# EUROPEAN ORGANIZATION FOR NUCLEAR RESEARCH

Proposal to the ISOLDE and Neutron Time-of-Flight Committee

# Emission channeling with short-lived isotopes (EC-SLI) of acceptor dopants in nitride semiconductors

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#### Abstract

Despite the overwhelming technological success of GaN-related devices in optoelectronics, the detailed structure of possible acceptor impurities in nitride semiconductors has been heavily debated ever since the realization of *p*-type GaN through Mg doping by Akasaki, Amano, and Nakamura in 1989-1992.

The implementation of the Emission Channeling with Short-Lived Isotopes (EC-SLI) lattice location method with  ${}^{27}Mg(t_{1/2}=9.45 \text{ min})$  and  ${}^{11}Be(t_{1/2}=13.8 \text{ s})$  at ISOLDE has now for the first time provided an experimental technique for direct study of the structural properties of Mg and Be in nitrides. While this has already lead to valuable results, including direct evidence for the existence of interstitial Mg and Be, the fact that interstitial and substitutional fractions of these implanted probes can now be detected with the precision of a few %, opens further possibilities of investigating how electrical activation processes and radiation damage influence these quantities in differently doped GaN samples.

As further studies using <sup>27</sup>Mg we propose to concentrate on detailed mapping of the site change from interstitial to substitutional Ga sites in different doping types of GaN (nid-GaN, GaN:Mg, *p*-GaN:Mg, *n*-GaN:Si,) as a function of implantation temperature and fluence.

In the case of <sup>11</sup>Be, where so far we could only achieve room temperature results in GaN, we aim at establishing the temperatures where the site changes from interstitial to substitutional positions occur in nid-GaN, *p*-GaN:Mg and AlN.

# Requested shifts: 24 (split into 3 runs over 2 years)

# **Motivation**

Mg-doped p-type GaN is nowadays a core component of many optoelectronic devices which we find in our homes, e.g. light emitting diodes (LEDs) for solid state white lighting or blue lasers [1-3]. Applications of GaN for power electronics, e.g. as high voltage transistors or power converters, also exist already but are less widespread; others, as in photovoltaics may still emerge. Despite the technological maturity of GaN, there are some basic properties related to its *p*-type doping that are still poorly understood and also limit the performance of devices. One such major problem is an inherent doping limit: once the Mg concentration in GaN surpasses  $\sim 10^{19} - 10^{20}$  cm<sup>-3</sup>, further introduction of Mg does not lead to an increase in the hole concentration, see e.g. Ref. [4] and Refs. therein. The electrical activation of Mg as a p-type dopant requires its incorporation on the substitutional Ga site  $S_{Ga}$  (Mg<sub>Ga</sub>) while Mg on other sites, e.g. interstitial Mg (Mg<sub>i</sub>), or Mg replacing N (Mg<sub>N</sub>), should be electrically inactive or even exhibit donor character. In the earlier stages of GaN research it was suggested from ab initio density functional theory calculations by Neugebauer and Van de Walle [5-6] that "incorporation of the Mg atoms on the N site or in an interstitial configuration were found to be negligible." This view, however, was not shared by all theorists, e.g. Reboredo and Pantelides stated that "interstitial Mg plays a major role in limiting *p*-type doping" [7]. The failure of experimental methods in establishing the existence of interstitial Mg and investigating its properties probably contributed to the fact that in the following years many researchers adopted the opinion that the doping limitations at high Mg concentrations are, in addition to passivation by H [1-3,6,8], mostly a consequence of native defects with donor character, in particular N vacancies  $V_{\rm N}$ , either in their isolated form [9-10] or in Mg<sub>Ga</sub>- $V_{\rm N}$  complexes with Mg [11-12].

Recently this controversial discussion was revived by the results of hybrid density functional calculations of Miceli and Pasquarello [13] who concluded that "...the amphoteric nature of the Mg impurity is critical to explain the dropoff in the hole density observed experimentally": once the doping limit has been reached, additional Mg atoms are not incorporated on substitutional Ga sites any more but on interstitial sites where they form compensating double donors, thus pinning the Fermi level, cf Ref. [13] for a more detailed discussion of several arguments that support this theory from the viewpoint of electrical and optical characterization methods. Low formation energies for interstitial Mg<sub>i</sub> if the Fermi level is located below mid gap were also theoretically predicted by Reshchikov *et al.* [14].

For the lighter alkaline earth metal Be, theory has not reached a consensus whether in GaN its acceptor energy is similar to Mg [6,10,15-16] or much higher [17-18]. It has also been suggested that Be suffers to larger extent than Mg from self-compensation in the form of Be<sub>i</sub> interstitials [6,15-16]. In the case of the extremely wide band gap material AlN, the quest for *p*-type AlN has not been successful and besides a deep nature of Mg and Be acceptors [18] compensation by interstitials may also play a leading role.

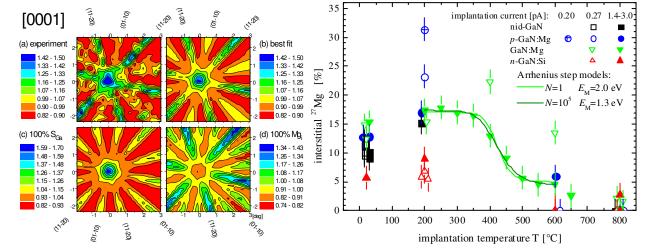
In contrast to the various refined but often contradictory theoretical models on the structural properties of Mg- and Be-related defects in GaN, no experimental data on the Mg and Be lattice sites existed so far. The often used ion beam lattice location technique of Rutherford backscattering spectrometry/channeling (RBS/C) is not applicable in this case since Mg and Be are much lighter than the Ga host atoms. Alternative ion beam detection methods such as particle-induced X-ray emission (PIXE) or nuclear reaction analysis fail at the low concentrations of dopants. Attempts to use extended X-ray absorption fine structure (EXAFS) for characterizing the distances from Mg atoms to their nearest neighbors were not successful since the Mg K X-ray absorption edge overlaps with the Ga L-edge [19].

# Summary of previous results

Within our previous experiment IS453 we have recently provided first direct lattice location measurements for <sup>27</sup>Mg ( $t_{1/2}$ =9.5 min) in AlN [20-21] and GaN [20,22] using the  $\beta^-$  emission

channeling technique with short-lived isotopes (EC-SLI). We found that the majority of ionimplanted Mg occupies substitutional Ga or Al sites. However, we presented also direct experimental proof for the existence of interstitial Mg<sub>i</sub> on positions near the so-called octahedral sites. In the case of GaN we could show that the interstitial Mg fraction depends on the GaN doping character, being enhanced in samples that were pre-doped *p*-type with stable Mg during growth, and suppressed in Si-doped *n*-GaN, which was recently accepted in Phys. Rev. Lett. [22]

