

Differences between amyloid- β aggregation in solution and on the membrane: insights into elucidation of the mechanistic details of Alzheimer's disease

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The association of the amyloid- β ($A\beta$) peptide with cellular membranes is hypothesized to be the underlying phenomenon of neurotoxicity in Alzheimer's disease. Misfolding of proteins and peptides, as is the case with $A\beta$, follows a progression from a monomeric state, through intermediates, ending at long, unbranched amyloid fibers. This tutorial review offers a perspective on the association of toxic $A\beta$ structures with membranes as well as details of membrane-associated mechanisms of toxicity.

Key learning points

This review seeks to explain/explore:

- (1) The amyloid hypothesis and how it relates to the toxicity of $A\beta$ assemblies.
- (2) How the toxicity of $A\beta$ is linked to both AD as well as mechanisms of membrane disruption.
- (3) The current structural models of amyloid intermediates.
- (4) The influence of the cell membrane on the changes in secondary structure along the misfolding pathway of $A\beta$.
- (5) The differences in the prominent mechanisms of membrane disruption: ion channel-like pores and membrane fragmentation.

1.0 Introduction

Amyloid deposits characterize more than twenty different clinical syndromes, each of which is associated with a distinct amyloid-forming protein. Neurodegenerative diseases such as Alzheimer's, Parkinson's, and Huntington's diseases, in addition to localized diseases such as Type II Diabetes, are associated with the aggregation and misfolding of amyloidogenic peptides/proteins. Of these amyloid-related diseases, Alzheimer's disease (AD) is the most known, affecting 5.2 million people in the United States and more than 24 million people worldwide.¹ AD is clinically characterized by the presence of intracellular neurofibrillary tangles and extracellular senile plaques.² The plaques consist of insoluble amyloid deposits composed primarily of aggregates of amyloid-beta ($A\beta$) in their fibril form. The $A\beta$ peptide is produced through proteolytic cleavage of the amyloid precursor protein (APP) by the β - and γ -secretases.³ The predominant $A\beta$ species formed are the $A\beta_{1-40}$ and $A\beta_{1-42}$ peptides (consisting of 40 and 42 residues, respectively);

with the $A\beta_{1-42}$ variant being an indicator of a progressive AD state.⁴ The accumulation of $A\beta_{1-40}$ and $A\beta_{1-42}$ in long, unbranched fibrils is a hallmark of the disease, as is the loss of neurons due to cell death in parallel with the $A\beta$ aggregation process.^{2,5} Amyloid fibrils of $A\beta_{1-40}$ and $A\beta_{1-42}$ form a parallel, in-register cross β -sheet structure that binds to fibril-specific dyes such as congo red and thioflavin-T.⁶ Prior to fibril formation, $A\beta$ forms a myriad of structures in the monomeric and oligomeric states, all of which result in similar fibril structures.³ Neurotoxicity by $A\beta$ is convoluted as it is unknown which misfolded species causes cell death, in addition to the mechanism by which a particular misfolded state causes toxicity. These data have been correlated with brain samples from patients with and without AD, and, to this day, the $A\beta$ fibrils define and confirm the diagnosis of AD patients' post-mortem.⁴

Early AD research pointed to $A\beta$ fibrils specifically as the neurotoxic agent leading to cellular death, memory loss, and other AD characteristics.⁷ Over the last two decades, further investigation has suggested that the fully-matured fibrils are no longer considered to be the main toxic agent; rather, oligomeric, prefibrillar species of the $A\beta$ peptide have been shown to be most damaging to neuronal cells.^{3,5,8-10} There are several hypotheses regarding the mode of $A\beta$ toxicity and they include

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1 the generation of reactive oxygen species, interaction with cell
receptors, interactions with metals and direct disruption of
cellular membranes (Fig. 1).⁵ One widely accepted theory, called
5 the amyloid hypothesis, deals with the notion that misfolded
intermediate states are responsible for cell death.³ Therefore, it is
important to establish the relationship between misfolded enti-
ties, their structural properties, and how they confer toxicity.

