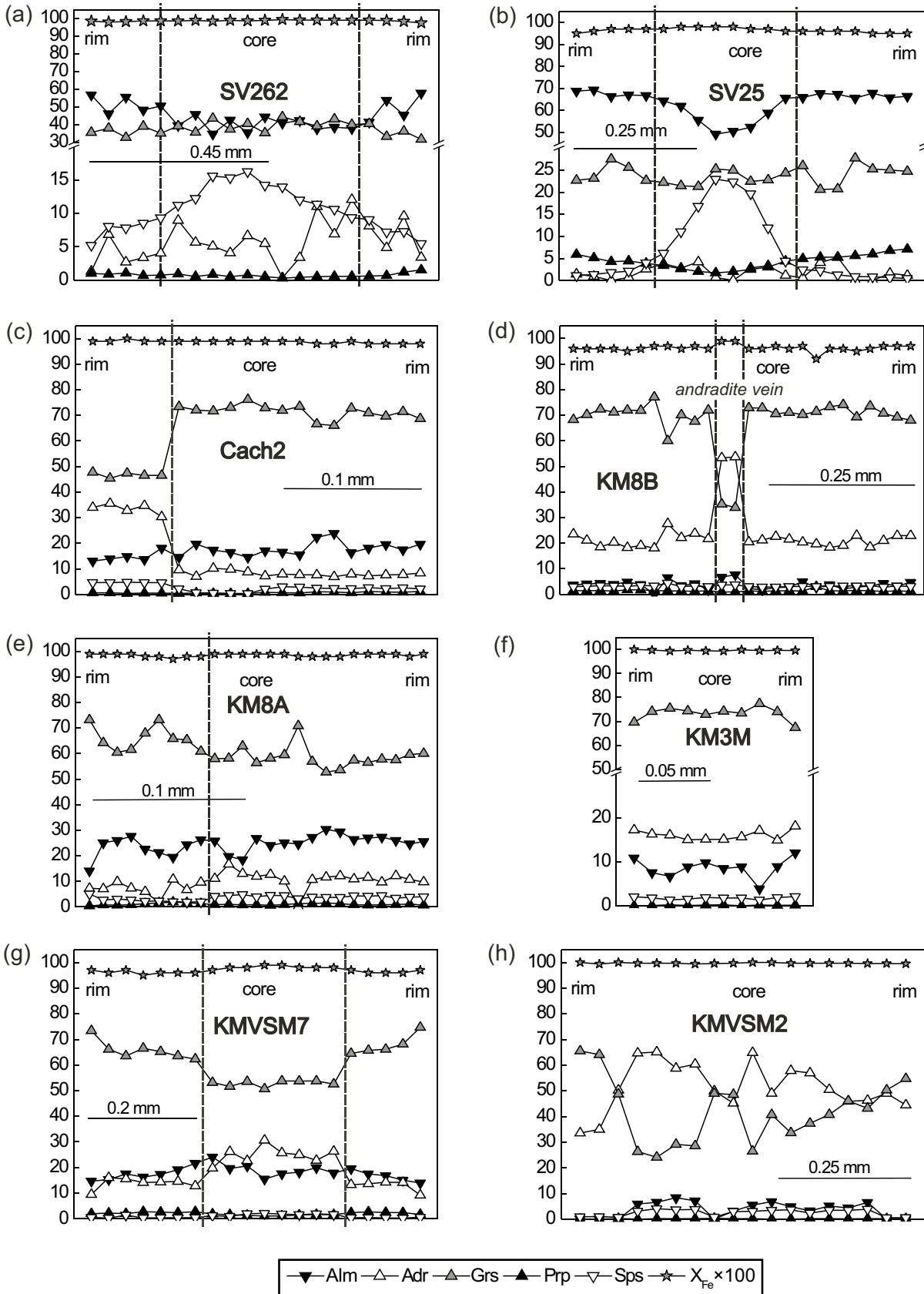


Fig. 5 Representative garnet profiles (mol %): a-c – Svatka Unit, Svatouch, Čachnov; d-f – Moldanubian Zone, Budeč, Vepřová; g-h – Kutná Hora Complex, Malešov.

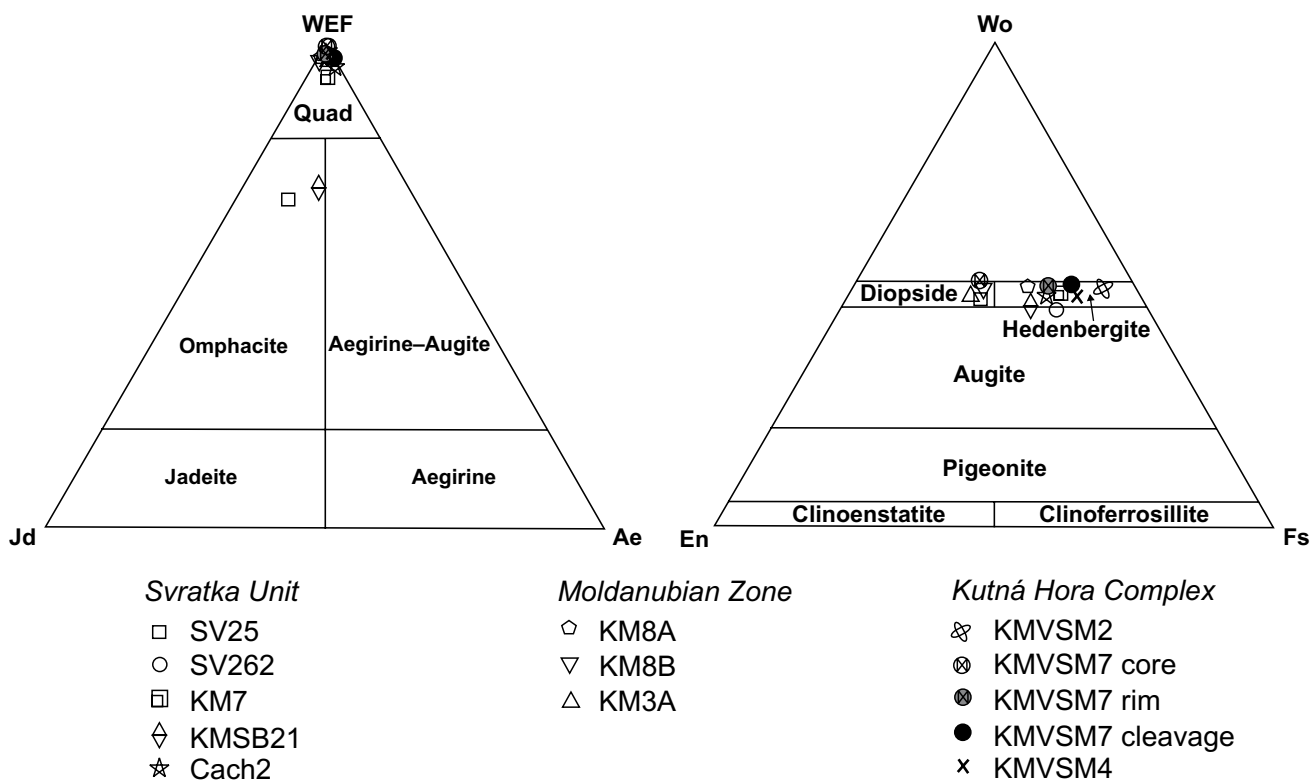


Fig. 6 Representative chemical compositions of pyroxenes from the Svatka Unit, the Moldanubian Zone and the Kutná Hora Complex in the classification diagram of Cawthorn and Collerson (1974).

Tab. 5 Representative chemical analyses of amphiboles from the Svatka Unit, the Moldanubian Zone and the Kutná Hora Complex

location sample position	Svatka Unit					Moldanubian Zone			Kutná Hora Complex
	SV262 matrix	SV25/98 matrix	Cach2 matrix	SV27/98 matrix	SV1/03 matrix	KM8C matrix	KM8B matrix	KM3A matrix	KMVSM5 matrix
(wt. %)									
SiO ₂	47.24	40.56	37.84	49.69	49.71	40.50	37.54	47.12	37.43
TiO ₂	0.14	0.97	0.39	0.01	0.11	1.00	0.30	0.74	0.47
Cr ₂ O ₃	0.00	0.03	0.00	0.00	0.00	0.00	0.00	0.10	0.00
Al ₂ O ₃	5.12	14.90	12.60	0.64	0.65	11.35	13.79	8.06	11.65
FeO	29.28	21.49	27.17	38.68	38.12	27.69	21.28	18.39	30.63
MnO	0.19	0.11	0.62	0.68	0.51	0.63	0.41	0.89	0.49
MgO	5.96	6.25	3.71	6.08	6.49	3.18	7.36	9.64	1.65
CaO	9.63	10.80	10.91	0.84	1.10	11.49	11.79	11.95	11.31
Na ₂ O	1.46	3.23	1.56	0.20	0.70	1.33	0.95	0.66	1.37
K ₂ O	0.00	0.02	2.08	0.02	0.00	1.56	3.62	0.46	2.40
F	0.00	0.15	0.00	0.12	0.00	0.00	0.55	0.00	0.00
Total	99.02	98.51	96.88	96.96	97.39	98.73	97.59	98.02	97.39
(apfu)*									
Si	7.29	6.11	6.08	7.99	7.96	6.34	5.87	6.98	6.10
Ti	0.02	0.11	0.05	0.00	0.01	0.00	0.00	0.08	0.06
Cr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.01	0.00
Al	0.93	2.65	2.38	0.13	0.12	2.09	2.53	1.41	2.23
Fe ²⁺	3.37	2.24	3.19	3.44	3.36	3.20	2.10	1.89	3.65
Fe ³⁺	0.02	0.47	0.46	0.00	0.00	0.42	0.69	0.39	0.53
Mn	0.00	0.01	0.08	0.00	0.00	0.08	0.05	0.11	0.07
Mg	1.37	1.40	0.89	1.46	1.55	0.74	1.72	2.13	0.40
Ca	1.59	1.74	1.88	0.15	0.19	1.93	1.98	1.90	1.97
Na	0.44	0.95	0.49	0.06	0.22	0.33	0.26	0.18	0.43
K	0.00	0.00	0.43	0.00	0.00	0.31	0.72	0.09	0.50
F	0.00	0.07	0.00	0.06	0.00	0.00	0.27	0.00	0.00
				grunerite	grunerite				

*recalculated on the basis of Richard and Clarke (1990), Fe²⁺/Fe³⁺ was estimated according to Robinson et al. (1981)

gioclase (bytownite to anorthite), epidote, clinozoisite and amphibole (hastingsite, grunerite). Titanite and apatite are the common accessories. Opaque minerals include a number of mineral species. The most common mineral is magnetite, corresponding to 20 vol. % in some samples. Pyrrhotite, pyrite and arsenopyrite are minor components. Rare löllingite and native bismuth are identified as inclusions in arsenopyrite.

The results of petrological study of skarns in the Moldanubian Zone (Holšice, Zliv, Vápenka, and Vlastějovice localities) were reported by Potužák (1996) and Drahotka et al. (2005), data from MZ in south-western Moravia (Županovice) by Pertoldová (1986), and from the Gföhl Unit in MZ (Slatina and Rešice) by Pertold et al. (1997).

4.2.1. Garnet

Garnet forms subhedral grains, which exhibit irregular compositional zoning. The compositional inhomogeneity is depicted in Figs 4, 5d–f. It includes variations in grossular (~50 to 80 mol. %), almandine (~1 to 30 mol. %) and andradite (~1 to 30 mol. %) components. In some instances, there is an outer rim of nearly pure grossular mantling the garnet grains. Some garnets contain fluorine in significant amounts (0.35–1.12 wt. %); the highest contents were recorded in grossular-rich outer rims. Garnet-pyroxene skarn with magnetite, sample KM8B03, contains two garnet generations with contrasting optical and compositional characteristics. The older garnet grains with predominating grossular component ($\text{Grs}_{77-60} \text{Alm}_{1-5} \text{Sps}_3 \text{Adr}_{8-24} \text{Prp}_{1-2}$, $X_{\text{Fe}} = 0.95-0.97$ mol. %) are corroded and fractured. The fractures are filled by a younger garnet with increased andradite component ($\text{Grs}_{34-35} \text{Alm}_{6-8} \text{Sps}_4 \text{Adr}_{53-54} \text{Prp}_1$, $X_{\text{Fe}} = 0.99$ mol. %; Figs 3f, 5d, Tab. 3). Garnet in sample KM3M is compositionally rather inhomogeneous and exhibits slight retrograde zoning, especially in the contents of grossular, almandine and andradite (from core to rim $\text{Grs}_{74-67} \text{Alm}_{9-12} \text{Adr}_{15-18}$, Fig. 5f, Tab. 6).

4.2.2. Clinopyroxene and amphibole

Clinopyroxenes belong to the hedenbergite–diopside group (Fig. 6). Replacement of clinopyroxene by secondary amphibole with hastingsite composition is common (Tab. 5).

Tab. 6 Representative chemical analyses of plagioclases from the Svatka Unit, the Moldanubian Zone and the Kutná Hora Complex

location sample position	Svatka Unit		Moldanubian Zone		Kutná Hora Complex
	SV25/98	Cach2	KM8C	KM3A	KMVM5
	matrix	matrix	matrix	matrix	matrix
(wt. %)					
SiO ₂	66.52	63.71	44.92	43.95	61.37
TiO ₂	0.00	0.14	0.00	0.00	0.00
Al ₂ O ₃	20.74	22.92	35.59	35.57	23.47
FeO	0.36	0.00	0.18	0.19	0.30
MnO	0.00	0.00	0.00	0.00	0.00
MgO	0.00	0.01	0.00	0.00	0.00
CaO	1.41	3.83	18.92	19.19	5.99
Na ₂ O	10.94	9.38	0.62	0.61	8.57
K ₂ O	0.00	0.22	0.01	0.02	0.25
Total	99.97	100.19	100.24	99.52	99.95
(apfu)*					
Si	11.69	11.23	8.27	8.17	10.95
Ti	0.00	0.00	0.00	0.00	0.00
Al	4.29	4.76	7.72	7.79	4.93
Fe ²⁺	0.05	0.00	0.03	0.03	0.05
Fe ³⁺	0.00	0.00	0.00	0.00	0.00
Mn	0.00	0.00	0.00	0.00	0.00
Mg	0.00	0.00	0.00	0.00	0.00
Ca	0.27	0.72	3.73	3.82	1.15
Na	3.73	3.21	0.22	0.22	2.96
K	0.00	0.05	0.00	0.00	0.06
(mol. %)					
Ab	93.30	80.60	5.60	5.40	71.10
An	6.70	18.20	94.30	94.50	27.50
Or	0.10	1.20	0.10	0.10	1.40

*recalculated on the basis of Deer et al. (1966)

4.2.3. Plagioclase and titanite

Compositionally homogeneous calcic plagioclase occurs in minor amounts interstitially among garnet and clinopyroxene grains. It contains over 90 mol. % of anorthite component (Tab. 5). Titanite is compositionally heterogeneous. The malayaite component is present in some samples, as indicated by increased tin (up to 1.23 wt. % SnO), and fluorine contents (up to 0.52 wt. %) (Tab. 7).