The EC-SLI method allows probing the sites of radioactive isotopes in single-crystalline samples [23-25]. The radioactive probe atoms are implanted at low fluence and the emitted  $\beta^{-}$  particles are guided by the crystal potential. Within our on-line setup [26] at the GHM beam line, a twodimensional position-sensitive detector (PSD) [24-25] is used to measure the angle-dependent emission yield of electrons in the vicinity of major crystallographic directions, providing patterns which are characteristic for the probe atom sites in the sample. In order to derive the fractions of probe atoms on different lattice sites, the experimental  $\beta^-$  emission yields are fitted by a linear combination of theoretical patterns corresponding to emitter atoms residing on different lattice sites. The theoretical  $\beta^{-}$  patterns are calculated using the so-called "manybeam" formalism [23-24]. In case of Mg and Be, the only radioactive isotopes suitable for EC-SLI studies are <sup>27</sup>Mg (9.45 min) and <sup>11</sup>Be (13.8 s). The GaN layers used were specially grown by metal-organic chemical vapour deposition (MOCVD) at the University of Cambridge and based on 4.8 µm thick wurtzite single-crystal GaN epilayers on (0001) sapphire. The first type of sample simply consisted of the not intentionally doped epilayer ("nid-GaN"), while for "*n*-GaN:Si" samples a 150 nm thick layer doped with  $1 \times 10^{19}$  cm<sup>-3</sup> Si was deposited in addition. Mg predoped samples contained a 150 nm thick top layer doped during growth with 2×10<sup>19</sup> cm<sup>-3</sup> Mg. However, while "GaN:Mg" samples were used as-grown, "p-GaN:Mg" samples were annealed for 20 min at 800°C under nitrogen atmosphere in order to drive out H and electrically activate the Mg [27].



**Fig. 1.** (a) Experimental <sup>27</sup>Mg  $\beta^-$  emission pattern around the [0001] axis of *p*-GaN:Mg during 200°C implantation at a beam current of 0.20 pA, in comparison to (b) the best fit of simulated patterns for 72% aligned with the *c*axis, e.g. on S<sub>Ga</sub>, and 31% on the Mg<sub>i</sub> interstitial sites near the octahedral position. The theoretical patterns for S<sub>Ga</sub> and Mg<sub>i</sub> sites are shown in panels (c) and (d). From [22].

**Fig. 2.** Interstitial fractions of <sup>27</sup>Mg in the four different types of GaN samples as a function of the implantation temperature. Open symbols were implantations at low beam currents of 0.20-0.27 pA, while filled symbols were measured at 1.4-3.0 pA. The solid lines show the fractions using two Arrhenius models: N is the assumed number of jumps required before encountering a Ga vacancy and  $E_{\rm M}$  is the activation energy for migration of interstitial Mg. From [22].

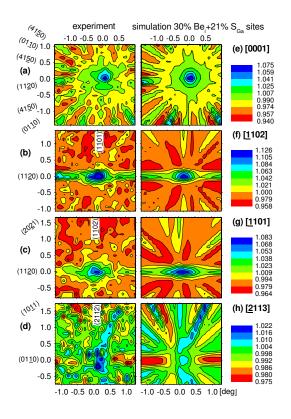
Fig. 1(a) shows the  $\beta^-$  emission distribution around the [0001] axis from <sup>27</sup>Mg implanted at 200°C into *p*-GaN:Mg at the very low beam current of 0.20 pA into a 1 mm beam spot, total fluence  $1.1 \times 10^{11}$  cm<sup>-2</sup>. The fact that the [0001] direction and all major planes exhibit channeling effects proofs that the majority of <sup>27</sup>Mg probes are aligned with the *c*-axis. However, comparing the experimental results to the theoretical pattern expected for 100% of <sup>27</sup>Mg aligned with the *c*-axis [Fig. 1(c)], one recognizes that the set of [01–10] planes has a lower intensity in comparison to [11–20] planes. This is a consequence of a minority of probes occupying positions in the wide open interstitial region of the wurtzite lattice, such as O, HA, HB or HAB sites since for those sites the (01–10) planes show blocking effects while the (11–20) planes keep the same anisotropy as for sites aligned with the *c*-axis [Fig. 1(d)]. Indeed, the best fit to this experimental pattern was obtained for 31% of Mg on interstitial sites and 72% aligned with the *c*-axis [Fig. 2(b)].

In Fig. 2 the fraction of  $^{27}$ Mg on interstitial Mg<sub>i</sub> sites is plotted as a function of the implantation temperature for the four different doping types of GaN, investigated at low and high implantation beam currents. For collecting these data each sample was once oriented with the [0001] facing the detector and then not moved again, only varying the implantation temperature and current, thus assuring maximum reproducibility of emission patterns and good control of the implantation fluence. Samples were first measured using beam currents of 0.20-0.27 pA at 20°C or 200°C, followed by implantations and measurements at higher temperatures, then the beam current was increased to 1.4-3.0 pA and the procedure repeated. Several characteristics are obvious. For measurements within the same beam current series the interstitial <sup>27</sup>Mg fraction is clearly correlated with the doping type: interstitial <sup>27</sup>Mg<sub>i</sub> is more prominent in *p*-GaN:Mg than in nid-GaN and lowest in n-GaN:Si. The second very clear characteristic is that for high implantation temperatures the interstitial fraction of <sup>27</sup>Mg is reduced, reaching in all cases values around 0% at 800°C. Third, for implantations at 200°C into p-GaN:Mg, there is a clear and pronounced influence of the fluence visible: during the very first measurement (0.20 pA, Fig. 2) Mg<sub>i</sub> reached 31%, dropping to 23% in a subsequent measurement at 0.27 pA; having finished the annealing sequence to 800°C (a temperature at which no more interstitial Mg was found), reimplanting at 200°C with 1.4 pA resulted in 17% Mg<sub>i</sub> only. Remarkably, in the case of *n*-GaN:Si, this trend is reversed: for the second sequence of measurements at higher implantation current the initially small fraction of Mgi increased somewhat. Overall, we attribute these fluence-related effects as caused by the introduction of implantation damage, which, as it accumulates, shifts the Fermi level of a sample towards mid gap and also introduces more and more Ga vacancies. While the latter favours the formation of  $Mg_{Ga}$  in all samples, the Fermi level shift is of minor consequences in nid-GaN but has opposite effects in p- and n-type samples (decrease of Mg, in pbut increase in *n*-GaN), which are overlaid on the influence of temperature and doping.

The temperature dependence of the interstitial fraction  $f_i(T)$  for high-current implantation into the GaN:Mg sample was used to estimate the migration energy ( $E_M$ ) of interstitial Mg<sub>i</sub> applying two simple Arrhenius models. The models assume that Mg<sub>i</sub> that remained interstitial following implantation starts to migrate due to its thermal energy and that it requires a certain number of jumps N until it encounters a Ga vacancy during the life time  $\tau$ =13.6 min of <sup>27</sup>Mg, which will lead to the formation of Mg<sub>Ga</sub> (see Ref. [20-21] for details):

$$f_{i}(T) = f_{i0} \exp\left[-\frac{v_{0}\tau}{N} \exp\left(-\frac{E_{M}}{k_{B}T}\right)\right].$$

Besides an attempt frequency  $v_0$ , for which we have taken  $2 \times 10^{13}$  Hz, N was assumed either 1 or  $10^5$ , where N=1 represents the limiting case in which the Mg<sub>i</sub> has a neighbouring Ga vacancy, and  $N=10^5$  is the upper limit when the diffusion-induced widening of the Mg<sub>i</sub> profile becomes comparable to the implantation depth, which can be excluded since it would considerably deteriorate the channeling effects. According to the least square fits for  $E_M$  shown by the solid lines in Fig. 2, the migration energies are thus estimated between 2.0 eV (N=1) and 1.3 eV ( $N=10^5$ ), respectively.