1.1 Structurally dissimilar A β oligomers give rise to varied neuronal toxicity

While the toxicity of A β is under constant debate, the amyloid hypothesis has been supported by bodies of work suggesting that amyloid oligomers are the toxic amyloid species.¹¹ The toxicity of soluble amyloid oligomers is not only relevant to AD,

but to other amyloid diseases as well such as Parkinson's disease and Type-II Diabetes where α -synuclein and human islet amyloid polypeptide (hIAPP) are indicated as the aggregative proteins, respectively.^{5,12} In the case of A β , it is known that amyloid oligomers are toxic *in vitro*,¹³ with further evidence identifying oligomers as the culprit for *in vivo* toxicity.¹⁴ The problem herein arises due to the fact that oligomers ranging in size and structural morphology have exhibited cytotoxicity.¹⁵

A wealth of structural data exists for the relatively benign fibrils of A β ;⁶ however, intermediate structures such as structured monomers and oligomers have remained relatively unexplored, largely due to their transient nature. In this regard, the inability to crystallize and/or trap a pure (single state) sample has presented much of the difficulty in performing



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Jeffrey R. Brender

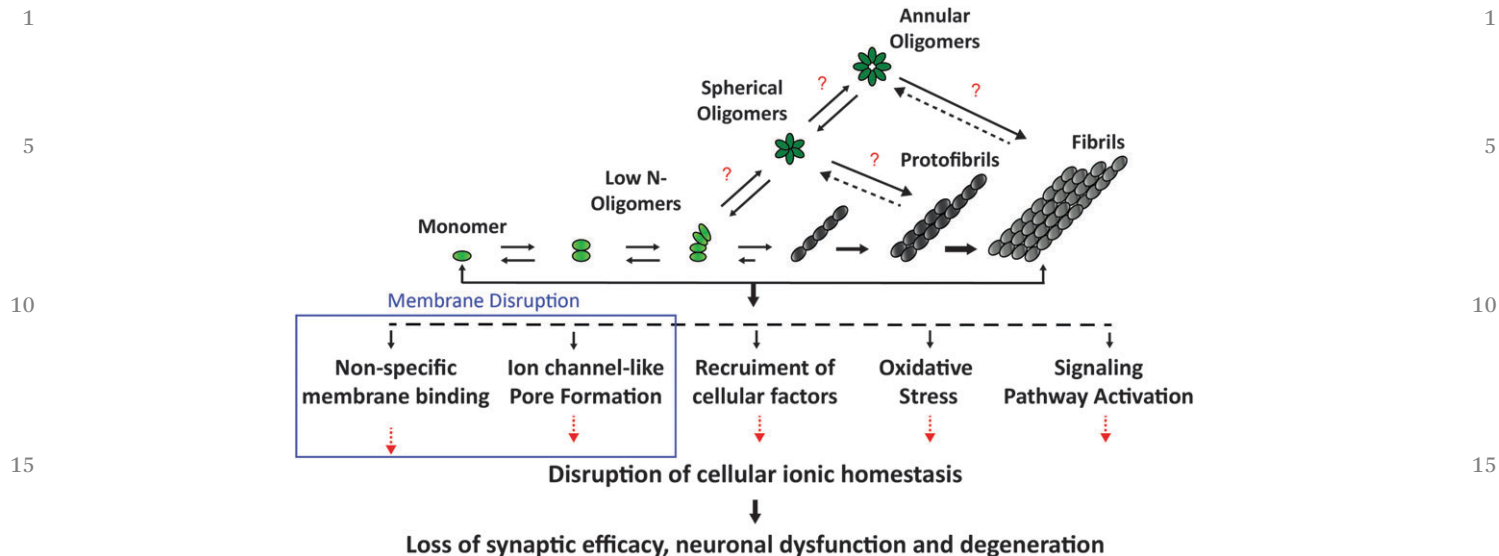
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revealed atomic-level mechanisms of membrane permeation/disruption by amyloid peptides and antimicrobial peptides, and reported the high-resolution structure of membrane proteins. More details about his current research can be found at <http://www.umich.edu/~ramslab>.