4.2.4. Metamorphic development

The chemical composition of garnet in skarns indicates weak retrograde zoning. The conditions of amphibolite-facies metamorphism presumably varied during the crystallization of skarn minerals, as probably did the oxygen and fluorine fugacity. The oxygen fugacity was probably increased only locally.

Tab. 7 Representative chemical analyses of titanites from the Svratka Unit, the Moldanubian Zone and the Kutná Hora Complex

location sample position (wt. %)	Svratka Unit					Moldanubian Zone		Kutná Hora Complex
	SV25/98 matrix	KMSB21c pure titanite	KMSB21c pure titanite	KMSB21c symplectite	KMSB21c symplectite	KM3A matrix	KM3C matrix	KMV20A matrix
SiO ₂	31.07	31.37	30.69	30.05	30.20	30.26	30.40	29.91
TiO ₂	37.30	31.24	31.51	34.92	34.71	36.70	29.02	36.19
Cr ₂ O ₃	0.00	0.00	0.01	0.00	0.00	0.00	0.00	0.03
Al ₂ O ₃	3.01	5.47	4.72	1.76	1.91	3.28	7.38	2.67
FeO	1.26	1.06	0.98	0.76	0.75	0.79	0.69	1.19
MnO	0.15	0.04	0.01	0.00	0.00	0.19	0.00	0.02
MgO	0.00	0.01	0.01	0.00	0.00	0.00	0.00	0.00
CaO	28.17	29.24	28.84	28.20	28.29	28.94	30.21	29.02
Na ₂ O	0.02	0.00	0.04	0.04	0.01	0.01	0.00	0.00
ZrO ₂	0.00	0.00	0.00	0.01	0.02	0.00	0.00	0.00
Nb ₂ O ₅	0.00	0.07	0.12	0.12	0.10	0.11	0.09	0.00
Ta ₂ O ₅	0.00	0.00	0.00	0.07	0.04	0.20	0.00	0.00
F	0.00	1.51	1.31	0.47	0.52	0.00	0.52	0.00
V ₂ O ₅	0.00	0.05	0.00	0.00	0.00	0.20	0.18	0.00
SnO	0.00	0.11	1.51	2.67	2.57	0.00	1.23	0.00
Total	100.98	100.17	99.75	99.07	99.12	100.68	99.72	99.03
(apfu)*								
Si	4.01	3.82	3.81	3.92	3.92	3.87	3.79	3.96
Ti	3.62	2.86	2.94	3.42	3.39	3.53	2.72	3.60
Cr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Al	0.46	0.79	0.69	0.27	0.29	0.49	1.08	0.42
Fe ²⁺	0.14	0.11	0.10	0.08	0.08	0.08	0.07	0.13
Fe ³⁺	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Mn	0.02	0.00	0.00	0.00	0.00	0.02	0.00	0.00
Mg	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Ca	3.90	3.82	3.84	3.94	3.94	3.96	4.04	4.12
Na	0.01	0.00	0.01	0.01	0.00	0.00	0.00	0.00
Zr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Nb	0.00	0.00	0.01	0.01	0.01	0.01	0.01	0.00
Ta	0.00	0.00	0.00	0.00	0.00	0.01	0.00	0.00
F	0.00	0.04	0.03	0.01	0.01	0.00	0.01	0.00
V	0.00	0.00	0.00	0.00	0.00	0.02	0.01	0.00
Sn	0.00	0.01	0.08	0.16	0.15	0.00	0.07	0.00

*recalculated on the basis of Deer et al. (1966, 1972)

4.3. Kutná Hora Complex

Inhomogeneous lenses and bodies of skarn in the Kutná Hora Complex are located in variously migmatized two-mica gneisses, some garnet bearing. These skarns are massive, fine-grained rocks with compact, banded or schlieren-like structure. The texture is granoblastic or nemato-granoblastic. The following skarn types were identified: garnet-pyroxene skarns, pyroxene skarns, pyroxene skarns with garnet, pyroxene-garnet skarns grading to garnetites. Garnet-pyroxene skarn is the most abundant type. Clinopyroxenes in these rocks are partly or strongly replaced by amphiboles. The typical mineral assemblage is hedenbergite, grossular-andradite garnet, epidote and magnetite. Accessory minerals are apatite,

titanite, allanite and cerite. Secondary minerals are represented by amphiboles and four generations of epidote.

4.3.1. Garnet

Garnet forms clusters or irregular layers. Based on composition, two groups of garnets are distinguished (Figs 4, 5g–h). The first group includes three-component andradite-grossular-almandine garnets (Fig. 5g) formed by blastic growth in epidosite layers; the second group contains grossular-andradite garnets (Fig. 3g, 5h). Prograde zoning in three-component garnets is expressed by variation in the abundances of all three components (from core to rim $\text{Grs}_{50-75} \text{Alm}_{24-15} \text{Sps}_{2-0} \text{Adr}_{31-9} \text{Prp}_{1-3}$, $X_{\text{Fe}} = 0.99-0.96$ mol. %, Fig. 5g, Tab. 3). The three-component

garnets enclose epidote and clinopyroxene inclusions. In some cases, garnets are affected by secondary replacement by younger epidote. Garnets of the second group also exhibit compositional zoning, even though the individual grains are often compositionally inhomogeneous. There is a general tendency to andradite enrichment in the crystal cores and to a grossular increase toward rims (from core to rim $\text{Grs}_{24-65} \text{Alm}_{9-0} \text{Sps}_{4-0} \text{Adr}_{65-33}$, Fig. 5h, Tab. 3). Younger grossular-rich garnet follows interfaces among grains. Hydrogrossular containing up to 0.46 wt. % fluorine is also a two-component garnet.

4.3.2. Clinopyroxene

Clinopyroxene is the most abundant mineral. Three generations of clinopyroxene are distinguished. Hedenbergite, as the oldest pyroxene phase, is enclosed in garnet and contains *c.* 1 mol. % of the jadeite component. The second generation includes matrix clinopyroxenes, which are unzoned, or rarely show a weak compositional zoning. The crystal cores have diopside composition, while the rims are enriched in FeO (20–22 %) and can be classified as hedenbergite (Fig. 6, Tab. 4). The youngest generation of clinopyroxene fills cleavage in the form of veinlets (Fig. 3h). It has hedenbergite composition (Fig. 6, Tab. 4).

4.3.3. Plagioclase and amphibole

Both plagioclase and amphibole are rare and only secondary. Plagioclase is oligoclase ($\sim \text{An}_{27}$; Tab. 6); amphibole corresponds to hastingsite (Tab. 5).

4.3.4. Epidote

Microscopic study resulted in identification of five generations of epidote–clinozoisite series minerals. Probably the oldest is epidote I in epidosite bands in sample KMVSM7, which shows replacement of epidote by garnet. This newly formed garnet contains inclusions of epidote I. Epidote of the second generation occurs as euhedral crystals enclosing, in their cores, allanite (cerite?). This may represent a phase enriched in REE, released during replacement of epidote I by garnet in epidosite layers. Epidote III was produced by replacement of certain compositional zones in garnet, leaving behind atoll-like (hollow) garnet crystals. The youngest epidote IV occurs together with clinopyroxene, filling the cleavage in the form of veinlets (Fig. 3h). The table of representative epidote analyses (Tab. 8) contains data for all four epidote generations.

Epidotes are typically free of compositional zoning. The chemical composition of epidote enclosed in garnet,

epidote formed via replacement of garnet, and epidote in late cleavage veinlets (I, II, IV) shows very limited variation ($\text{Si} = 2.96\text{--}2.99$, $\text{Al} \sim 2.25$ apfu; Tab. 8). Epidotes enriched in REE contain up to 5 wt. % Ce_2O_3 , 3 wt. % La_2O_3 , and 2 wt. % Nd_2O_3 . The contents of ThO_2 , UO_2 , Sm_2O_3 and F are lower than 1 wt. %.

4.3.5. Metamorphic development

Garnets in skarns in the Kutná Hora Complex are rather compositionally heterogeneous, including variation in the grossular and andradite components (Fig. 5g–h). The relative abundance of mineral phases rich in REE (mainly metamict allanite) is probably related to enrichment in REE in the precursor epidosite bands. These layers were probably derived by hydrothermal alteration of Ca- and Fe-enriched layers. Due to the extensive stability field of epidote in amphibolite facies, epidotes persisted up to a stage of epidote replacement by garnet. The prograde compositional zoning was observed in some garnets (Fig. 5g). The andradite component in the garnet indicates fluctuations in oxygen fugacity (Fig. 5h). The rather constant composition of the individual generations of pyroxene and epidote, belonging to the successive stages of metamorphism, suggests that variation in the P–T conditions was small.

5. Whole-rock geochemistry of the skarns

Altogether 59 whole-rock samples were used for chemical characterization of skarns in the studied units. The samples are listed in Tab. 1; major- and trace-element analyses are presented in Tabs 9–11.

5.1. Svratka Unit

Skarns from this unit are represented by samples from 13 localities. The mineral assemblages are dominated by Grt–Cpx, Cpx and Grt–Amph. The analyses of major and trace elements are presented in Tab. 9. The abundances of the major elements are highly variable: 35–55 wt. % SiO_2 , 1–18 wt. % Al_2O_3 , 1–23 wt. % CaO , 2–47 wt. % Fe_2O_3 and 2–8 wt. % MgO (Fig. 10).

The chondrite-normalized REE patterns for most of the samples resemble those for post-Archaeon sediments (Taylor and McLennan 1985; Condie 1993). Overall, the abundances of the trace elements, including REE, are highly variable. Skarns from the Svratka Unit match the skarns from the Moldanubian Zone in their geochemical characteristics. A positive Eu anomaly is present in samples 35 and 36 from Teplá ($\text{Eu}/\text{Eu}^* = 1.94$ and 1.39), 49 from Čachnov ($\text{Eu}/\text{Eu}^* = 1.99$) and 52 from Ruda ($\text{Eu}/\text{Eu}^* = 1.19$). Samples 57 (Pernštejn) and 42 and

Tab. 8 Representative chemical analyses of epidotes from the Svatka Unit, the Moldanubian Zone and the Kutná Hora Complex

location sample position	Svatka Unit	Moldanubian Zone	Kutná Hora Complex				Kutná Hora Complex KMVSM1 with REE	
	Cach2 matrix	KM3E matrix	KMVSM7 phase I.	KMVSM1 after Grt con.	KMVSM7 incl.in Grt cleavage			
(wt. %)							(wt. %)	
SiO ₂	37.80	39.22	37.61	37.81	37.53	37.49	SiO ₂	39.62
TiO ₂	0.00	0.06	0.00	0.00	0.00	0.00	TiO ₂	0.25
Cr ₂ O ₃	0.00	0.04	0.00	0.00	0.00	0.00	Al ₂ O ₃	12.93
Al ₂ O ₃	24.35	25.92	24.32	24.01	24.23	24.02	FeO	14.17
FeO	10.58	10.20	11.47	11.65	11.16	11.53	MnO	0.66
MnO	0.00	0.08	0.00	0.00	0.00	0.00	MgO	0.13
MgO	0.00	0.00	0.03	0.00	0.02	0.01	La ₂ O ₃	3.04
CaO	22.55	23.32	24.24	23.70	24.50	23.86	Ce ₂ O ₃	5.28
Na ₂ O	0.00	0.01	0.00	0.00	0.00	0.00	Pr ₂ O ₃	0.58
K ₂ O	0.00	0.00	0.00	0.00	0.00	0.00	Nd ₂ O ₃	2.03
F	0.00	0.00	0.00	0.00	0.00	0.00	Sm ₂ O ₃	0.19
Cl	0.00	0.00	0.00	0.00	0.00	0.00	ThO ₂	0.55
Total	95.28	98.85	97.66	97.17	97.43	96.90	UO ₂	0.22
(apfu)							CaO	12.58
Si	3.03	3.02	2.96	2.99	2.97	2.98	P ₂ O ₅	0.05
Ti	0.00	0.00	0.00	0.00	0.00	0.00	F	0.24
Cr	0.00	0.00	0.00	0.00	0.00	0.00	H ₂ O	3.00
Al	2.30	2.35	2.26	2.24	2.26	2.25	Total	95.52
Fe ²⁺	0.00	0.00	0.00	0.00	0.00	0.00	(apfu)	
Fe ³⁺	0.71	0.66	0.76	0.77	0.74	0.76	Si	3.75
Mn	0.00	0.01	0.00	0.00	0.00	0.00	Ti	0.02
Mg	0.00	0.00	0.00	0.00	0.00	0.00	Al	1.44
Ca	1.94	1.93	2.05	2.01	2.07	2.03	Fe ³⁺	1.05
Na	0.00	0.00	0.00	0.00	0.00	0.00	Mn	0.05
K	0.00	0.00	0.00	0.00	0.00	0.00	Mg	0.02
F	0.00	0.00	0.00	0.00	0.00	0.00	La	0.11
Cl	0.00	0.00	0.00	0.00	0.00	0.00	Ce	0.18
							Pr	0.02
							Nd	0.07
							Sm	0.01
							Th	0.01
							U	0.01
							Ca	1.28
							P	0.00
							F	0.07
							H	1.93

*recalculated on the basis of Deer et al. (1966, 1972), all Fe was considered as Fe³⁺

"after grt con." – "after garnet consumption"

43 (Kuklík) exhibit positive Eu anomalies ($Eu^*/Eu = 2.00-8.62$) and low total REE contents.