**Fig. 3.** (a-d) Experimental <sup>11</sup>Be  $\beta^-$  emission patterns from nid-GaN during RT implantation, in comparison to (e-h) the best fit of simulated patterns for 21% on S<sub>Ga</sub>, and 30% on the Be<sub>i</sub> interstitial sites near the octahedral position. Fractions are not yet corrected for background, typically this will increase them by a factor of ~2.

Our first test measurements with <sup>11</sup>Be implanted at RT in nid-GaN, performed during a parasitic run in 2012, gave encouraging results (Fig. 3). <sup>11</sup>Be is a challenging isotope for the EC-SLI method, since due to its high  $\beta^-$  endpoint energy of 11.5 MeV, extensive manybeam simulations at many different energies are required. These simulations have been performed in the meantime and it will be possible to directly analyse future experiments based on them. According to the best fit results in Fig. 3 the majority of Be was actually found on interstitial sites.

#### Proposed future experiments and summary of requested shifts:

The implementation of the EC-SLI method with <sup>27</sup>Mg and <sup>11</sup>Be at ISOLDE has for the first time provided an experimental method for direct study of the structural properties of Mg and Be in nitride semiconductors. While this has already lead to valuable results [20-22], the fact that interstitial and substitutional fractions of these implanted probes can now be detected with the precision of a few %, opens further possibilities to investigate how electrical activation processes and radiation damage influence these quantities in differently doped GaN samples. Within a wider picture, our studies also aim at comparing the properties of Mg and Be in nitrides with those of the chemically similar probes <sup>8</sup>Li [28], <sup>24</sup>Na [21,29-30], <sup>42</sup>K, <sup>45</sup>Ca [30-31], and <sup>89</sup>Sr [30-31], for which emission channeling results are already available. This will allow a systematic correlation of occupied lattice sites and interstitial migration energies with e.g. ionic radii, electronegativity, and other chemical properties.

In the case of <sup>27</sup>Mg we would like to concentrate on detailed mapping of the site change from interstitial to substitutional Ga sites in differently doped GaN as a function of implantation temperature and fluence. Particular focus shut be put on further characterizing as-grown GaN:Mg which has not yet been electrically activated by annealing. Here, our first results at low fluence showed a considerable increase in the interstitial Mg fraction upon implantation at 400°C. If this can be confirmed and systematically measured in further experiments, it would allow to study in more detail the processes that take place when Mg-doped GaN:Mg is thermally activated and thus converted to *p*-GaN:Mg. This is commonly associated with the release of H from the sample, hence we will also explore first measurements with intentionally hydrogenated GaN samples. Moreover, precise Arrhenius curves, as are shown for high-fluence implantation in GaN:Mg in Fig. 2, should be measured for other doping types and especially at low implantation currents. This should establish whether the precise site change temperature depends on the doping type of the sample, which would indicate varying activation energies for Mg migration, or differing concentrations of Ga vacancies depending on doping type.

Our shift request assumes two beam times for <sup>27</sup>Mg and one for <sup>11</sup>Be during the 2-year period 2017+2018 before the ISOLDE shutdown. The total amount of data contained in Fig. 2 was measured during a 4 shift beam time, so a total of 2×8 shifts should be sufficient to perform the

detailed studies outlined above. We would like to point out that dedicated Mg RILIS beam times with a Ti target (the only target that is suitable for our purpose since it allows reducing stable <sup>27</sup>Al and radioactive <sup>27</sup>Na contaminations to acceptable levels) with less than 2 days duration are usually too short to be accommodated in the ISOLDE schedule.

In 2014, within the last addendum of IS453, INTC had approved 3 shifts for <sup>11</sup>Be experiments. However, such a minor amount of shifts could not be scheduled in the years 2014-2016. Therefore, the only data we have measured so far are shown in Fig. 3 and concern RT implantation into GaN. These were achieved during a parasitic run in 2012 during the times when the primary user did not take beam. For <sup>11</sup>Be we aim at establishing at which temperature the site changes to substitutional sites occur in undoped nid-GaN, *p*-GaN:Mg and in AlN, which should take ~8 shifts, since <sup>11</sup>Be yields are lower than <sup>27</sup>Mg. This will allow assessing the relevance of interstitial Be<sub>i</sub> as compensating centers, and first experimental values for the interstitial migration energy of Be<sub>i</sub> in these materials, for which theoretical predictions in GaN suggest  $E_{\rm M}({\rm Be}_i)=1.2-2.9$  eV [15].

isotope	yield (atoms/µC)	target – ion source	Shifts (8h)
<sup>27</sup> Mg	1×10 <sup>7</sup>	Ti-W – RILIS Mg	16
<sup>11</sup> Be	6×10 <sup>6</sup>	UC <sub>x</sub> -W or Ta-W – RILIS Be	8

Table1: Number of shifts requested for each isotope

Total shifts: 24

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# Appendix

## **DESCRIPTION OF THE PROPOSED EXPERIMENT**

The experimental setup comprises:

*EC-SLI* on-line implantation and measurement chamber at ISOLDE GPS-GHM beam line

Part of the Choose an item.	Availability	Design and manufacturing
EC-SLI, on-line emission channeling chamber at GHM	Existing	ig  To be used without any modification
[insert lines if needed]		

#### HAZARDS GENERATED BY THE EXPERIMENT

*(if using fixed installation)* Hazards named in the document relevant for the fixed installation. [EC-SLI, On-line Emission Channeling Chamber at GHM]

Additional hazards:

Hazards	EC-SLI Chamber at GHM	
Thermodynamic and f	luidic	
Pressure	[-][Bar], [-][l]	
Vacuum	10 <sup>-6</sup> mbar, 70 l	
Temperature	[Room temperature] [K]	
Heat transfer		
Thermal properties of	At the sample holder inside	
materials	the chamber, under vacuum,	
	there is an annealing system	
	up to 1173 K. A closed cycle	
	refrigerator can cool samples	
	down to 50 K	
Cryogenic fluid	[He gas], [13 ] <b>[Bar]</b> , [3] <b>[l]</b>	
	Closed cycle He refrigerator	
Electrical and electron	nagnetic	
Electricity	[220] <b>[V],</b> [25] <b>[A]</b>	
Static electricity		
Magnetic field	[magnetic field] [T]	
Batteries		
Capacitors		
Ionizing radiation		
Target material	Solid-state ,single crystal	
	samples for analysis, mounted	
	into a goniometer under	
	vacuum. The short-lived	
	isotopes are directly implanted	
	into the sample and	
	measurements are done in	
	situ.	
Beam particle type (e, p,	Isolde radioactive ion beams	
ions, etc)		
Beam intensity	From pA to nA	
Beam energy	From 30 to 60 keV	
Cooling liquids	[liquid]	
Gases	[gas]	 
Calibration sources:	Calibrations are done on-	
	line with real samples during	