Q4 Fig. 1 The amyloid hypothesis for Alzheimer's disease. The aggregation of amyloid- β is highly diverse and poorly understood. Mounting evidence points to oligomers as being the most toxic agent in Alzheimer's disease; however, intermediate structures are transient and heterogeneous. Additionally, the mechanism by which A β can be neurotoxic has not been fully elucidated. One prevailing hypothesis suggests that A β can be toxic through a membrane disruption mechanism.

structural studies of oligomeric species. Relatively low-resolution methods, such as AFM, hydrogen-deuterium exchange mass spectrometry, and circular dichroism (CD) have shown the wide size distribution of oligomers and variance in secondary structure. To further complicate matters, increasing the concentration to levels which are more amenable to atomic-resolution structural experiments can affect the aggregation time, decreasing the longevity of the oligomeric sample. Consequently, the sensitivity of aggregation results from a multitude of factors generating heterogeneous samples and hampering structural studies by techniques like nuclear magnetic resonance (NMR) and X-ray crystallography.

The larger oligomers of A β have been mostly described as spherical, although studies have shown that circular, annular oligomers are producible *in vitro*.^{16,17} One interesting aspect of A β oligomers and amyloid oligomers in general, is their ability to bind a common antibody regardless of structure. This ability

to bind the A11 antibody indicates that amyloid oligomers across diseases and protein sequences share a common structure.¹³ This common β -sheet-containing structure is fascinating, especially given structural evidence that A β_{1-40} forms a 3_{10} helix in solution (Fig. 2a), meaning a helix-to- β -sheet transition may occur.¹⁸ That being said, secondary structure-dependent toxicity has been the source of some debate. While most of the oligomers studied to date have contained β -sheet secondary structure, for relatively small oligomers of A β_{1-42} , it was determined that hydrophobic exposure and not the presence of β -sheets is required for toxicity. The study also found that aggregate number is also not necessarily linked to toxicity; two similar sized globular oligomers were characterized and the more-hydrophobic exposed oligomer was toxic.¹⁹

Some structural studies have utilized detergents or small molecules to stabilize both large²⁰ and small²¹ oligomeric states with some success while others have utilized the freeze-trapping

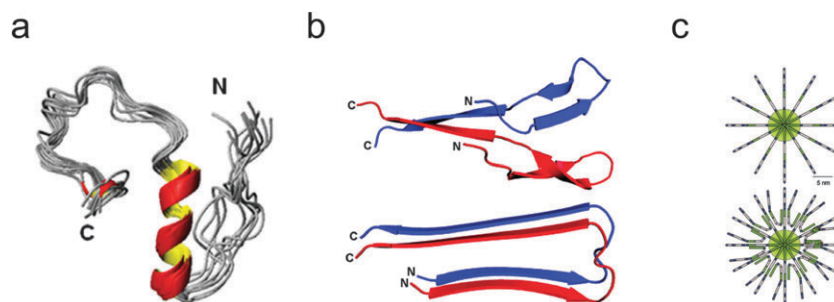


Fig. 2 Various structures of A β in solution and detergents characterized by NMR. (a) A β_{1-40} as a partially folded structure in the presence of 50 mM NaCl, with residues 13 to 23 forming a 3_{10} helix.¹⁸ (b) Solution NMR structure of a 0.05% SDS-stabilized pre-globulomer of A β_{1-42} (top) compared with the basic fold of the fibrils of A β_{1-42} (bottom). Adapted with permission from Yu *et al.*²¹ Copyright 2009 American Chemical Society. (c) Structural schematic of β -balls formed at low pH in the absence (top) and presence (bottom) of DSS. These structures both show a pinwheel or micelle-like arrangement of monomers. Reprinted with permission from Laurents *et al.* 2005. Copyright 2005 Journal of Biological Chemistry.²⁰

of purified oligomers.⁹ Very large oligomers formed at low pH, described as amyloid “ β -balls” and having a molecular weight of 764 kDa, have been studied at low-resolution. These very large assemblies can also be stabilized by DSS (4,4-dimethyl-4-silapentane-1-sulfonic acid), and are hypothesized to have a pin-wheel-like structure, with monomers protruding radially outward (Fig. 2c).²⁰ In the case of freeze-trapped large oligomers of $A\beta_{1-40}$, structural studies by solid state NMR have identified fibril-like contacts.⁹ These large spherical oligomers contain primarily β -sheet secondary structure and are toxic to cultured neurons. Smaller $A\beta_{1-42}$ oligomers, thought to be pentamers, show a more loose arrangement with turns and C-termini contained in the center of the oligomer.¹⁰ Polymerization of these “disk-shaped” oligomers into fibrils is thought to occur by association and lateral extension of parallel β -strands into fibrils. These pentamers are also toxic to cultured mouse neurons, even in the absence of β -sheet secondary structure. The collective theme associated with the three aforementioned structures is the idea of a pinwheel, or micelle-like, structure which may confirm the notion that oligomers of various size and sequence share a common structural element.