5.2. Moldanubian Zone

The studied samples from nine localities represent a varied set with different mineralogical compositions. The abundances of major- and selected trace elements are given in Tab. 10. Chemistry of the individual samples depends on their modal composition and variability in the composition of the minerals, especially

of garnet. Major elements exhibit a wide variation: 20–60 wt. % SiO₂, 1–17 wt. % Al₂O₃, 5–30 wt. % CaO, 1–66 wt. % Fe₂O₃ and 0.5–14 wt. % MgO. A majority of samples shows a negative correlation between SiO₂ and CaO or FeO₁ (Fig. 7). On the other hand, Al₂O₃ correlates positively with TiO₂, Zr and total REE (Fig. 9). For example, samples from Budeč and Vepřová (Nos 1, 6, 7 and 8), with those from Rešice and Slatina (Nos 14 and 18), contain Al₂O₃ > 12 wt. % and TiO₂ > 0.39 wt. %. Their composition is comparable with trends in the trace-element abundances in the paragneisses

Tab. 9 Whole-rock major- and trace-element compositions from the Svatka Unit. Major oxides in wt. %, trace elements in ppm, Au in ppb

Sample No.	SiO ₂	TiO ₂	Al ₂ O ₃	Fe ₂ O ₃	MgO	MnO	CaO	Na ₂ O	K ₂ O	P ₂ O ₅	Total	Ba	Co	Rb	Sn	Sr	Zr	Pb
28	41.09	0.60	17.64	3.26	2.13	0.48	21.35	0.69	0.42	0.14	99.67	n.a.	n.a.	22.2	79.1	1080.9	n.a.	25.1
29	45.00	0.40	10.58	2.47	6.56	0.63	20.65	0.58	0.14	0.14	99.80	n.a.	n.a.	27.5	46.3	645.5	n.a.	20.4
30	42.63	0.37	8.37	5.84	2.95	1.41	20.83	0.33	0.08	0.14	100.40	n.a.	n.a.	21.0	9.4	250.3	n.a.	9.6
32	48.26	0.07	2.46	2.00	5.21	0.42	11.61	0.66	0.03	0.08	99.23	1.2	35.1	0.5	35.0	94.4	24.4	0.3
34	47.92	0.11	3.76	30.04	5.05	0.59	12.16	0.62	0.04	0.12	99.89	2.0	41.9	0.6	40.0	75.8	35.3	0.6
35	44.81	0.62	12.32	4.50	3.87	0.28	19.77	0.99	0.66	0.09	99.86	n.a.	n.a.	38.0	131.0	158.0	n.a.	14.0
36	47.28	0.61	13.43	2.92	3.39	0.30	18.90	1.86	0.41	0.14	100.01	n.a.	n.a.	37.0	70.0	557.0	n.a.	11.0
37	44.82	0.36	9.64	3.58	2.24	0.42	22.28	0.28	0.35	0.16	99.90	n.a.	n.a.	14.4	49.6	243.3	n.a.	9.8
38	43.63	0.55	11.53	21.60	1.96	0.35	19.53	0.47	0.26	0.14	100.02	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
39	42.07	0.50	11.30	23.24	1.69	0.40	19.81	0.27	0.30	0.18	99.76	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
40	44.41	0.42	10.44	19.54	2.71	0.37	20.35	0.85	0.05	0.22	99.36	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
41	41.78	0.54	12.39	19.44	2.16	0.36	21.72	0.79	0.05	0.18	99.79	15.0	16.4	3.5	91.0	384.7	151.2	1.1
42	47.00	0.02	0.88	23.94	5.60	1.60	19.86	0.38	0.01	0.09	99.79	7.0	18.7	0.8	23.0	66.2	10.0	9.7
43	35.29	0.02	0.91	46.84	2.69	2.05	12.18	0.18	0.08	0.07	99.78	9.0	19.6	1.8	70.0	61.1	5.0	2.5
44	46.30	0.31	7.58	5.78	3.11	0.44	16.92	1.89	0.11	0.09	99.17	n.a.	n.a.	23.5	77.2	248.3	n.a.	6.7
45	42.76	0.56	10.88	19.79	2.45	0.37	21.58	0.94	0.08	0.18	99.78	21.0	17.3	5.2	84.0	466.1	154.9	1.1
47	41.28	0.58	13.00	18.77	1.97	0.41	22.93	0.13	0.02	0.18	99.72	13.0	10.0	2.6	254.0	843.5	142.2	2.6
48	52.26	0.57	12.51	2.30	2.94	0.41	15.87	3.89	0.30	0.14	100.19	97.8	12.0	7.5	57.0	837.1	188.0	1.4
49	50.20	0.49	12.18	4.24	2.45	0.48	16.94	2.62	0.23	0.14	99.80	54.0	12.5	6.2	136.0	490.7	178.1	0.8
50	54.58	0.39	8.55	1.86	3.01	0.61	15.20	3.17	0.20	0.16	99.35	25.6	9.7	2.7	45.0	246.8	215.0	0.7
51	49.47	0.65	14.56	4.34	2.15	0.36	10.53	2.12	0.62	0.18	99.83	160.2	14.2	18.0	60.0	555.5	143.4	1.7
52	55.06	0.41	9.93	2.29	3.36	0.29	10.82	1.25	0.47	0.13	100.23	187.1	13.7	15.6	80.0	313.0	100.4	1.5
53	44.94	0.39	11.35	2.28	3.59	0.40	17.48	1.58	0.49	0.14	99.47	59.3	22.8	20.9	64.0	1370.9	107.8	1.4
54	47.17	0.46	11.87	2.77	5.66	0.63	19.26	0.95	0.71	0.16	99.91	n.a.	n.a.	42.7	84.4	486.8	n.a.	45.1
55	40.80	0.83	18.17	37.81	2.60	0.43	1.12	0.09	0.01	0.13	99.82	29.0	35.1	0.9	116.0	21.1	208.2	1.2
56	39.22	0.87	18.45	39.49	2.63	0.35	0.87	0.02	0.02	0.20	99.87	26.0	40.3	1.2	135.0	5.2	183.0	1.0
57	47.01	0.15	3.58	18.56	7.56	1.00	19.92	0.46	0.12	0.16	99.69	78.0	22.2	7.0	160.0	153.7	66.4	17.1

Sample No.	Zn	Bi	Au	La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
28	102.1	n.a.	n.a.	33.03	66.43	7.68	30.94	5.97	1.25	5.32	0.62	4.87	1.50	2.52	0.51	2.88	0.46
29	105.1	n.a.	n.a.	22.49	43.84	5.68	19.18	4.53	0.77	3.77	n.a.	3.29	0.86	1.39	0.29	1.78	0.24
30	218.5	n.a.	n.a.	22.14	43.52	5.36	21.26	4.23	0.74	3.59	0.00	3.72	0.81	1.52	0.32	1.67	0.23
32	3.0	2.2	26.9	4.40	8.00	1.02	4.30	1.00	0.30	1.15	0.17	0.97	0.17	0.61	0.10	0.63	0.10
34	6.0	n.a.	n.a.	5.40	10.10	1.25	4.50	1.06	0.34	1.32	0.25	1.23	0.31	1.34	0.30	2.51	0.41
35	48.0	n.a.	n.a.	25.20	52.70	4.90	25.40	7.02	4.40	6.86	0.00	6.16	1.54	2.95	0.40	3.68	0.52
36	36.0	n.a.	n.a.	38.80	72.70	7.00	29.70	4.55	2.34	5.81	0.92	5.13	1.54	3.46	0.49	3.42	0.50
37	57.2	5.8	0.1	54.20	94.60	9.90	43.80	8.86	1.67	6.55	1.11	5.53	1.48	2.49	0.50	3.00	0.43
38	n.a.	n.a.	n.a.	62.90	108.80	11.42	38.60	5.90	1.64	5.05	0.81	5.04	0.91	2.38	0.35	2.35	0.30
39	n.a.	n.a.	n.a.	31.50	56.90	6.71	26.50	5.40	1.26	4.65	0.86	4.50	0.91	2.45	0.39	2.23	0.34
40	n.a.	n.a.	n.a.	20.60	38.80	4.88	20.30	4.20	1.42	4.53	0.90	4.56	0.98	2.84	0.43	2.32	0.43
41	8.0	0.8	0.5	28.80	63.10	7.28	29.60	5.15	1.20	4.68	0.81	4.39	0.94	2.58	0.41	2.44	0.34
42	70.0	21.4	1.4	3.80	1.70	0.34	1.10	0.11	0.18	0.18	0.04	0.29	0.06	0.14	0.03	0.14	0.02
43	120.0	5.0	0.8	2.40	3.00	0.45	1.90	0.45	1.35	0.51	0.08	0.46	0.10	0.28	0.04	0.21	0.03
44	77.2	n.a.	n.a.	7.13	20.81	3.38	14.33	3.64	0.84	3.02	0.00	2.91	0.52	1.59	0.29	1.52	0.20
45	8.0	0.8	0.5	35.70	71.40	8.73	34.20	5.77	1.14	5.03	0.82	4.49	0.92	2.64	0.40	2.53	0.36
47	23.0	0.4	0.5	37.70	70.80	9.14	35.80	6.21	1.64	5.32	0.86	4.68	0.92	2.72	0.43	2.82	0.40
48	18.0	0.1	1.2	37.00	71.00	8.27	32.90	6.80	1.62	5.99	0.94	5.01	0.93	2.78	0.41	2.96	0.42
49	7.0	0.2	0.9	38.90	69.80	8.52	33.60	6.70	4.27	6.40	1.03	5.71	1.07	3.24	0.48	2.97	0.45
50	3.0	0.1	0.9	28.20	54.50	6.35	25.90	4.80	1.33	3.69	0.48	2.80	0.47	1.28	0.20	1.44	0.21
51	15.0	0.4	1.8	50.20	80.30	11.00	44.60	7.40	2.20	5.40	0.83	4.83	0.94	2.87	0.44	3.16	0.42
52	15.0	0.3	19.7	25.90	39.80	6.28	25.50	5.00	1.86	4.54	0.70	3.79	0.77	2.20	0.31	1.94	0.28
53	8.0	2.6	2.2	28.00	49.40	6.37	26.70	5.30	1.23	4.76	0.77	4.54	0.87	2.62	0.36	2.53	0.33
54	448.1	n.a.	n.a.	31.38	59.04	6.45	27.70	4.41	1.41	4.38	0.89	4.67	0.82	2.63	0.29	2.46	0.35
55	71.0	0.1	0.5	18.00	37.90	4.93	19.70	4.12	1.42	5.90	1.05	6.29	1.55	4.70	0.79	4.95	0.70
56	57.0	0.1	0.5	36.10	69.90	8.74	35.60	6.06	1.21	5.61	0.90	4.72	0.94	2.74	0.46	2.87	0.43
57	62.0	4.5	1.8	9.60	17.10	2.18	7.20	1.51	0.93	1.32	0.23	1.26	0.30	0.84	0.15	1.02	0.14

n.a. = not analyzed

Tab. 10 Whole-rock major- and trace-element compositions of skarns from the Moldanubian Zone. Major oxides in wt. %, trace elements in ppm, except for Au which is in ppb

Sample No.	SiO ₂	TiO ₂	Al ₂ O ₃	Fe ₂ O ₃	MgO	MnO	CaO	Na ₂ O	K ₂ O	P ₂ O ₅	Total	Ba	Co	Rb	Sn	Sr	Zr	Pb
1	44.98	0.39	12.37	2.63	3.76	0.51	19.35	0.69	0.87	0.14	99.19	n.a.	n.a.	129.6	567.2	459.7	n.a.	17.5
2	39.39	0.33	8.71	6.46	7.61	0.57	17.16	0.39	1.53	0.24	99.03	n.a.	n.a.	150.4	550.6	128.7	n.a.	15.7
4	36.94	0.23	10.49	32.67	3.58	0.43	12.16	0.84	1.07	0.07	99.86	66.0	6.2	25.7	494.0	163.9	47.0	2.5
5	20.45	0.19	3.26	66.39	3.09	0.16	4.99	0.39	0.38	0.11	99.71	50.0	6.8	13.0	419.0	20.2	38.9	1.0
6	51.27	0.67	16.84	1.88	0.51	0.13	17.72	1.16	0.36	0.17	99.70	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
7	55.77	0.60	15.29	0.66	2.48	0.15	12.90	3.02	0.96	0.19	99.54	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
8	50.97	0.61	16.56	1.68	0.58	0.13	18.70	0.97	0.20	0.19	97.80	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.
13	44.17	0.51	4.51	22.53	2.03	1.03	24.41	0.15	0.03	0.16	99.89	26.0	23.8	2.6	183.0	58.4	105.9	1.5
14	60.32	0.86	15.56	4.93	2.45	0.05	5.62	3.96	5.24	0.23	99.85	450.0	8.8	255.8	42.0	146.7	196.0	0.8
15	43.22	0.33	9.10	16.32	4.28	0.34	24.59	0.62	0.08	0.12	99.99	6.0	10.3	5.2	128.0	101.2	92.5	1.1
16	37.77	0.23	10.82	18.33	1.30	0.41	29.22	0.24	0.03	0.14	99.91	4.0	4.5	2.8	211.0	72.1	84.2	1.3
17	50.92	0.07	1.10	9.09	14.27	0.13	23.34	0.33	0.02	0.06	99.70	3.0	8.3	1.3	55.0	29.5	19.8	0.8
18	41.24	0.83	13.84	26.51	2.49	0.22	13.59	1.59	0.24	0.20	99.73	32.0	18.1	17.6	554.0	67.9	438.6	0.3
19	39.37	0.34	16.81	27.45	1.99	0.25	13.79	0.87	0.10	0.19	99.91	11.0	15.1	5.2	277.0	32.8	151.6	0.9
20	47.41	0.08	1.52	26.05	3.27	0.81	21.06	0.19	0.03	0.09	99.75	6.0	29.4	2.6	9.0	100.9	20.8	0.6
21	39.00	0.36	8.82	18.31	1.64	0.44	30.11	0.06	0.01	0.12	99.94	2.0	5.4	0.4	67.0	71.5	134.3	0.5
22	45.00	0.34	8.66	22.94	2.15	0.56	19.15	0.90	0.19	0.12	99.91	42.0	12.3	11.5	34.0	287.3	117.2	1.6
23	46.02	0.05	1.23	23.66	5.06	1.15	21.87	0.67	0.01	0.08	99.79	7.0	25.3	0.3	143.0	20.6	15.5	22.2
24	45.08	0.05	1.22	23.63	4.84	1.10	23.00	0.63	0.01	0.07	99.84	6.0	22.9	0.3	139.0	22.6	13.5	33.8
25	35.12	0.54	8.53	10.47	4.88	0.98	27.85	0.58	0.45	0.37	99.74	41.0	3.6	26.7	313.0	262.6	176.0	10.3
26	47.99	0.08	2.04	21.69	5.85	1.90	17.93	0.54	0.07	0.12	99.81	265.0	25.5	2.2	47.0	62.0	25.2	22.9
27	43.79	0.08	2.26	30.66	4.75	2.08	16.10	0.37	0.15	0.12	99.87	19.0	25.9	2.8	89.0	78.5	26.2	6.3