<ul> <li>the runs, or using weak sources implanted with long lived isotopes, e.g. 73 As (80d - c.e. emitter) or 595e (4.5.d - β-emitter)</li> <li>Open source</li> <li>27Mg (9.5 min) 1.2MBg (stur.)</li> <li>12Be (13.8 s) 0.8MBg (stur.)</li> <li>12Be (13.8 s) 0.0MBg (stur.)</li> <li>Sealed source</li> <li>Seated source</li> <li>IBe (13.8 s) 0.0MBg (stur.)</li> <li>Seated source</li> <li>Seated source</li> <li>Max 500 µSV/h at contact at the chamber wals, during a distance</li> <li>addition at the working zones</li> <li>Sotope</li> <li>Sotope and yield. Typical is "300 µSV/h at contact at the chamber wals, "500 µSV/h in contact. The Chamber wals, "500 µSV/h in contact.</li> <li>Isotope</li>     &lt;</ul>		1	1
Ived isotopes, e.g. 73 A st (8d)           - 0pen source         Demitter)           23         27Mg (9.5 min) 1.2MBg (satur.)           - 11Be (13.8.5) 0.8MBg (satur.)		the runs, or using weak	
-c.e. emitter) or 59Fe (44.5d - B-emitter)     - Eventiter)		sources implanted with long	
-c.e. emitter) or 59Fe (44.5d - B-emitter)     - Eventiter)		lived isotopes, e, g, 73 As (80d	
Beamter)         Common source           Open source         Z7Mg (9 S min) 1.2MBg (stur.)           11Be (13.8 s) 0.2MBg (stur.)           • Sealed source           • Isotope           • Isotope           • Activity           Use of activated material:           • Description           • Description           • Description           • Description           • Description           • Description           • Doscription           • Doscription           • Doscription           • Doscription           • Doscription           • Doscription           • Statuments, depending on lostore and yield. Typical Is ~100 LSv/h on contact. at the chamber walls, Soly Si/h in 10 cm distance.           • The chamber is enclosed by concrete blocks that shield radiation at the working zones to ambient background of the ISDLDE hall, < Si LSv/h.           Experiments are remotely monitored.           • Isotope           • Activity           Non-Lonizing radiation           Hardifferoupency (1- 300MHz)           OPMI-Ionizing radiation           Chemical agent, [quantity]           Hardifferoupency (1- 300MHz)           OChemical agent, [quantity]           Chemical agent, [quantity]      <			
Open source     Zimage (9.5 min) 1.2MBq (sour.)      Sealed source     [ISO standard]     Isotope     Activity     Second matrix     destinated material:     Description     Description     Description     Description     Conserved material:     Description     Description     Description     Description     Conserved material:     Description			
27Mg (9.5 min) 1.2MBg (sotur.)       11Be (13.8 s) 0.8MBg (sotur.)       • Sealed source       • Isotope       • Activity       Use of activated material:       • Description       • Distributed material:       • Description       • State       • Distributed material:       • Description       • Distributed material:       • Description       • State       • Distributed material:       • State       • State       • Chardial and the working zones to ambient background of the isotope       • Nort-ionizing radiation       • Activity       • Activity       • Activity	Open source		
(satur.)       118e (13.8 s) 0.8MBg (satur.)         • Sealed source       [ISO standard]         • Seated source       [ISO standard]         • Sestope	• Open source		
11Be (13.8 s) 0.8MBg (stur.)         • Sealed source         Isotope         • Activity         Use of activated material:         • Description         • Activity         • Description         • Activity         • Activity         • Activity         • Activity         • Activity         • Isotope         • Activity         • Isotope         • Isoto			
<ul> <li>Sealed source</li> <li>Iso standard)</li> <li>Iso standard)</li> <li>Isotope</li> <li>Activity</li> <li>Use of activated material:</li> <li>Description</li> <li>Description</li> <li>Max. 500 µSv/h at contact at the chamber walls, during experiments, depending on isotope and yield. Typical is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h. Experiments are remotely monitored.</li> <li>Isotope</li> <li>Isotope</li> <li>Activity</li> <li>Non-ionizing radiation</li> <li>Laser</li> <li>Uv light</li> <li>Microwaves (300MHr-30</li> <li>GHz)</li> <li>GHzi (Learnical agent), (quantity)</li> <li>Chemical agent, (quantity)</li></ul>		(satur.)	
<ul> <li>Sealed source</li> <li>Iso standard)</li> <li>Iso standard)</li> <li>Isotope</li> <li>Activity</li> <li>Use of activated material:</li> <li>Description</li> <li>Description</li> <li>Max. 500 µSv/h at contact at the chamber walls, during experiments, depending on isotope and yield. Typical is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h on contact. at the chamber walls, of yield is "100 µSv/h. Experiments are remotely monitored.</li> <li>Isotope</li> <li>Isotope</li> <li>Activity</li> <li>Non-ionizing radiation</li> <li>Laser</li> <li>Uv light</li> <li>Microwaves (300MHr-30</li> <li>GHz)</li> <li>GHzi (Learnical agent), (quantity)</li> <li>Chemical agent, (quantity)</li></ul>			
<ul> <li>Isotope <ul> <li>Activity</li> <li>Use of activated material:</li> <li>Description</li> <li>Description</li> <li>Max. 500 µSv/h at contact at and in 10 cm distance</li> <li>isotope and yield. Typical is "100 µSv/h on contact. at the chamber wills, during experiments, depending on isotope and yield. Typical is "100 µSv/h on contact. at the chamber wills." 500 µSv/h in 10 cm distance.</li> <li>The chamber is enclosed by concrete blocks that shield radiation at the working zones to ambient background of the ISODE hall, &lt;50 µSv/h.</li> <li>Experiments, depending on isotope and yield. Typical is "100 µSv/h on contact. at the chamber wills, during experiments, depending on isotope and yield. Typical is "100 µSv/h on contact. at the chamber wills, during experiments, depending on isotope and yield. Typical is "100 µSv/h on contact.</li> <li>Isotope</li> <li>Activity</li> <li>Non-ionizing radiation</li> <li>Laser</li> <li>UV light</li> <li>Microwaves (300MHz-30</li> <li>GHz)</li> <li>Radiofrequency (1-300 µSv/h)</li> <li>Gorosive (chemical agent], (quantity)</li> <li>Chemical agent, (quantit</li></ul></li></ul>		11Be (13.8 s) 0.8MBq (satur.)	
<ul> <li>Isotope <ul> <li>Activity</li> <li>Use of activated material:</li> <li>Description</li> <li>Description</li> <li>Max. 500 µSv/h at contact at and in 10 cm distance</li> <li>isotope and yield. Typical is "100 µSv/h on contact. at the chamber wills, during experiments, depending on isotope and yield. Typical is "100 µSv/h on contact. at the chamber wills." 500 µSv/h in 10 cm distance.</li> <li>The chamber is enclosed by concrete blocks that shield radiation at the working zones to ambient background of the ISODE hall, &lt;50 µSv/h.</li> <li>Experiments, depending on isotope and yield. Typical is "100 µSv/h on contact. at the chamber wills, during experiments, depending on isotope and yield. Typical is "100 µSv/h on contact. at the chamber wills, during experiments, depending on isotope and yield. Typical is "100 µSv/h on contact.</li> <li>Isotope</li> <li>Activity</li> <li>Non-ionizing radiation</li> <li>Laser</li> <li>UV light</li> <li>Microwaves (300MHz-30</li> <li>GHz)</li> <li>Radiofrequency (1-300 µSv/h)</li> <li>Gorosive (chemical agent], (quantity)</li> <li>Chemical agent, (quantit</li></ul></li></ul>			