Sample No.	Zn	Bi	Au	Cu	La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
1	131.7	n.a.	n.a.	16.6	40.54	81.98	9.97	35.89	7.55	2.21	6.65	0.34	7.14	1.34	3.39	0.67	4.43	0.57
2	250.2	n.a.	n.a.	143.2	9.29	24.84	3.66	17.30	5.24	2.29	4.38	0.68	4.14	0.72	1.87	0.23	2.10	0.31
4	121.0	0.8	1.1	n.a.	16.80	40.50	5.51	21.30	5.08	1.80	5.27	1.10	6.38	1.32	3.82	0.69	4.46	0.64
5	71.0	1.8	1.8	n.a.	7.50	15.40	2.07	8.90	1.52	0.72	1.22	0.20	1.03	0.23	0.64	0.11	0.62	0.09
6	n.a.	n.a.	n.a.	n.a.	45.15	89.35	10.42	42.12	8.69	1.46	7.35	0.67	6.28	1.51	2.82	0.53	3.56	0.49
7	n.a.	n.a.	n.a.	255.5	42.57	77.02	9.20	35.55	9.69	1.50	6.74	0.90	5.60	1.21	2.34	0.46	3.53	0.51
8	n.a.	n.a.	n.a.	457.9	39.73	77.26	8.70	34.82	7.52	2.59	6.59	0.76	5.21	1.45	2.59	0.51	2.95	0.40
13	13.0	5.5	0.7	1.2	15.90	30.10	3.66	14.30	2.71	0.53	2.85	0.50	3.00	0.63	1.75	0.28	1.77	0.24
14	10.0	0.7	0.5	1.7	62.40	132.20	16.97	65.00	12.65	2.26	9.87	1.54	7.84	1.50	4.04	0.62	3.97	0.56
15	5.0	0.4	0.5	1.2	14.80	33.30	4.73	19.20	4.72	3.52	4.93	0.87	4.82	1.07	3.05	0.52	3.35	0.45
16	4.0	0.4	0.5	0.6	10.20	20.00	3.05	15.10	4.61	4.43	5.12	0.96	5.76	1.32	4.28	0.78	4.87	0.62
17	9.0	0.8	0.5	0.8	4.90	13.10	1.83	7.40	1.58	0.41	1.30	0.20	0.95	0.20	0.49	0.07	0.52	0.07
18	11.0	1.3	0.5	1.0	46.00	124.60	18.63	78.00	16.29	4.65	12.70	2.43	15.06	3.29	9.77	1.65	10.79	1.39
19	8.0	0.8	2.8	0.4	13.70	51.30	5.74	25.40	6.19	1.96	6.53	1.46	9.75	2.30	6.65	1.11	7.32	0.92
20	7.0	0.2	0.5	4.1	19.50	34.20	4.44	15.00	1.62	0.39	1.11	0.12	0.63	0.10	0.27	n.a.	0.19	0.03
21	10.0	0.3	3.1	1.0	6.80	14.20	2.32	14.00	3.71	0.72	3.34	0.54	3.02	0.56	1.61	0.26	1.61	0.24
22	9.0	0.5	1.6	1.0	20.40	42.40	5.01	19.00	3.52	0.75	3.25	0.54	2.90	0.61	1.66	0.27	1.61	0.24
23	90.0	1.4	3.3	2.1	3.10	6.20	0.78	3.10	0.53	0.34	0.50	0.09	0.52	0.08	0.27	0.05	0.28	0.04
24	72.0	2.5	0.5	2.0	2.40	4.40	0.61	2.40	0.39	0.36	0.50	0.08	0.53	0.08	0.26	0.04	0.20	0.03
25	139.0	9.1	24.8	210.5	22.00	45.30	5.75	22.70	5.20	1.09	5.38	0.92	5.51	1.15	3.73	0.70	4.03	0.68
26	38.0	126.3	6.3	0.1	6.10	10.80	1.43	5.90	1.07	0.65	0.98	0.17	1.00	0.19	0.60	0.09	0.60	0.09
27	76.0	29.4	4.5	1.4	6.00	12.30	1.44	6.00	1.10	1.26	1.08	0.18	0.91	0.17	0.52	0.08	0.45	0.07

n.a. = not analyzed

of the Varied Group, Moldanubian Zone (Hrdličková ed. 2008). The skarns are characterised by significant variation in the REE contents. The normalized REE patterns resemble those for sediments, 0.2–0.8 Ga old (Taylor and McLennan 1985; Condie 1993) with typical 100-fold enrichment in LREE, 10-fold enrichment

in HREE and a slightly negative Eu anomaly. The LREE/HREE ratio is thus ~ 10 , which corresponds to the values for post-Archaean sedimentary rocks (Fig. 8).

The Eu/Eu* values span a wide range of 0.5 to 3.5 but the majority of samples are characterised by negligible

magnitude of the Eu anomaly. The highest Eu/Eu* values are typical of samples with low Σ REE. Differences in the contents of the major elements, total REE contents and Eu/Eu* values occur not only among individual localities but also among samples from the same locality.

Skarns with abundant garnet are enriched in HREE, which results in a rather flat REE distribution pattern. Clinopyroxene skarns from Rešice (sample 17) and Budeč (5), exhibiting the lowest values of Σ REE, contain low Al₂O₃ and TiO₂.

Magnetite-bearing skarn from Rešice contains extremely low SiO₂, high Fe₂O₃, and somewhat elevated Cu, As, Sn and Zn. Skarn from Budeč (sample 5) shows remarkably low Zr, Sr and Rb. All the samples from Budeč and Slatina are enriched in Sn, up to 560 ppm.

Skarns from Vlastějovice, Holšice and Zliv (samples 20–25) are split in two groups. The first group features REE abundances resembling detrital sediments (Holšice, Vápenka and Vlastějovice, sample Nos 20–22, 25). Sample 20 exhibits extreme enrichment in LREE and low HREE abundances, combined with low contents of lithophile elements Rb, Ba and K. The second group is represented by skarn from Zliv (samples 23 and 24) with extremely low LREE as well as HREE abundances (10× chondrite) and a prominent positive Eu anomaly (Eu/Eu* = 2.02–2.49). The two samples have high Mg and Ca contents, as well as increased Co, Pb and Zn with low Al₂O₃, TiO₂ and Zr concentrations.

5.3. Kutná Hora Complex

Samples from four localities represent mainly Grt–Cpx assemblages with variable contents of the major elements: 22–55 wt. % SiO₂, 2–14 wt. % Al₂O₃, 14–31 wt. % CaO, 2–58 wt. % Fe₂O₃ and 0.5–4 wt. % MgO. The trace-element abundances are variable for the individual localities as well as for the individual samples (Tab. 11). The samples from Točice and Církvice (Nos 66, 68–69) exhibit REE and other trace-element contents alike post-Archaeon sedimentary rocks. The chondrite-normalized patterns are also similar to many samples from the Moldanubian Zone and the Svatka Unit, with typical 100-fold enrichment in LREE and 10 fold enrichment in HREE.

Skarns from Malešov (samples 60 and 61), Jakub near Církvice (sample 67) and Církvice (sample 65) have lower contents of REE and other trace elements. The samples from Malešov (Nos 58 and 59) are largely different, with low abundances of REE and absence of enrichment in LREE compared with HREE. All these samples with relatively low REE (Nos 58–59, 61, 65 and 67) have low Al₂O₃ and TiO₂ contents. Sample 67 has a positive Eu anomaly (Eu/Eu* = 3.5).

6. Oxygen isotopes

The oxygen isotopic composition in garnets, pyroxenes and amphiboles from skarns of the Svatka Unit is given in Tab. 12. The Fig. 11 indicates that equilibrium was probably reached for most of the co-existing garnet–clinopyroxene pairs. The $\delta^{18}\text{O}$ values for garnet and clinopyroxene vary in the interval of –0.1 to +4.1 ‰. Only the sample SV 297 exhibits isotope disequilibrium between garnet and clinopyroxene.

7. Geochronological data

The isotopic ratios obtained from 106 zircon grains from samples SV 25 and SV 262 (Svatouch) are concordant or plot close to the Concordia (Fig. 12, Tab. 13). The U–Pb ages for individual zircon grains range broadly from 500 Ma to 2.6 Ga with a marked maximum between 540 and 600 Ma, accompanied by minor peaks at ~1.8 Ga, ~1.9–2.0 Ga, ~2.45 Ga and ~2.6 Ga. Such a span of ages and the presence of marked age maxima are typical of detrital zircon populations and thus indicative of a sedimentary protolith of the studied sample.

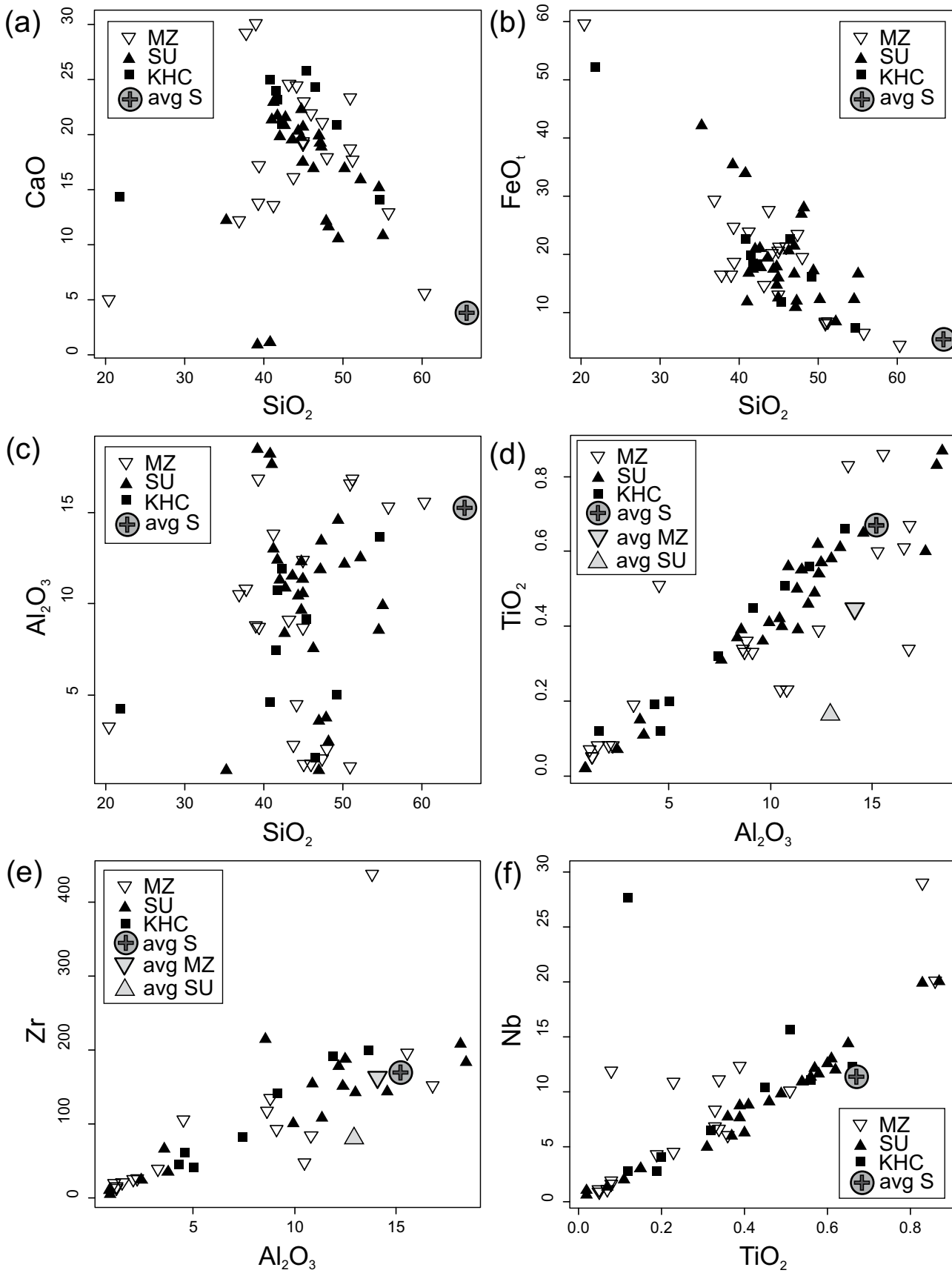
8. Discussion

8.1. Metamorphic development

8.1.1. The Svatka Unit

Pressure-temperature development was studied in mica schists and paragneisses in the proximity of skarn bodies (Pertoldová et al. 2007). Structural relations and element partition among minerals indicate three metamorphic stages: M₁ early prograde, M₂ prograde/peak and M₃ retrograde. The metamorphic conditions attained high-pressure conditions, corresponding to the kyanite stability field, followed by nearly isothermal decompression characterized by crystallization of sillimanite.

The effects of stage M₁, characterized by prograde garnet zoning, were observed in all the studied samples. Stage M₂ is defined on the basis of mineral assemblages with garnet, clinopyroxene, muscovite, biotite, staurolite and kyanite, indicating a metamorphic peak. This stage was characterised by an increase in the P–T conditions straddling the boundary of the amphibolite-facies field to the adjacent lower granulite- and eclogite-facies domains. Phase M₃ was related to decompression returning back to the amphibolite-facies P–T conditions. Chlorite and sillimanite formed as a result of re-equilibration of the M₁ assemblages. Retrogressive sillimanite appears as a reaction product from staurolite or micas in mica schists.