<ul> <li>Activity</li> <li>Use of activated material:</li> <li>Description</li> <li>Description</li> <li>Max. 500 µSv/h at contact at the chamber walls, during experiments, depending on isotope and yield. Typical is ~100 µSv/h on contact. at the chamber walls, ~50 µSv/h in 10 cm distance.</li> <li>The chamber walls, ~50 µSv/h in 10 cm distance.</li> <li>The chamber is enclosed by concrete blocks that shield radiation at the working zones to ambient background of the ISOLDE hall, &lt; 5 µSv/h. Experiments are remotely monitored.</li> <li>Isotope</li> <li>Activity</li> <li>Non-ionizing radiation</li> <li>Laser</li> <li>UV light</li> <li>Microwaves (300MHz-30 GH)</li> <li>GH2</li> <li>Radiofrequency (1-300MHz-30 GH)</li> <li>GH2</li> <li>Chemical</li> <li>Toxic</li> <li>[chemical agent], [quantity]</li> <li>CMR (carcingens, [chemical agent], [quantity]</li> <li>CMR (carcingens, [chemical agent], [quantity]</li> <li>Carcisve</li> <li>[chemical agent], [quantity]</li> <li>Finamable</li> <li>[chemical agent], [quantity]</li> <li>Finamable</li> <li>[chemical agent], [quantity]</li> <li>Finamable</li> <li>[chemical agent], [quantity]</li> <li>Finamable</li> <li>[chemical agent], [quantity]</li> <li>Pinamable</li> <li>[chemical agent], [quantity]</li> <li>Pination</li> <li>[location]</li> <li>[l</li></ul>	<ul> <li>Sealed source</li> </ul>	[ISO standard]	
Use of activated material:	<ul> <li>Isotope</li> </ul>		
Use of activated material:	Activity		
Description     Dose rate on contact     and in 10 cm     distance     Max. 500 µSv/h at contact at     and in 10 cm     distance     Max. 500 µSv/h at contact at     the chamber walls, during     experiments, depending on     isotope and yield. Typical is     "100 µSv/h on contact. at the     chamber walls, "50 µSv/h in     10 cm distance.     The chamber is enclosed by     concrete blocks that shield     radiation at the working zones     to ambient background of the     ISOLDE hall, < 5 µSv/h.     Experiments are remotely     monitored.     Isotope     Activity     Non-ionizing radiation     Laser     Vilight     UV light     Chemical agent, [quantity]     Chemical agent, [quantity]     Chemical agent, [quantity]     Irritant     (chemical agent, [quantity]     Planmable     (chemical agent, [quantity]     Planmable     (chemical agent, [quantity]     Irritant     (chemical agent, [quantity]     Planmable     (chemical agent, [			
Dose rate on contact and in 10 cm distance the chamber is enclosed by concrete blocks that shield radiation at the working zones to ambient background of the ISOUE halt, < S isych, Experiments are remotely monitored. lostope distive uv light Mon-lonizing radiation laser Uv light Microwaves (300MHz-30 GHz) Ghaptical distance toxic [chemical agent], [quantity] Chemical Corrosive infrant ichemical agent], [quantity] Corrosive infrant ichemical agent], [quantity] infrant ichemical agent], [quantity]			
and in 10 cm distancethe chamber walls, during experiments, depending on isotope and yield. Typical is ~100 µSy/h on contact. at the homber valls, 750 µSy/h in 10 cm distance. The chamber valls, 750 µSy/h in 10 cm distance. The chamber valls during zones radiation at the working zones monitored.• Isotope			
distance       experiments, depending on isotope and yield. Typical is "100 µSV/h on contact. at the chamber walls, "50 µSV/h in 10 cm distance. The chamber is enclosed by concrete blocks theids agents to ambient background of the ISOLDE hall, <5 µSV/h. Experiments are remotely monitored.         • Isotope		-	
isotope and yield. Typical is         ``100 µSv/h on contact. at the chamber walls, ~50 µSv/h in 10 cm distance.         The chamber is enclosed by concrete blocks that shield radiation at the working zones to ambient background of the ISOLE hall, <5 µSv/h.         Experiments are remotely monitored.         • Isotope         • Activity         Non-ionizing radiation         Laser         UV light         Radiofrequency (1- 300MHz-30 GHz)         GHz)         Chemical         Toxic         (chemical agent), [quantity]         Harmful         (chemical agent), [quantity]         CMR (carcinogens, [chemical agent], [quantity]         Corrosive       [chemical agent], [quantity]         Irritant       [chemical agent], [quantity]         Corrosive       [chemical agent], [quantity]         Oxidizing       [chemical agent], [quantity]         Oxidizing       [chemical agent], [quantity]         Oxidizing       [chemical agent], [quantity]         Dangerous for the [chemical agent], [quantity]       [chemical agent], [quantity]         Oxidizing       [chemical agent], [quantity]         Dangerous for the [chemical agent], [quantity]       [chemical agent], [quantity]         Dangerous for the [chemical agent], [quantity]       [chemi			
"100 µSv/h on contact, at the chamber walls, "S0 µSv/h in 10 cm distance. The chamber is enclosed by concrete blocks that shield radiation at the working zones to ambient background of the ISOLDE hall, S µSv/h. Experiments are remotely monitored.• Isotope• ActivityNon-ionizing radiation LaserLaserUV lightMicrowaves (300MH2:30 GH2)GH2)Chemical 200MH2Chemical agent), [quantity]Harmful[chemical agent], [quantity]Chemical Toxic[chemical agent], [quantity]CMR (carcinogens, Lick (chemical agent], [quantity]Irritant[chemical agent], [quantity]Explosiveness(chemical agent], [quantity]Dangerous for the [chemical agent], [quantity]Mechanical[chemical agent], [quantity]Physical impact or mechanical energy (moving parts)Mechanical roperties (location][location]Vehicles and Means of Icoation][location]Vehicles and Means of Icoation][location]	distance		
chamber walls, ~50 µSv/h in 10 cm distance. The chamber is enclosed by concrete blocks that shield radiation at the working zones to ambient background of the ISOLDE hall, <5 µSv/h. Experiments are remotely monitored.• Isotope• ActivityNon-ionizing radiationLaserUV lightMicrowaves (300MHz-30 GH2)GH2)Chemical 300MHz)Chemical to reproduct on the substance to reproduct on			
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concrete blocks that shield radiation at the working zones to ambient background of the ISOLDE hall, < 5 µSv/h. Experiments are remotely monitored.• Isotope		10 cm distance.	
radiation at the working zones to ambient background of the ISOLDE hall, < \$ jusyh. Experiments are remotely monitored.• Isotope• ActivityNon-ionizing radiationLaserUV lightGHz)Radiofrequency (1- 300MHz)300MHz)ChemicalToxic[chemical agent], [quantity]ChemicalCorrosive[chemical agent], [quantity]InstrainedCorrosive[chemical agent], [quantity]InstrainedCorrosive[chemical agent], [quantity]InstrainedCorrosive[chemical agent], [quantity]InstrainedCorrosive[chemical agent], [quantity]InstrainedExplosiveness[chemical agent], [quantity]Explosiveness[chemical agent], [quantity]Instrained[chemical agent], [quantity]Explosiveness[chemical agent], [quantity]Explosiveness[chemical agent], [quantity]Dangerous for the environmentMechanical environmentMechanical energy (moving parts)Mechanical energy (moving parts)Mechanical energy 		The chamber is enclosed by	