Tab. 11 Whole-rock major- and trace-element compositions of skarns from the Kutná Hora Complex. Major oxides in wt. %, trace elements in ppm, Au in ppb

Sample No.	SiO ₂	TiO ₂	Al ₂ O ₃	Fe ₂ O ₃	MgO	MnO	CaO	Na ₂ O	K ₂ O	P ₂ O ₅	Total	Ba	Co	Rb	Sn	Sr	Zr	Pb
58	21.91	0.19	4.28	58.05	0.69	0.19	14.31	0.29	0.08	0.07	99.90	3	17.3	2.2	75	23.9	45.0	1.0
59	40.82	0.12	4.60	25.13	2.93	0.32	24.98	0.37	0.01	0.09	99.99	2	12.9	0.4	103	38.1	60.8	0.8
60	41.72	0.51	10.74	6.18	1.88	0.96	23.15	0.27	0.04	0.15	100.02	n.a.	n.a.	3.4	51.7	122.5	n.a.	10.7
61	46.52	0.12	1.59	6.64	2.02	1.07	24.34	0.22	0.03	0.05	99.84	n.a.	n.a.	2.3	10.3	87.9	n.a.	11.6
65	41.52	0.32	7.43	22.09	1.96	0.52	23.95	0.17	0.01	0.14	99.85	2	19.4	0.3	285	419.4	82.8	1.1
66	42.32	0.56	11.91	20.38	1.98	0.52	20.90	0.26	0.30	0.19	99.77	48	18.0	35.2	103	839.0	191.4	1.7
67	49.26	0.20	5.03	1.95	4.01	0.54	20.83	0.26	0.03	0.12	99.77	n.a.	n.a.	19	275	n.a.	41	n.a.
68	45.41	0.45	9.16	13.10	2.79	1.75	25.81	0.18	0.02	0.17	99.77	4	13.0	0.9	242	259.6	142.0	1.4
69	54.71	0.66	13.67	8.21	2.70	0.80	14.07	3.12	1.53	0.18	99.76	462.1	9.8	60.4	89	1335.5	199.5	1.1

Sample No.	Zn	Bi	Au	La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
58	36	0.5	1.4	1.50	3.70	0.62	4.20	1.27	0.29	1.29	0.19	1.05	0.21	0.68	0.09	0.57	0.08
59	4	5.3	1.8	1.30	4.20	0.73	4.40	1.24	0.40	1.22	0.20	1.11	0.21	0.58	0.09	0.63	0.11
60	104.3	n.a.	n.a.	14.36	29.12	3.75	18.33	4.97	1.12	4.50	0.00	4.18	0.74	1.36	0.22	2.48	0.41
61	51.5	n.a.	n.a.	10.82	19.33	1.78	10.19	1.90	0.32	1.66	0.39	1.94	0.38	1.28	n.a.	0.85	0.14
65	6	1.7	1.2	9.90	18.60	2.97	12.80	3.13	0.52	3.02	0.50	2.57	0.55	1.56	0.24	1.45	0.21
66	10	12.3	72.2	31.90	60.30	7.63	29.70	5.30	1.62	5.00	0.80	4.30	0.84	2.48	0.39	2.35	0.35
67	60	n.a.	n.a.	13.33	22.64	2.09	11.25	2.60	2.36	1.54	n.a.	2.92	0.62	1.98	0.24	1.72	0.27
68	12	<0.1	<0.5	38.20	65.40	8.78	34.70	5.66	2.37	4.91	0.80	3.98	0.78	2.15	0.33	2.17	0.30
69	6	0.1	3.4	26.60	51.70	6.82	26.40	5.05	0.92	4.15	0.75	4.26	0.79	2.24	0.32	2.22	0.33

n.a. = not analyzed

Chlorite is a late retrogression mineral, which locally replaced biotite and garnet.

Overall, the prograde P-T path for rocks of the Svratka Unit begun at *c.* 610 °C and 0.55 GPa, continuing to the peak at ~0.7–0.85 GPa and max. 720 °C for mica schists, paragneisses and migmatites and up to 1.4 GPa for skarns (see Fig. 13). The retrograde part of the path started with sillimanite and chlorite crystallization in mica schists as well as by symplectite and corona structures growth in skarns.

Our samples with titanite-plagioclase symplectite (titanite II) replacing a Al- and F-bearing titanite (titanite I) are similar to titanites in eclogites from Central China (Carswell et al. 1996). These authors described two types of breakdown reactions for high-Al titanites: 1) ilmenite-amphibole symplectite, 2) symplectite of low-Al, F titanite with plagioclase. Aluminium and fluorine contents in titanite I in our samples are distinctly lower

↩

Fig. 7 Geochemical variation diagrams for skarns from all three studied units – the Moldanubian zone (MZ), the Svratka Unit (SU) and the Kutná Hora Complex (KHC) exhibit common features. The compositions of average post-Archaean sedimentary rocks (avg S, Taylor and McLennan 1985), average metasediments from the Moldanubian Zone – Strážek Unit (avg MZ, Hrdličková ed. 2008) and average metasediments from the Svratka Unit (avg SU, Hanžl ed. 2008) are plotted for comparison. There is a fair negative correlation between SiO₂ and CaO, FeO₁ and positive correlation for Al₂O₃ and TiO₂. The Al₂O₃ contents are not dependent on the SiO₂, as is a common feature for sedimentary clastic rocks. The contents of Zr and Nb in most skarns are lower than in clastic sediments, as are those of Al and Ti.

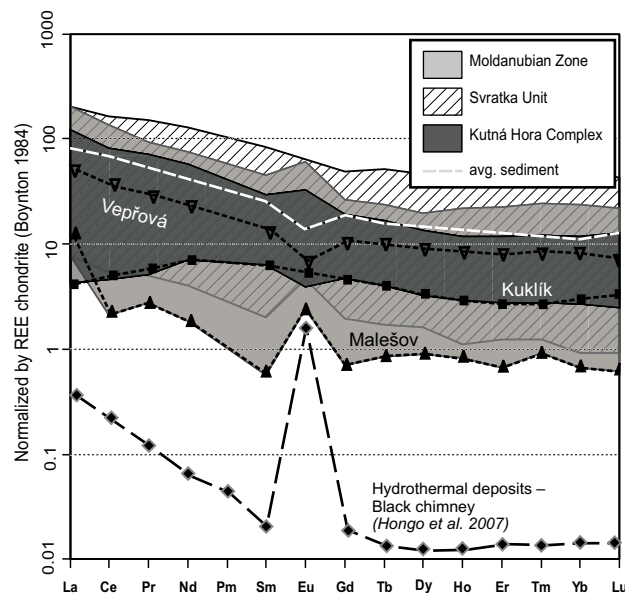


Fig. 8 The REE contents in skarns from all three studied units normalized to chondritic abundances (Boynton 1984). High variability of the REE contents within the individual units can be seen and compared with the REE trend for average post-Archaean sediments. The progressive fractionation of LREE and increasing magnitude of the positive Eu anomaly are characteristic for some of the studied skarns, resembling the trend known from hydrothermal rift-related deposits on the sea floor (e.g. Hongo et al. 2007). A pattern typical of skarns with prevailing clastic sedimentary content is depicted on an example from Vepřová (No. 13), a lower content of detritic material is demonstrated by the REE trend of Kuklík (42) and the influence of hydrothermal activity on the REE composition is characteristic of the Malešov (59) sample.

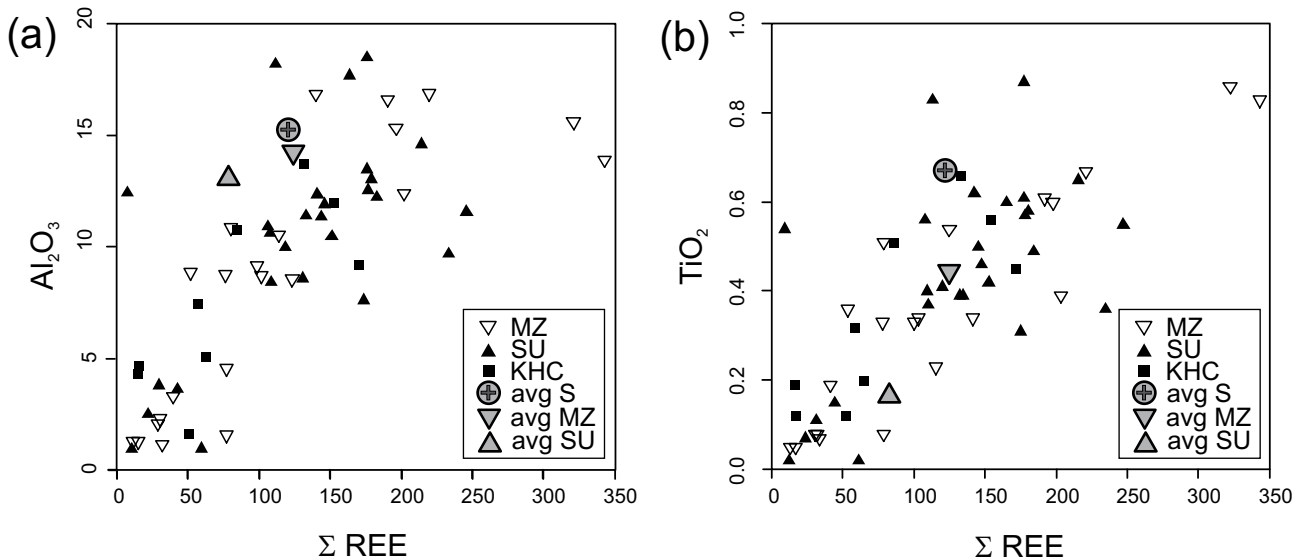


Fig. 9 Geochemical variation diagrams Al_2O_3 vs. ΣREE and TiO_2 vs. ΣREE for the skarns from the Moldanubian Zone (MZ) and the Svratka Unit (SU).

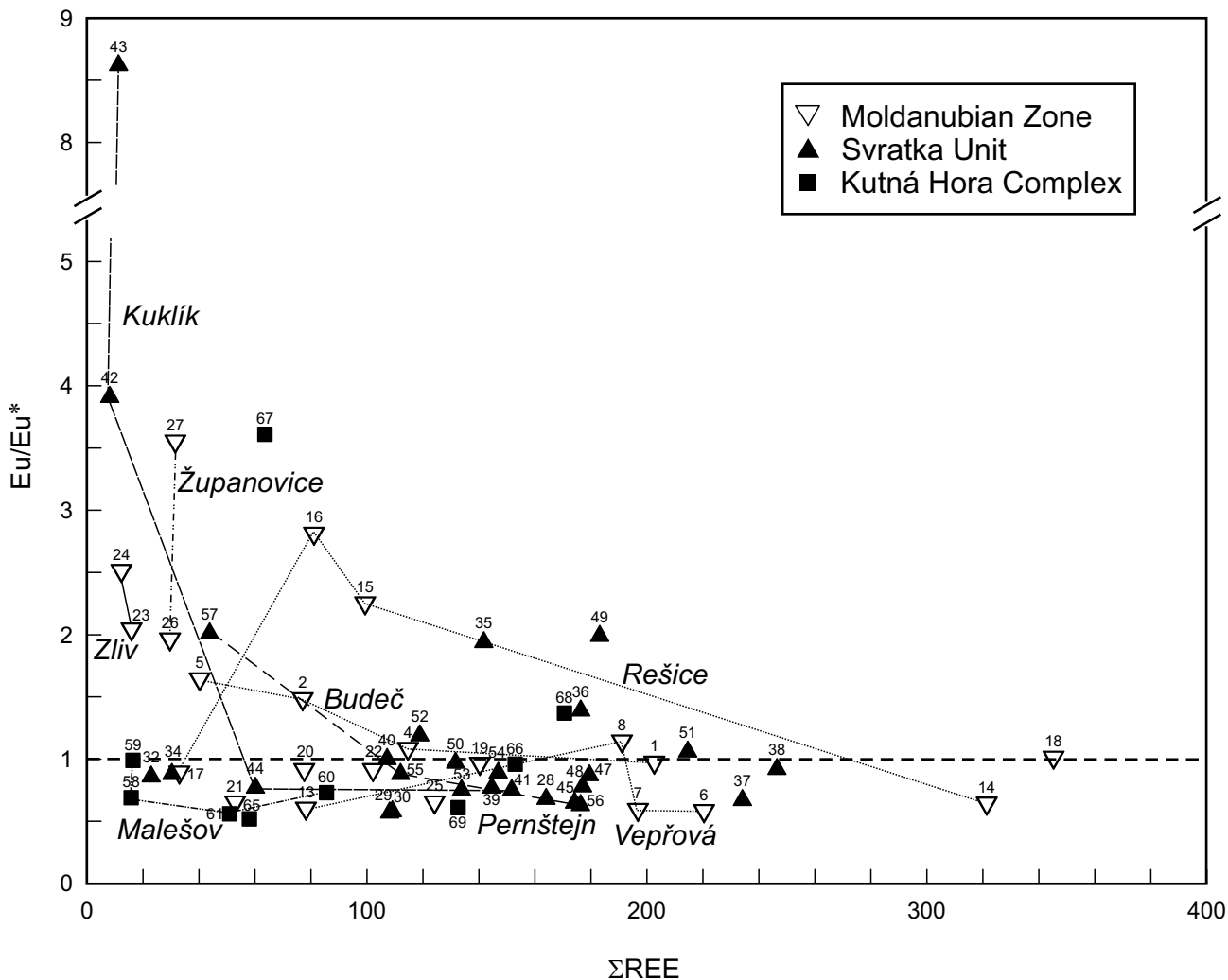


Fig. 10 The diagram of Eu anomaly (Eu/Eu^*) vs. ΣREE after normalization to chondrites (Boynton 1984). A positive Eu anomaly is characteristic for products of hydrothermal activity near submarine hydrothermal veins. The lower the hydrothermal activity and the lower temperature of circulating fluids, the lesser is their influence on the composition of the surrounding sediments (Chavagnac et al. 2005). Even the skarn samples from the same locality show large variations in their REE contents and patterns, suggesting that the influence of hydrothermal fluids was very local.

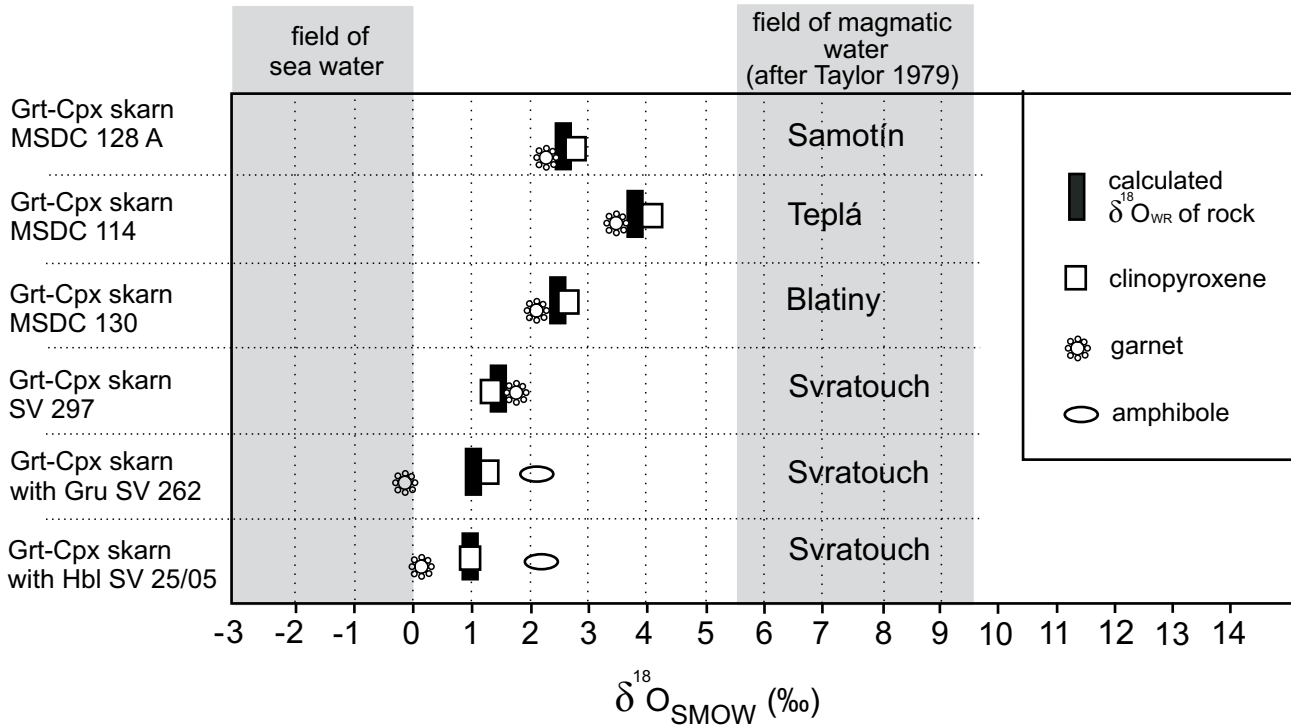


Fig. 11 The $\delta^{18}\text{O}_{\text{SMOW}}$ (‰) values for garnet, clinopyroxene, grunerite and hornblende from skarns (the Svatka Unit). Sea water after Hoefs (2009).

Tab. 12 Oxygen isotope composition expressed as $\delta^{18}\text{O}\%$ (SMOW) for skarn minerals from the Svatka Unit

locality	sample	$\delta^{18}\text{O}\%$ Cpx	$\delta^{18}\text{O}\%$ Grt	$\delta^{18}\text{O}\%$	
				Hbl, Gru	calculated
Samotín	Gr-Cpx skarn MSDC128A	2.80	2.30		2.58
Teplá	Gr-Cpx skarn MSDC114	4.10	3.50		3.80
Blatiny	Gr-Cpx skarn MSDC130	2.70	2.10		2.44
Svratouch	Gr-Cpx skarn SV 297	1.30	1.80		1.46
Svratouch	Gr-Cpx skarn with Gru SV262	1.30	-0.10	2.10	1.05
Svratouch	Gr-Cpx skarn with Hbl SV25	1.00	0.10	2.20	0.86

Tab. 13 Analytical data obtained by LA ICP-MS U-Pb zircon dating

Analysis #	Isotopic ratios						Calculated ages (Ma)							
	$^{207}\text{Pb}/^{235}\text{U}$	$\pm 1\sigma$	$^{206}\text{Pb}/^{238}\text{U}$	$\pm 1\sigma$	$^{207}\text{Pb}/^{206}\text{Pb}$	$\pm 1\sigma$	$^{207}\text{Pb}/^{235}\text{U} \pm 1\sigma$	$^{206}\text{Pb}/^{238}\text{U} \pm 1\sigma$	$^{207}\text{Pb}/^{206}\text{Pb} \pm 1\sigma$	Concordia	$\pm 1\sigma$			
1	0.8197	0.0360	0.1014	0.0024	0.0586	0.0020	608	20	623	14	554	73	618	13
2	0.7605	0.0357	0.0901	0.0018	0.0612	0.0025	574	21	556	11	648	88	559	10
3	0.7896	0.0364	0.0904	0.0018	0.0633	0.0027	591	21	558	10	719	91	562	10
4	0.6923	0.0223	0.0890	0.0023	0.0564	0.0013	534	13	550	14	469	52	542	11
5	0.6068	0.0279	0.0851	0.0026	0.0517	0.0014	482	18	526	15	274	61	508	13
6	0.7030	0.0368	0.0952	0.0020	0.0536	0.0021	541	22	586	12	353	89	578	11
7	6.2488	0.1607	0.3648	0.0064	0.1242	0.0025	2011	23	2005	30	2018	36	2009	21
8	3.3171	0.1711	0.2461	0.0092	0.0978	0.0034	1485	40	1418	47	1582	65	1459	36
9	0.6511	0.0443	0.0903	0.0028	0.0523	0.0026	509	27	557	17	298	115	547	15
10	1.5603	0.0531	0.1706	0.0034	0.0663	0.0016	955	21	1016	19	816	51	988	16
11	0.7164	0.0227	0.0883	0.0019	0.0588	0.0016	549	13	546	11	560	60	547	10
12	11.9667	0.3294	0.4833	0.0118	0.1796	0.0030	2602	26	2542	51	2649	27	2599	26
13	0.7583	0.0383	0.0980	0.0026	0.0561	0.0025	573	22	603	16	457	99	594	14
14	0.7204	0.0410	0.0892	0.0020	0.0585	0.0027	551	24	551	12	550	103	551	11

Tab. 13 continued

Analysis #	Isotopic ratios						Calculated ages (Ma)							
	$^{207}\text{Pb}/^{235}\text{U}$	$\pm 1\sigma$	$^{206}\text{Pb}/^{238}\text{U}$	$\pm 1\sigma$	$^{207}\text{Pb}/^{206}\text{Pb}$	$\pm 1\sigma$	$^{207}\text{Pb}/^{235}\text{U} \pm 1\sigma$	$^{206}\text{Pb}/^{238}\text{U} \pm 1\sigma$	$^{207}\text{Pb}/^{206}\text{Pb} \pm 1\sigma$	Concordia	$\pm 1\sigma$			
15	0.7255	0.0157	0.0913	0.0016	0.0576	0.0008	554	9	563	9	516	31	559	8
16	0.7073	0.0242	0.0908	0.0017	0.0565	0.0015	543	14	560	10	473	58	555	9
17	0.7626	0.0317	0.0954	0.0023	0.0580	0.0019	575	18	588	13	528	71	584	12
18	0.6070	0.0434	0.0828	0.0017	0.0532	0.0034	482	27	513	10	337	144	510	10
19	0.7874	0.0381	0.0911	0.0015	0.0627	0.0029	590	22	562	9	699	98	564	9
20	0.6782	0.0315	0.0850	0.0019	0.0579	0.0025	526	19	526	11	525	96	526	10
21	4.6722	0.1532	0.3112	0.0082	0.1089	0.0014	1762	27	1747	40	1781	23	1759	26
22	0.7210	0.0317	0.0929	0.0027	0.0563	0.0018	551	19	573	16	463	73	565	14
23	0.7070	0.0218	0.0888	0.0022	0.0577	0.0017	543	13	549	13	520	65	546	11
24	5.5354	0.1738	0.3347	0.0098	0.1200	0.0018	1906	27	1861	47	1956	27	1902	27
25	0.6449	0.0265	0.0864	0.0016	0.0541	0.0020	505	16	534	10	376	82	528	9
26	5.8659	0.1864	0.3394	0.0106	0.1254	0.0025	1956	28	1884	51	2034	35	1953	28
27	0.8466	0.0346	0.1058	0.0027	0.0581	0.0019	623	19	648	16	532	70	639	14
28	0.7012	0.0647	0.0955	0.0037	0.0532	0.0043	539	39	588	22	338	183	578	20
29	0.7538	0.0287	0.0951	0.0019	0.0575	0.0018	570	17	586	11	510	70	582	10
30	0.7226	0.0404	0.0933	0.0027	0.0562	0.0024	552	24	575	16	458	94	569	15
31	0.7411	0.0306	0.0940	0.0018	0.0572	0.0018	563	18	579	11	499	69	576	10
32	4.7553	0.1118	0.3069	0.0067	0.1124	0.0016	1777	20	1725	33	1838	26	1772	20
33	0.6957	0.0416	0.0968	0.0023	0.0521	0.0024	536	25	596	13	291	107	585	13
34	1.1507	0.0665	0.1356	0.0039	0.0616	0.0025	778	31	820	22	659	88	807	20
35	0.7174	0.0425	0.0917	0.0024	0.0568	0.0026	549	25	565	14	483	101	562	13
36	0.6744	0.0250	0.0899	0.0020	0.0544	0.0016	523	15	555	12	387	68	544	10
37	0.6446	0.0257	0.0870	0.0019	0.0537	0.0018	505	16	538	11	360	75	528	10
38	0.6656	0.0227	0.0905	0.0017	0.0533	0.0017	518	14	559	10	343	70	546	9
39	4.5514	0.1145	0.3145	0.0065	0.1050	0.0016	1740	21	1763	32	1714	27	1744	20
40	5.8817	0.1398	0.3671	0.0086	0.1162	0.0022	1959	21	2016	40	1899	34	1959	21
41	5.9880	0.1655	0.3596	0.0087	0.1208	0.0025	1974	24	1980	41	1968	37	1975	24
42	0.7416	0.0293	0.0972	0.0026	0.0554	0.0019	563	17	598	15	427	75	583	13
43	5.6645	0.1743	0.3452	0.0093	0.1190	0.0019	1926	27	1912	44	1941	29	1924	26
44	0.7047	0.0412	0.0909	0.0029	0.0562	0.0026	542	25	561	17	462	104	555	16
45	0.6666	0.0415	0.0914	0.0025	0.0529	0.0027	519	25	564	15	324	117	554	14
46	4.8740	0.2954	0.3362	0.0101	0.1051	0.0044	1798	51	1869	49	1717	77	1834	39
47	0.6743	0.0232	0.0911	0.0020	0.0537	0.0015	523	14	562	12	359	62	547	10
48	0.6685	0.0223	0.0880	0.0022	0.0551	0.0012	520	14	544	13	417	50	533	11
49	0.6972	0.0226	0.0883	0.0019	0.0573	0.0015	537	13	545	11	502	59	542	10
50	0.7999	0.0451	0.0932	0.0023	0.0622	0.0034	597	25	575	13	682	116	578	13
51	0.6673	0.0316	0.0897	0.0023	0.0539	0.0024	519	19	554	14	369	100	544	12
52	0.6611	0.0217	0.0903	0.0022	0.0531	0.0014	515	13	557	13	333	59	536	11
53	0.6647	0.0368	0.0879	0.0024	0.0548	0.0023	517	22	543	14	405	92	537	13
54	0.6660	0.0568	0.0883	0.0025	0.0547	0.0034	518	35	545	15	401	139	542	14
55	0.7281	0.0203	0.0988	0.0025	0.0534	0.0009	555	12	607	15	348	40	571	11
56	1.1501	0.0886	0.1248	0.0062	0.0668	0.0023	777	42	758	35	832	73	765	31
57	4.5589	0.1307	0.3117	0.0080	0.1061	0.0017	1742	24	1749	39	1733	30	1743	23
58	10.6351	0.2784	0.4569	0.0124	0.1688	0.0022	2492	24	2426	55	2546	21	2495	24
59	0.6522	0.0278	0.0829	0.0024	0.0570	0.0020	510	17	514	14	493	78	512	12
60	5.5391	0.2076	0.3448	0.0104	0.1165	0.0018	1907	32	1910	50	1903	28	1907	31
61	9.8736	0.1911	0.4525	0.0078	0.1582	0.0018	2423	18	2406	34	2437	19	2422	18
62	0.6906	0.0396	0.0874	0.0023	0.0573	0.0029	533	24	540	14	503	110	539	13
63	5.7865	0.1279	0.3652	0.0054	0.1149	0.0020	1944	19	2007	26	1878	31	1962	17
64	0.8394	0.0408	0.0975	0.0018	0.0625	0.0029	619	23	600	11	690	98	602	10
65	0.7408	0.0255	0.0931	0.0019	0.0577	0.0011	563	15	574	11	520	42	570	10
66	0.7971	0.0277	0.0945	0.0019	0.0612	0.0018	595	16	582	11	646	65	585	10