to ambient background of the ISOLDE hall, < 5 µSv/h. Experiments are remotely monitored.• Isotope• ActivityNon-ionizing radiationLaserUV lightMicrowaves (300MHz-30 GHz)GHz)Radiofrequency (1- 300MHz)300MHz)ChemicalToxic(chemical agent), [quantity]ChemicalCorrosive(chemical agent), [quantity]Irritant(chemical agent), [quantity]Physical ingect or envicomentMechanical energy (moving parts)[location]Mechanical energy (moving parts)[location]Vibration[location]Vibration[location]Vehicles and Means of Transport[location]		concrete blocks that shield	
to ambient background of the ISOLDE hall, < 5 µSv/h. Experiments are remotely monitored.• Isotope• ActivityNon-ionizing radiationLaserUV lightMicrowaves (300MHz-30 GHz)GHz)Radiofrequency (1- 300MHz)300MHz)ChemicalToxic(chemical agent), [quantity]ChemicalCorrosive(chemical agent), [quantity]Irritant(chemical agent), [quantity]Physical ingect or envicomentMechanical energy (moving parts)[location]Mechanical energy (moving parts)[location]Vibration[location]Vibration[location]Vehicles and Means of Transport[location]		radiation at the working zones	
ISOLDE hall, < 5 µSv/h. Experiments are remotely monitored.ISOLDE hall, < 5 µSv/h. Experiments are remotely monitored.• Isotope• ActivityNon-ionizing radiationLaserUV lightMicrowaves (300MH2-30 GH2)GH2)Radiofrequency (1- 300MHz)ChemicalToxic[chemical agent], [quantity]Harmful[chemical agent], [quantity]Mitages and substances toxic to reproductionCorrosive[chemical agent], [quantity]Irritant[chemical agent], [quantity]Irritant[chemical agent], [quantity]Oxidizing[chemical agent], [quantity]Explosiveness[chemical agent], [quantity]Dangerous for the envinement[chemical agent], [quantity]Dangerous for the envinement[chemical agent], [quantity]Mechanical[chemical agent], [quantity]Physical impact or mechanical energy (moving parts)[location]Mechanical properties (Sharp, rough, slippery)[location]Vibration[location][location]Vehicles and Means of Transport[location]		_	
Experiments are remotely monitored.• Isotope• ActivityNon-ionizing radiationLaserUV lightActivityMicrowaves (300MHz-30 GHz)GHz)Radiofrequency (1- 300MHz)ChemicalToxic[chemical agent], [quantity]Harmful[chemical agent], [quantity]CMR (carcinogens, toxic to reproduction)[chemical agent], [quantity]Corrosive[chemical agent], [quantity]Irritant[chemical agent], [quantity]Irritant[chemical agent], [quantity]Banmable[chemical agent], [quantity]Irritant[chemical agent], [quantity]Irritant[chemical agent], [quantity]Planmable[chemical agent], [quantity]Asphyxiant[chemical agent], [quantity]Dangerous for the environmentMechanical Properties[location]Mechanical Properties(Iocation]Wethariacl nergy (moving parts)Wethariacl nergy (moving parts)VibrationVehicles and Means of Transport			
monitored.monitored.• Isotope			
Isotope     Activity     Non-ionizing radiation Laser     UV light     UV light     Microwaves (300MHz-30     GHz)     Radiofrequency (1- 300MHz)     Chemical     Toxic     [chemical agent], [quantity]     Harmful     [chemical agent], [quantity]     CMR (carcinogens,     [chemical agent], [quantity]     UCMR (carcinogens,     [chemical agent], [quantity]     Irritant     [chemical agent], [quantity]     Irritant     [chemical agent], [quantity]     Flammable     [chemical agent], [quantity]     Explosiveness     [chemical agent], [quantity]     Asphyxiant     [chemical agent], [quantity]     Explosiveness     [chemical agent], [quantity]     Asphyxiant     [chemical agent], [quantity]     [Ammable			
Activity     Non-ionizing radiation Laser     UV light     UV light     UV light     GHz) Radiofrequency (1- 300MHz)     Chemical     Toxic     (chemical agent), [quantity] Chemical     (chemical agent), [quantity]     (chemical agent), [quantity]     (CMR (carcinogens,     mutagens and substances     toxic to reproduction)     Corrosive     (chemical agent), [quantity]     (paired)     (chemical agent), [quantity]     (paired)     (chemical agent), [quantity]     (paired)     (chemical agent), [quantity]     (chemical agent), [quantity]     (paired)     (pai		monitoreu.	
Non-ionizing radiation         Laser          UV light          Microwaves (300MHz-30          GHz)          Radiofrequency (1-          300MHz)          Chemical          Toxic       [chemical agent], [quantity]         Harmful       [chemical agent], [quantity]         Katoriogens,       [chemical agent], [quantity]         mutagens and substances          toxic to reproduction)          Corrosive       [chemical agent], [quantity]         Irritant       [chemical agent], [quantity]         Flammable       [chemical agent], [quantity]         Explosiveness       [chemical agent], [quantity]         Explosiveness       [chemical agent], [quantity]         Dangerous for the       [chemical agent], [quantity]         Physical impact or       [location]         mechanical energy       [moving parts]         Mechanical properties       [location]         (Sharp, rough, slippery)       [location]         Vibration       [location]         Vehicles and Means of       [location]			
Laser       UV light         UV light       Image: State		<u> </u>	
UV light       Improves (300MHz-30         Microwaves (300MHz-30       Improves (300MHz-30         GHz)       Improves (300MHz)         300MHz)       Improves (200MHz)         Chemical         Toxic         [chemical agent], [quantity]         Harmful       [chemical agent], [quantity]         CMR (carcinogens, mutagens and substances toxic to reproduction)       [chemical agent], [quantity]         Corrosive       [chemical agent], [quantity]         Irritant       [chemical agent], [quantity]         Irritant       [chemical agent], [quantity]         Explosiveness       [chemical agent], [quantity]         Explosiveness       [chemical agent], [quantity]         Asphyxiant       [chemical agent], [quantity]         Dangerous for the [chemical agent], [quantity]       Improves (200MHz)         environment       [chemical agent], [quantity]         Mechanical       [location]         Mechanical properties       [location]         (moving parts)       [location]         Wibration       [location]         Vehicles and Means of       [location]	Non-ionizing radiation		
Microwaves (300MHz-30 GHz)       Image: Character of the second sec	Laser		
GHz)       Image: Chemical agent (quantity)         Radiofrequency (1- 300MHz)       Image: Chemical agent (quantity)         Chemical       [chemical agent], [quantity]         Harmful       [chemical agent], [quantity]         CMR (carcinogens, mutagens and substances toxic to reproduction)       [chemical agent], [quantity]         Corrosive       [chemical agent], [quantity]         Irritant       [chemical agent], [quantity]         Flammable       [chemical agent], [quantity]         Oxidizing       [chemical agent], [quantity]         Explosiveness       [chemical agent], [quantity]         Dangerous for the environment       [chemical agent], [quantity]         Mechanical nergy (moving parts)       [location]         Mechanical properties (Sharp, rough, slippery)       [location]         Vibration       [location]         Vehicles and Means of       [location]	UV light		