Tab. 13 continued

Analysis #	Isotopic ratios						Calculated ages (Ma)							
	$^{207}\text{Pb}/^{235}\text{U}$	$\pm 1\sigma$	$^{206}\text{Pb}/^{238}\text{U}$	$\pm 1\sigma$	$^{207}\text{Pb}/^{206}\text{Pb}$	$\pm 1\sigma$	$^{207}\text{Pb}/^{235}\text{U} \pm 1\sigma$	$^{206}\text{Pb}/^{238}\text{U} \pm 1\sigma$	$^{207}\text{Pb}/^{206}\text{Pb} \pm 1\sigma$	Concordia $\pm 1\sigma$				
67	1.1928	0.0545	0.1377	0.0032	0.0628	0.0017	797	25	832	18	702	56	822	16
68	0.6605	0.0302	0.0846	0.0023	0.0566	0.0016	515	18	524	14	476	61	521	13
69	0.6740	0.0229	0.0839	0.0021	0.0583	0.0011	523	14	519	12	541	41	521	11
70	12.3879	0.3170	0.4917	0.0109	0.1827	0.0023	2634	24	2578	47	2678	20	2631	24
71	0.8161	0.0302	0.0963	0.0023	0.0614	0.0020	606	17	593	14	655	71	597	12
72	0.6319	0.0226	0.0823	0.0019	0.0557	0.0014	497	14	510	11	441	57	505	10
73	0.6717	0.0129	0.0869	0.0011	0.0561	0.0009	522	8	537	6	455	36	532	6
74	0.7357	0.0226	0.0928	0.0014	0.0575	0.0016	560	13	572	8	511	60	569	8
75	4.9088	0.2308	0.3292	0.0098	0.1081	0.0021	1804	40	1834	48	1768	35	1815	35
76	0.7453	0.0204	0.0905	0.0014	0.0598	0.0013	565	12	558	8	595	48	560	8
77	0.7294	0.0321	0.0920	0.0013	0.0575	0.0020	556	19	567	8	512	77	566	8
78	0.6953	0.0191	0.0866	0.0015	0.0582	0.0013	536	11	535	9	539	48	536	8
79	6.4373	0.1658	0.3690	0.0081	0.1265	0.0020	2037	23	2025	38	2050	28	2036	22
80	0.7626	0.0365	0.0959	0.0028	0.0577	0.0016	575	21	590	16	517	62	585	15
81	0.7845	0.0277	0.0960	0.0022	0.0593	0.0013	588	16	591	13	577	48	590	11
82	0.7002	0.0181	0.0911	0.0014	0.0557	0.0013	539	11	562	8	442	53	555	7
83	0.8703	0.0239	0.1039	0.0015	0.0607	0.0016	636	13	637	9	630	59	637	8
84	0.6627	0.0224	0.0838	0.0019	0.0573	0.0014	516	14	519	12	504	54	518	10
85	0.7478	0.0239	0.0934	0.0015	0.0581	0.0016	567	14	575	9	533	60	573	8
86	11.8982	0.4510	0.4895	0.0144	0.1763	0.0032	2596	36	2568	62	2618	30	2593	35
87	0.7416	0.0217	0.0921	0.0016	0.0584	0.0014	563	13	568	9	545	51	567	8
88	0.8085	0.0327	0.0971	0.0025	0.0604	0.0024	602	18	597	15	618	86	599	13
89	4.8100	0.2406	0.3149	0.0105	0.1108	0.0029	1787	42	1765	52	1812	48	1779	38
90	10.5631	0.3155	0.4558	0.0082	0.1681	0.0034	2485	28	2421	36	2539	34	2464	25
91	0.7800	0.0281	0.0960	0.0019	0.0589	0.0015	585	16	591	11	565	55	589	10
92	0.7037	0.0345	0.0919	0.0018	0.0555	0.0021	541	21	567	11	433	83	563	10
93	0.7076	0.0321	0.0905	0.0019	0.0567	0.0017	543	19	559	11	479	65	556	11
94	0.7479	0.0233	0.0859	0.0018	0.0631	0.0019	567	14	531	11	712	63	543	10
95	0.7540	0.0245	0.0952	0.0015	0.0574	0.0019	571	14	586	9	509	71	583	8

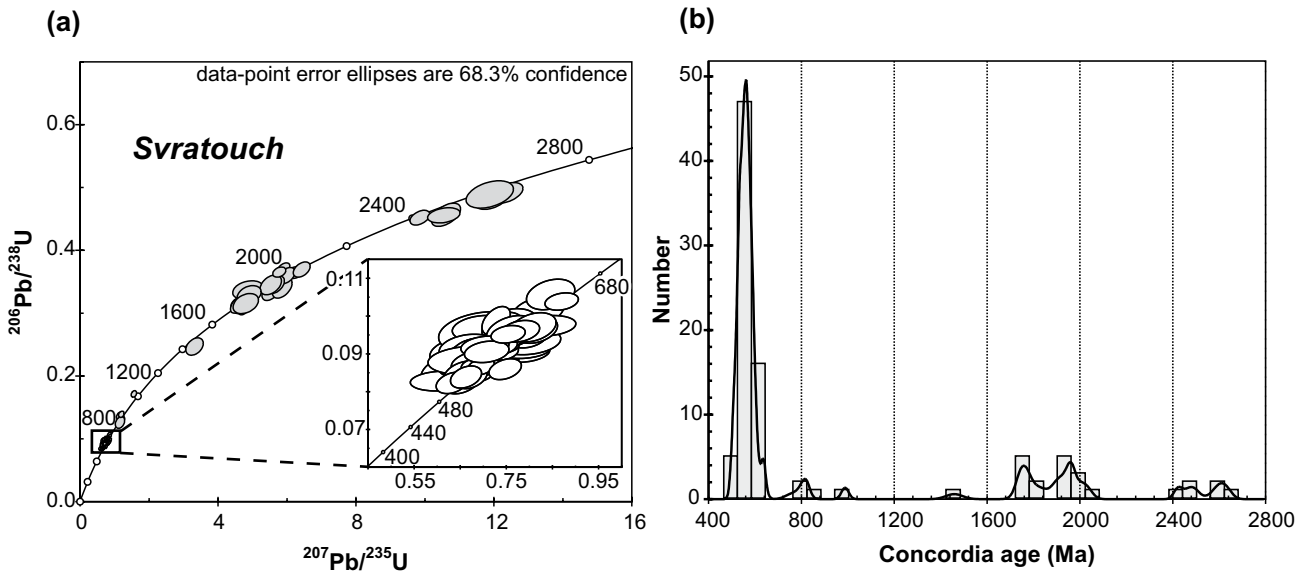


Fig. 12a – Concordia diagram for concordant population of detrital zircons from the Svatouch sample. Lower-right inset presents the measured data in zoomed portion of the Concordia diagram for interval between 400 and 700 Ma; b – Frequency curve and histogram of zircon ages for the same sample as in (a).

than in titanites described from China. Nevertheless, the decompression-related symplectites in our samples, composed of titanite II and plagioclase, clearly formed from early titanite I with increased Al, and were preserved locally as small relics.

The proposed P-T-t evolutionary scheme for the skarn bodies and surrounding rocks is based on integrated data on petrology, geochronology and the structural patterns. On a regional scale, there are preserved some unequivocal relics of the original intrusive contacts between the Svatka metagranites and the surrounding migmatites (Verner et al. this volume). An old generation of isoclinal and rootless folds in the skarns and migmatites pre-dates this intrusion (Verner et al. this volume). As a consequence of widespread and intense Variscan deformation and recrystallization of paragneisses and migmatites in the Svatka Unit, relict domains of pre-Variscan mineral assemblages are rare. The emplacement of Svatka metagranite was dated at ~515 Ma (Schulmann et al. 2005). These rocks were affected only by a relatively weak deformation and partial recrystallization during the Variscan Orogeny (see Buriánek et al. this volume).

8.1.2. Moldanubian Zone

The pressure-temperature conditions in the northern part of the Moldanubian Zone were estimated for rocks in the proximity of skarns, i.e., calc-silicate gneisses, granulites and migmatized paragneisses. Pertoldová et al. (2007) reported P-T conditions for three stages of metamorphism. Phase M₁ corresponded to the peak metamorphic conditions (T= 660 °C and P = 0.6 GPa). The P-T conditions for the HT/LP phase M₂, leaving its imprint in impure marbles, corresponded to 550 °C and 0.2 GPa (Novák 1989). The retrograde metamorphism M₃ resulted in crystallization of chlorite, replacing amphibole, plagioclase and epidote. The presence of the younger andradite-rich garnet generation, cutting the older, corroded garnet grains, indicates a sharp increase in oxygen fugacity, most probably caused by ingress of hydrothermal fluids. It is so far uncertain whether the hydrothermal fluids were igneous-related or of regional metamorphic origin.

Tajčmanová et al. (2006) assessed the P-T conditions of equilibration in granulites and migmatized paragneisses. The P-T conditions were estimated at ~850 °C, 1.8 GPa for the granulites, and at 870 °C and 0.8–1.1 GPa for migmatized paragneisses. These authors described the polyphase retrogressive character of decompression, corresponding to a temperature of 620 °C and pressure of 0.4–0.6 GPa. Schulmann et al. (2008) also reported results for several samples documenting prograde and retrograde paths of metamorphism (see Fig. 13).

8.1.3. Kutná Hora Complex

Vrána et al. (this volume) deciphered three metamorphic events in gneisses, migmatites and granulites of the Běstvína Unit (Kutná Hora Complex). Event M₁ resulted in crystallization of HP/HT granulites in the Běstvína Unit at 840–920 °C and 1.8–2.2 GPa (i.e., eclogite-facies conditions). Nahodilová et al. (2006) reported peak conditions of 831 ± 53 °C and 1.65 ± 1.8 GPa followed by retrogression at 705 ± 97 °C and 1.4 ± 0.2 GPa, based on study of different samples from the Běstvína Unit. For the second event M₂, Nahodilová et al. (2006) estimated the P-T conditions in migmatites at 875 ± 95 °C and 1.56 ± 0.14 GPa. Data for MP/LT retrogression indicate T = 712 ± 39 °C and P = 1.06 ± 0.18 GPa. The P-T path of eclogite from the new Roztěž locality in the KHC was estimated by means of the pseudosection method. The modelling indicated minimum pressures above 2.15–2.30 GPa and temperatures 600–650 °C reached during the prograde part of metamorphism (Štědrá and Nahodilová this volume). The high-jadeite clinopyroxene in the kyanite-bearing eclogite from Bořetice indicates even higher pressures than those inferred from the Roztěž eclogite and from the Běstvína Unit – 761°C/4.3 GPa (Štědrá and Nahodilová this volume)

Metamorphic event M₃ was characterized by pressure of 0.9 to 1.2 GPa in the Ky–Grt felsic gneisses (Vrána et al. this volume). No suitable mineral pair was found for a temperature estimate.

The multistage tectonometamorphic history of the Kutná Hora Complex has already been assumed by Synek and Oliveriová (1993) but their study lacked exact geothermobarometric data. Observations by these authors as well as by Nahodilová et al. (2006) and Vrána et al. (this volume) show that the metamorphic development of the Kutná Hora Complex was essentially different from that of the Moldanubian Zone. In the former unit, decompression took place mainly in the kyanite stability field. On the other hand, decompression in the Moldanubian Zone was marked by crystallization of sillimanite, replacement of kyanite by spinel or low-Ca garnet and production of cordierite (Tajčmanová et al. 2006).

8.2. Geochemistry

The studied skarn samples from the Svatka Unit, the Moldanubian Zone and the Kutná Hora Complex experienced a complicated history of polyphase metamorphism. Geochemical data document large variability in the major- and trace-element compositions, including REE, among individual localities and also among samples from the same locality. The differences in REE abundances cannot be explained solely on the basis of differences in clinopyroxene and garnet contents, or

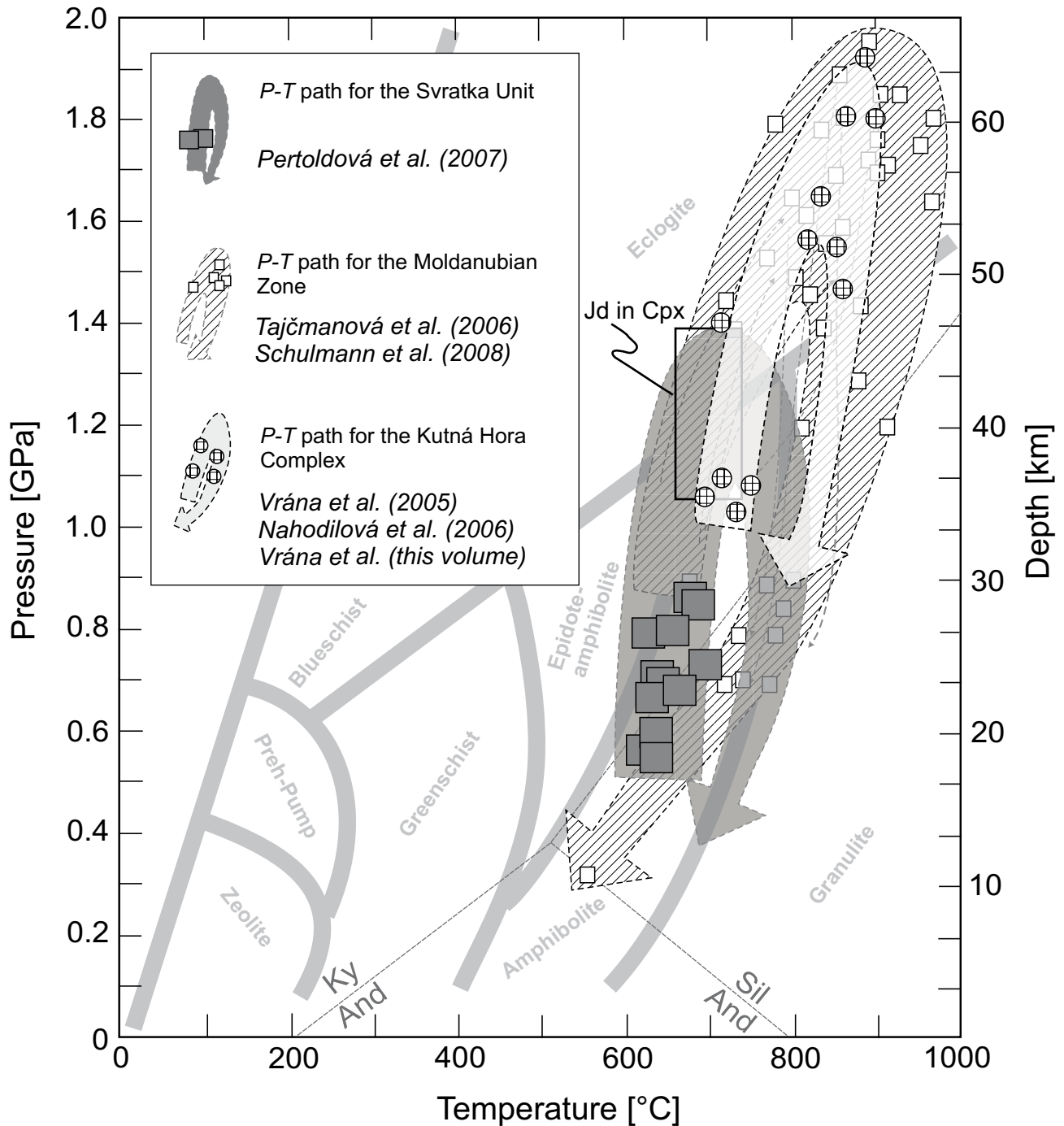


Fig. 13 Summary of the published P-T paths for the Svatka Unit, the Moldanubian Zone and the Kutná Hora Complex.

indeed any other rock-forming minerals, as the mobility of the most elements was strongly limited during Variscan metamorphism. Arguments pointing to the conservative character of skarn metamorphism (closed system) have already been published by Pertold et al. (1997). They include, for instance, different garnet and pyroxene chemistry, distinct whole-rock LREE/HREE ratios, and oxygen isotope compositions preserved in

individual skarn bands. Consequently, we consider the geochemical variability of individual skarn types (bands, layers) to be a close approximation to the original compositional/lithological variations of the pre-metamorphic protoliths.