GHz)       Image: Chemical agent (quantity)         Radiofrequency (1- 300MHz)       Image: Chemical agent (quantity)         Chemical       [chemical agent], [quantity]         Harmful       [chemical agent], [quantity]         CMR (carcinogens, mutagens and substances toxic to reproduction)       [chemical agent], [quantity]         Corrosive       [chemical agent], [quantity]         Irritant       [chemical agent], [quantity]         Flammable       [chemical agent], [quantity]         Oxidizing       [chemical agent], [quantity]         Explosiveness       [chemical agent], [quantity]         Dangerous for the environment       [chemical agent], [quantity]         Mechanical nergy (moving parts)       [location]         Mechanical properties (Sharp, rough, slippery)       [location]         Vibration       [location]         Vehicles and Means of       [location]	Microwaves (300MHz-30		
Radiofrequency (1- 300MHz)       Image: Chemical agent a			
300MHz)       Image: Chemical agent ag	· ·		
Chemical         Toxic       [chemical agent], [quantity]         Harmful       [chemical agent], [quantity]         CMR (carcinogens, mutagens and substances toxic to reproduction)       [chemical agent], [quantity]         Corrosive       [chemical agent], [quantity]         Irritant       [chemical agent], [quantity]         Flammable       [chemical agent], [quantity]         Oxidizing       [chemical agent], [quantity]         Explosiveness       [chemical agent], [quantity]         Dangerous for the environment       [chemical agent], [quantity]         Mechanical       [location]         Physical impact or mechanical energy (moving parts)       [location]         Mechanical properties (Sharp, rough, slippery)       [location]         Vibration       [location]         Vehicles and Means of Transport       [location]	naanon equency (1		
Toxic[chemical agent], [quantity]Harmful[chemical agent], [quantity]CMR (carcinogens, mutagens and substances toxic to reproduction)[chemical agent], [quantity]Corrosive[chemical agent], [quantity]Irritant[chemical agent], [quantity]Flammable[chemical agent], [quantity]Oxidizing[chemical agent], [quantity]Explosiveness[chemical agent], [quantity]Asphyxiant[chemical agent], [quantity]Dangerous for the environment[chemical agent], [quantity]Mechanical[location]Mechanical properties (Sharp, rough, slippery)[location]Vibration[location]Vehicles and Means of Transport[location]	300MHz)		
Harmful[chemical agent], [quantity]CMR (carcinogens, mutagens and substances toxic to reproduction)[chemical agent], [quantity]Corrosive[chemical agent], [quantity]Irritant[chemical agent], [quantity]Flammable[chemical agent], [quantity]Oxidizing[chemical agent], [quantity]Explosiveness[chemical agent], [quantity]Dangerous for the environment[chemical agent], [quantity]Mechanical[chemical agent], [quantity]Mechanical properties (Sharp, rough, slippery)[location]Vibration[location]Vehicles and Means of Transport[location]			
CMR (carcinogens, mutagens and substances toxic to reproduction)[chemical agent], [quantity]Corrosive[chemical agent], [quantity]Irritant[chemical agent], [quantity]Flammable[chemical agent], [quantity]Oxidizing[chemical agent], [quantity]Explosiveness[chemical agent], [quantity]Explosiveness[chemical agent], [quantity]Dangerous for the environment[chemical agent], [quantity]Mechanical (moving parts)[location]Mechanical properties (Sharp, rough, slippery)[location]Vibration[location]Vehicles and Means of Transport[location]	Chemical		
mutagens and substances toxic to reproduction)Image: construct of the second s	Chemical Toxic		
toxic to reproduction)(chemical agent), [quantity]Corrosive[chemical agent], [quantity]Irritant[chemical agent], [quantity]Flammable[chemical agent], [quantity]Oxidizing[chemical agent], [quantity]Explosiveness[chemical agent], [quantity]Asphyxiant[chemical agent], [quantity]Dangerous for the environment[chemical agent], [quantity]Mechanical[chemical agent], [quantity]Physical impact or mechanical energy (moving parts)[location]Mechanical properties (Sharp, rough, slippery)[location]Vibration[location]Vibration[location]	Chemical Toxic Harmful	[chemical agent], [quantity]	
Corrosive[chemical agent], [quantity]Irritant[chemical agent], [quantity]Flammable[chemical agent], [quantity]Oxidizing[chemical agent], [quantity]Explosiveness[chemical agent], [quantity]Asphyxiant[chemical agent], [quantity]Dangerous for the environment[chemical agent], [quantity]Mechanical[chemical agent], [quantity]Physical impact or mechanical energy (moving parts)[location]Mechanical properties (Sharp, rough, slippery)[location]Vibration[location]Vehicles and Means of Transport[location]	Chemical Toxic Harmful CMR (carcinogens,	[chemical agent], [quantity]	
Irritant[chemical agent], [quantity]Flammable[chemical agent], [quantity]Oxidizing[chemical agent], [quantity]Explosiveness[chemical agent], [quantity]Asphyxiant[chemical agent], [quantity]Dangerous for the environment[chemical agent], [quantity]Mechanical[chemical agent], [quantity]Physical impact or mechanical energy (moving parts)[location]Mechanical properties (Sharp, rough, slippery)[location]Vibration[location]Vehicles and Means of Transport[location]	Chemical Toxic Harmful CMR (carcinogens, mutagens and substances	[chemical agent], [quantity]	
Irritant[chemical agent], [quantity]Flammable[chemical agent], [quantity]Oxidizing[chemical agent], [quantity]Explosiveness[chemical agent], [quantity]Asphyxiant[chemical agent], [quantity]Dangerous for the environment[chemical agent], [quantity]Mechanical[chemical agent], [quantity]Physical impact or mechanical energy (moving parts)[location]Mechanical properties (Sharp, rough, slippery)[location]Vibration[location]Vehicles and Means of Transport[location]	Chemical Toxic Harmful CMR (carcinogens, mutagens and substances	[chemical agent], [quantity]	
Flammable       [chemical agent], [quantity]         Oxidizing       [chemical agent], [quantity]         Explosiveness       [chemical agent], [quantity]         Asphyxiant       [chemical agent], [quantity]         Dangerous for the environment       [chemical agent], [quantity]         Mechanical       [chemical agent], [quantity]         Physical impact or mechanical energy (moving parts)       [location]         Mechanical properties (Sharp, rough, slippery)       [location]         Vibration       [location]         Vehicles and Means of Transport       [location]	Chemical Toxic Harmful CMR (carcinogens, mutagens and substances toxic to reproduction)	[chemical agent], [quantity] [chemical agent], [quantity]	
Oxidizing[chemical agent], [quantity]Explosiveness[chemical agent], [quantity]Asphyxiant[chemical agent], [quantity]Dangerous for the environment[chemical agent], [quantity]Mechanical[chemical agent], [quantity]Physical impact or mechanical energy (moving parts)[location]Mechanical properties (Sharp, rough, slippery)[location]Vibration[location]Vehicles and Means of Transport[location]	Chemical Toxic Harmful CMR (carcinogens, mutagens and substances toxic to reproduction) Corrosive	[chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity]	
Explosiveness[chemical agent], [quantity]Asphyxiant[chemical agent], [quantity]Dangerous for the environment[chemical agent], [quantity]MechanicalPhysical impact or mechanical energy (moving parts)[location]Mechanical properties (Sharp, rough, slippery)[location]Vibration[location]Vehicles and Means of Transport[location]	Chemical Toxic Harmful CMR (carcinogens, mutagens and substances toxic to reproduction) Corrosive Irritant	[chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity]	
Asphyxiant       [chemical agent], [quantity]         Dangerous for the environment       [chemical agent], [quantity]         Mechanical       [chemical agent], [quantity]         Physical impact or mechanical energy (moving parts)       [location]         Mechanical properties (Sharp, rough, slippery)       [location]         Vibration       [location]         Vehicles and Means of Transport       [location]	Chemical Toxic Harmful CMR (carcinogens, mutagens and substances toxic to reproduction) Corrosive Irritant Flammable	[chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity]	
Dangerous for the environment       [chemical agent], [quantity]         Mechanical         Physical impact or mechanical energy (moving parts)       [location]         Mechanical properties (Sharp, rough, slippery)       [location]         Vibration       [location]         Vehicles and Means of Transport       [location]	Chemical Toxic Harmful CMR (carcinogens, mutagens and substances toxic to reproduction) Corrosive Irritant Flammable Oxidizing	[chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity]	
environment     Image: Constraint of the second secon	Chemical Toxic Harmful CMR (carcinogens, mutagens and substances toxic to reproduction) Corrosive Irritant Flammable Oxidizing Explosiveness	[chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity] [chemical agent], [quantity]	
Mechanical         Physical impact or mechanical energy (moving parts)       [location]         Mechanical properties (Sharp, rough, slippery)       [location]         Vibration       [location]         Vehicles and Means of Transport       [location]	Chemical Toxic Harmful CMR (carcinogens, mutagens and substances toxic to reproduction) Corrosive Irritant Flammable Oxidizing Explosiveness Asphyxiant	[chemical agent], [quantity] [chemical agent], [quantity]	
Physical impact or mechanical energy (moving parts)       [location]         Mechanical properties (Sharp, rough, slippery)       [location]         Vibration       [location]         Vehicles and Means of Transport       [location]	Chemical Toxic Harmful CMR (carcinogens, mutagens and substances toxic to reproduction) Corrosive Irritant Flammable Oxidizing Explosiveness Asphyxiant Dangerous for the	[chemical agent], [quantity] [chemical agent], [quantity]	
mechanical energy (moving parts)     Image: Constraint of the sector of th	Chemical Toxic Harmful CMR (carcinogens, mutagens and substances toxic to reproduction) Corrosive Irritant Flammable Oxidizing Explosiveness Asphyxiant Dangerous for the environment	[chemical agent], [quantity] [chemical agent], [quantity]	
(moving parts)     Image: Constraint of the second se	Chemical Toxic Harmful CMR (carcinogens, mutagens and substances toxic to reproduction) Corrosive Irritant Flammable Oxidizing Explosiveness Asphyxiant Dangerous for the environment Mechanical	[chemical agent], [quantity] [chemical agent], [quantity]	
Mechanical properties (Sharp, rough, slippery)       [location]         Vibration       [location]         Vehicles and Means of Transport       [location]	ChemicalToxicHarmfulCMR (carcinogens, mutagens and substances toxic to reproduction)CorrosiveIrritantFlammableOxidizingExplosivenessAsphyxiantDangerous for the environmentMechanicalPhysical impact or	[chemical agent], [quantity] [chemical agent], [quantity]	
Mechanical properties (Sharp, rough, slippery)       [location]         Vibration       [location]         Vehicles and Means of Transport       [location]	ChemicalToxicHarmfulCMR (carcinogens, mutagens and substances toxic to reproduction)CorrosiveIrritantFlammableOxidizingExplosivenessAsphyxiantDangerous for the environmentMechanicalPhysical impact or	[chemical agent], [quantity] [chemical agent], [quantity]	
(Sharp, rough, slippery)     Image: Constant of the state	ChemicalToxicHarmfulCMR (carcinogens, mutagens and substances toxic to reproduction)CorrosiveIrritantFlammableOxidizingExplosivenessAsphyxiantDangerous for the environmentMechanicalPhysical impact or mechanical energy	[chemical agent], [quantity] [chemical agent], [quantity]	
Vibration     [location]       Vehicles and Means of Transport     [location]	ChemicalToxicHarmfulCMR (carcinogens, mutagens and substances toxic to reproduction)CorrosiveIrritantFlammableOxidizingExplosivenessAsphyxiantDangerous for the environmentMechanicalPhysical impact or mechanical energy (moving parts)	[chemical agent], [quantity]	
Vehicles and Means of [location] Transport	ChemicalToxicHarmfulCMR (carcinogens, mutagens and substances toxic to reproduction)CorrosiveIrritantFlammableOxidizingExplosivenessAsphyxiantDangerous for the environmentMechanicalPhysical impact or mechanical energy (moving parts)Mechanical properties	[chemical agent], [quantity]	
Transport	ChemicalToxicHarmfulCMR (carcinogens, mutagens and substances toxic to reproduction)CorrosiveIrritantFlammableOxidizingExplosivenessAsphyxiantDangerous for the environmentMechanicalPhysical impact or mechanical energy (moving parts)Mechanical properties (Sharp, rough, slippery)	[chemical agent], [quantity]         [chemical agent], [quantity]	
	ChemicalToxicHarmfulCMR (carcinogens, mutagens and substances toxic to reproduction)CorrosiveIrritantFlammableOxidizingExplosivenessAsphyxiantDangerous for the environmentMechanicalPhysical impact or mechanical energy (moving parts)Mechanical properties (Sharp, rough, slippery)Vibration	[chemical agent], [quantity]         [location]         [location]	
NOISE	ChemicalToxicHarmfulCMR (carcinogens, mutagens and substances toxic to reproduction)CorrosiveIrritantFlammableOxidizingExplosivenessAsphyxiantDangerous for the environmentMechanicalPhysical impact or mechanical energy (moving parts)Mechanical properties (Sharp, rough, slippery)VibrationVehicles and Means of	[chemical agent], [quantity]         [location]         [location]	
	ChemicalToxicHarmfulCMR (carcinogens, mutagens and substances toxic to reproduction)CorrosiveIrritantFlammableOxidizingExplosivenessAsphyxiantDangerous for the environmentMechanicalPhysical impact or mechanical energy (moving parts)Mechanical properties (Sharp, rough, slippery)VibrationVehicles and Means of Transport	[chemical agent], [quantity]         [location]         [location]	

Frequency	[frequency],[Hz]	
Intensity		
Physical		
Confined spaces	[location]	
High workplaces	[location]	
Access to high workplaces	[location]	
Obstructions in	[location]	
passageways		
Manual handling	[location]	
Poor ergonomics	[location]	

#### 0.1 Hazard identification

3.2 Average electrical power requirements (excluding fixed ISOLDE-installation mentioned above): (make a rough estimate of the total power consumption of the additional equipment used in the experiment)

About 5 kW for the on-line chamber at GHM. Each of the 3 off-line setups about 2 kW.