Skarns from all three studied units share some common geochemical features. The $\text{Al}_2\text{O}_3/\text{TiO}_2$, $\text{Al}_2\text{O}_3/\text{Zr}$, TiO_2/Nb ratios are closely similar to the average composition of

post-Archaeon sediments but the elemental abundances are usually lower in the skarns. This suggests that skarn protoliths contained, apart from variable proportions of detrital sedimentary material, also other, non-detrital components. These admixtures were mainly Ca and Fe, supplied by chemogenic sedimentation. However, the balance of the latter elements is rather variable in our skarn set. The absence of correlation between FeO_t and CaO indicates that the two elements do not substitute for each other and that there was no replacement of Ca by Fe in the skarn protoliths.

The ΣREE values in the skarns (8 to 345 ppm) are both higher and lower than total REE contents in average sediments (121 ppm). If the abundances of Al_2O_3 , TiO_2 and Zr are considered as indicators for detrital sedimentary components in skarn protoliths, it is obvious from diagrams of Al_2O_3 vs. ΣREE and TiO_2 vs. ΣREE (Fig. 9) that the skarns contain mostly lower values of Al_2O_3 and TiO_2 compared to average sediments. However, their ratios to ΣREE point to relative enrichment of REE in the detritic component in the skarns. Comparison with paragneisses and migmatites of the Strážek Unit (Moldanubian Zone) (Hrdličková ed. 2008) and paragneisses of the Svatka Unit (Hanžl ed. 2008) shows that the latter have higher $\text{Al}_2\text{O}_3/\text{TiO}_2$ and $\text{Al}_2\text{O}_3/\text{Zr}$ ratios than Moldanubian paragneisses, skarns and average sediments. It is probable that formation of skarn protoliths included some processes resulting in relative enrichment in REE, which took place in a sedimentation environment with a lower supply of detrital material. This would account for the situation that many skarn samples contain ΣREE higher, but Al_2O_3 , TiO_2 and Zr lower, than the average sediments.

The lowest total REE contents (less than 100 ppm) occur in skarns that are known as former magnetite ore deposits or at least contain some magnetite accumulations (Zliv, Županovice, Rešice, Budeč, Vepřová, Kuklík, Pernštejn, Svatouch and Malešov). Skarns at Církvice and Jakub near Církvice also have such low ΣREE values, although no magnetite is known at present from these localities.

Europium anomalies in skarn samples have a wide range of 0.5–8.6. In many cases there are differences among samples from the same locality. The Fig. 10 indicates that the most pronounced positive anomalies of Eu occur predominantly in samples with the lowest ΣREE values. At the same time, these are the samples with the lowest contents of Al_2O_3 (less than 6 % wt. %), TiO_2 (less than 0.25 wt. %) and Zr (less than 100 ppm, with a few exceptions) and, in our interpretation, the samples with the lowest detrital component. The occurrence of a positive Eu anomaly in sediments and hydrothermal products is usually interpreted as resulting from the activity of reducing, high-temperature fluids. The presence of positive Eu anomalies in some skarn samples is in contrast with

the negative europium anomaly in metasediments of the Moldanubian Zone and of the Svatka Unit.

Hongo et al. (2007) recorded a distinct U-shaped REE patterns with a prominent positive Eu anomaly in hydrothermal fluids from the sea-floor black smoker (Fig. 8). These geochemical characteristics are in sharp contrast to the sea water, which does not have a Eu anomaly and the REE are not fractionated, except for Ce. According to Zheng et al. (2008), marine exhalites may exhibit enrichment in lanthanide elements by two or three orders of magnitude relative to the sea water, due to the hydrothermal solutions activity and adsorption phenomena. Oxidation processes take place in dependence on the distance from the source of solutions and residence time on the sea floor.

Some of the analysed skarns contain increased abundances of base metals: Cu (Budeč), Zn, Pb (Budeč, Zliv, Županovice, Býšovec, Kadov, Kuklík, Malešov), As (Budeč, Svatouch, Pernštejn, Kuklík), Bi and Au (Budeč, Zliv, Županovice, Církvice, Ruda near Čachnov, Svatouch, Kuklík). The tin contents are somewhat increased in all the skarn samples. The highest values were recorded in Budeč and Slatina (400 to 570 ppm), in Církvice, Jakub near Církvice, Líšné, Točice and Rešice (200 to 360 ppm), as well as in Vepřové, Pernštejn, Zliv, Čachnov, Teplá and Malešov (100 to 200 ppm). Skarns from other localities contain Sn in abundances that are one order of magnitude higher than in the metasediments of the Svatka Unit and the Moldanubian Zone (Hanžl ed. 2008; Hrdličková ed. 2008).

8.3. Oxygen isotopes

The $\delta^{18}\text{O}$ values for garnet and clinopyroxene are low, on average 1.62 ‰ for garnet, 2.2 ‰ for clinopyroxene and 2.15 ‰ for amphibole. Such compositions cannot be interpreted as resulting from crystallization of the mineral assemblages in the presence of granitoid-related magmatic fluids, i.e. during contact metamorphism. It is probable that a component of marine or meteoritic water was involved. The new data compare well with the results published by Pertoldová et al. (1998), Pertold et al. (1997), and Drahota et al. (2005) for other skarn localities in the Moldanubian Zone and the Svatka Unit. The determined $\delta^{18}\text{O}$ values for minerals from the Vrbík small contact metasomatic scheelite deposit (Pertold et al. 2000) and from the Vápenka contact skarn band (Drahota et al. 2005) exhibit distinctly higher heavy oxygen contents. The $\delta^{18}\text{O}$ values published by Meinert et al. 2003 for skarn deposits connected with igneous-related hydrothermal fluids are higher as well (Meinert et al. 2003: average $\delta^{18}\text{O}$ of 5.0 ‰ for garnet, 6.5 ‰ for clinopyroxene, and 7.1 ‰ for amphibole).

8.4. Geochronological data

The ages obtained from the detrital zircon population suggest that the youngest source of clastic material for the studied sample was of Cambrian–Neoproterozoic age. The most pronounced age maximum spans the interval of 500–640 Ma with maximum data density between 540–600 Ma. The Neoproterozoic protolith ages of 540–580 Ma are typical of granitoids and orthogneisses of the Brunovistulian (van Breemen et al. 1982; Scharbert and Batík 1985; Fritz et al. 1996; Friedl et al. 2000, 2004; see Leichmann and Höck 2008 for review) as well as Saxothuringian and Lugian domains (Hegner and Kröner 2000; Tichomirowa et al. 2001; Linnemann et al. 2008 and references therein) and they only rarely appear as protolith ages of Moldanubian orthogneisses (Schulmann et al. 2005). Importantly, Cambrian ages of 500–520 Ma are absent in the Moravo–Silesian Domain (Brunia), but they are known as the most frequent protolith ages of the Moldanubian orthogneisses, as well as of some orthogneisses in the Saxothuringian and Lugian domains (Vrána and Kröner 1995; Hegner and Kröner 2000; Tichomirowa et al. 2001; Friedl et al. 2004; Schulmann et al. 2005). Zircons of this age found in the studied samples can be interpreted either as clastic grains coming directly from eroded Cambrian granitoids or volcanics, or they represent recycled crystals from sediments rich in clastic material of Cambrian age. Ages older than 650 Ma are known from xenocrystic cores of younger zircons in Brunovistulian, Saxothuringian and Lugian orthogneisses as well as from orthogneisses and granulites of the Moldanubian Domain (Hegner and Kröner 2000; Kröner et al. 2000; Tichomirowa et al. 2001; Friedl et al. 2004 and references therein).

The presence of Cambrian zircons with an age of ~500 Ma shows that the deposition of the sedimentary protolith to the skarn sample occurred during Early Palaeozoic but not earlier than in the Mid–Late Cambrian times. Tectonometamorphic activity associated with the Variscan collision in the Bohemian Massif is known to be Mid–Late Devonian (Franke 2000) and this period can probably be regarded as the minimum age for the protolith sedimentation. The minimal age of metamorphism in the studied sample is constrained by the cooling ages from the associated gneisses, the Ar–Ar muscovite and amphibole age of which is Early Carboniferous (330–326 Ma; Fritz et al. 1996).

Interpretation of the sediment provenance is complicated by the fact that ages recognized in the detrital population are known from many geological units of the Bohemian Massif except for the Moldanubian Domain. The opening of a large oceanic domain (the Saxothuringian ocean) started during the Early–Middle Cambrian, resulting in separation of the Saxothuringian/

Lugian domains from the more easterly exposed units of the Bohemian Massif (Franke 2000). Given the major shortening of the Moldanubian Domain during the Variscan continental collision in the Late Devonian–Early Carboniferous, it seems unlikely that the Saxothuringian/Lugian domains acted as sediment source regions. They seem to have been remote from the Moldanubian Domain at the time of assumed sedimentation of the protolith, e.g. between Late Cambrian and Late Devonian. So far there is no reliable evidence for separation of the Brunovistulian and Moldanubian domains by a large ocean. The palaeogeographic reconstruction, together with the dense population of zircons with ages between 540 and 580 Ma, support the interpretation that much of the clastic material could have been derived from sources in the neighbouring Brunovistulian Domain. Archaean–Mesoproterozoic ages probably correspond to recycled zircons, either xenocrystic cores of magmatic zircons in Neoproterozoic granitoids, or clastic zircons in the sediments of Neoproterozoic–Cambrian age.

9. Conclusions

- Skarns form elongated lenticular bodies that are sub-parallel to regional fabrics in the surrounding rocks. The skarns have preserved the relict fabrics, which are in some cases overprinted by regional structures.
- Variations in mineral chemistry correspond to differences in the chemical compositions of the protoliths.
- The temperature or P-T conditions of the skarn mineral assemblages were determined in all three units. These conditions are compared to P-T intervals of metamorphic paths recognized in surrounding rocks.
- Three metamorphic stages (M_1 – early prograde, M_2 – prograde/peak, M_3 – retrograde) were determined in the skarns from the Svatka Unit, with minimal peak pressures of 1.4 GPa.
- The P-T conditions in skarns from the Moldanubian Zone are represented by a narrow interval within the amphibolite facies. The peak pressure conditions of metamorphism were not preserved and compositional zoning in garnets shows signs of retrograde Variscan evolution.
- Chemical zoning of some garnets from the Kutná Hora Complex indicates prograde development under amphibolite-facies conditions but peak pressure conditions were not preserved, similarly to the Moldanubian Zone. The chemically homogeneous compositions of individual generations of clinopyroxenes and epidotes do not reflect the prograde and retrograde events in the Kutná Hora Complex.
- In all three units, a local increase in the oxygen fugacity in the system played a significant role in

development of minerals rich in Fe³⁺, especially magnetite.

- The estimated P-T conditions for the skarns are in compliance with the metamorphic paths of the surrounding metasediments in all the studied units, except the peak pressures in the Svratka Unit. These results indicate that the protolith of skarns endured all the metamorphic events together with the surrounding sedimentary/volcano-sedimentary sequences.
- The light oxygen isotopic compositions in skarn silicates from the Svratka Unit do not support the idea of an open-system genesis due to a granitoid-magma derived fluid activity. The interaction with seawater was probable during the formation of the skarn protoliths.
- The large variability of the zircon ages (0.5–2.6 Ga) suggests their detrital origin. The most frequent ages fall in the interval of 500–640 Ma with majority of data clustering between 540–600 Ma. Zircons and other clastic material are therefore Early Palaeozoic (>500 Ma) in age and possibly linked to a source in rocks of Cambrian–Neoproterozoic age.
- Skarn protoliths were most probably mixed sedimentary rocks with a component of exhalites deposited on the sea floor. In addition to enrichment in Ca and Fe, and lowered Si, they contain increased abundances of some metals (Zn, Pb, As, Sn). The total REE contents with the type and magnitude of the Eu anomaly are highly variable. They indicate variation in the local temperature and redox conditions among individual layers at a single locality as well as regional changes between individual localities. There is also an indication of rather conservative character of the metamorphism with limited migration of most geochemical species within the garnet-clinopyroxene rocks. It is probable that hydrothermal solutions, instrumental for the development of skarn protoliths, debouched on the sea floor along fracture zones with the character of local extensional rifts. The deposition of exhalites proceeded along such zones, where they were mixed with detrital material in variable proportions.
- The geochemical features of the studied skarns do not indicate systematic differences among the three studied geological units, although the tectono-metamorphic development was indeed different. The sedimentary-exhalative protoliths of the skarns were mineralized by magnetite at some localities, giving rise to small deposits of iron ore, accompanied by a slight enrichment in some base metals, as is common in this type of mineralization.
- The geological positions of the skarn bodies and their structural and metamorphic record reflect probable collateral tectono-metamorphic development of skarns with the surroundings rocks. The geochemical signatures, the oxygen isotopic compositions and the

presence of detrital zircons with a wide range of ages exclude metasomatic, and instead point to a sedimentary-exhalative, mode of origin for the studied skarns.

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