Smaller oligomers, coined as pre-globulomers and globulomers, of $A\beta_{1-42}$ described by Yu *et al.* show similar molecular contacts to amyloid fibers of $A\beta$ (Fig. 2b).²¹ The pre-globulomers (16 kDa) and globulomers (64 kDa) were stabilized by 0.05% sodium dodecyl sulphate (SDS) and studied by solution NMR. As with larger freeze trapped oligomers of $A\beta_{1-40}$, the identification of these fibril-like structures provides a basis for the conversion of oligomers to fibrils, but do not directly explain the ability of $A\beta$ to permeabilize cell membranes.

Overall, high-resolution reports of fibril-like contacts for spherical oligomers of $A\beta_{1-40}$ have shown parallel β -sheets⁹ while oligomers of $A\beta_{1-42}$ demonstrated a lack of β -sheet secondary structure,¹⁰ highlighting the important difference observed between the structure and toxicity of these $A\beta$ alloforms. Others have pointed to oligomers containing anti-parallel β -sheets for the formation of pores and thus, the permeabilization of cell membranes.²²

Annular oligomers, or annular protofibrils, are 8–20 nm in diameter and contain, as with spherical oligomers, a high β -sheet secondary structure.¹⁶ Despite the β -sheet content, circular oligomers are quite different from their spherical counterparts. It has been shown that these annular oligomers may share a common structure with β -barrel pores, such as α -hemolysin, through the binding of an anti-annular protofibril antibody to assembled α -hemolysin.¹⁶ Despite having a preformed pore-like structure, these circular oligomers do not permeabilize membranes, but rather convert to spherical oligomers in the presence of lipids. Regardless of their inability to permeabilize membranes, annular protofibrils point to the possibility of a β -barrel pore as the $A\beta$ pore structure.

The relationship between structure and cellular toxicity has been of primary focus in recent amyloid research. Given the number of reportedly different oligomeric $A\beta$ species, relating these parameters could provide insights into how we might go about preventing unfavorable folds from occurring. Despite the propensity of $A\beta$ to form fibril-like intermediates early and late

in its aggregation pathway, existing evidence suggests that the fold of toxic oligomers may differ considerably from that of the $A\beta$ fibril.^{15,23} As a result of the variance in secondary structure between different oligomer preparations, $A\beta$ toxicity need not derive from a ‘single’ toxic species. In a recent study, Lashuel and co-workers argued that $A\beta$ toxicity results from an “ongoing polymerization process” by demonstrating that a heterogeneous mixture of monomers and protofibrils was prone to heightened aggregation and, in turn, cytotoxicity.²⁴ Neurotoxicity occurring by such a mechanism would only present greater difficulties in seeking therapeutic measures for alleviating AD symptoms, and attributes greater stress to understanding misfolding pathways as opposed to identifying individual toxic structures.

While it is important to understand the structural link between intermediates and fibrils, their ultimate mode of action should be considered. Therefore, it is of utmost importance to gain structural and functional information on intermediates at their site of action, namely the membrane. Solid state NMR has proven to be an effective tool for the characterization of $A\beta$ oligomer structures in the absence of the membrane. This technique has been a cornerstone of lipid biophysics as well, allowing researchers to effectively determine the integrity and dynamics of the lipid bilayer. With solid state NMR having been recently used to solve the structures and dynamics of several membrane proteins²⁵ and membrane disrupting antimicrobial peptides,²⁶ it stands to reason that this technique will be on the forefront of structural determination of $A\beta$ in the membrane environment.

2.0 The role of membrane composition and disruption in AD pathogenesis

A number of studies have identified various possible mechanisms for cytotoxicity mediated by $A\beta$, and a prominent focus on AD pathology has centered on $A\beta$'s ability to disrupt membranes.⁵ In order to be cytotoxic *via* membrane disruption, $A\beta$ aggregates must interact with the cellular surface by either a receptor or the lipid membrane. The mechanism of $A\beta$ -membrane interactions has not been fully elucidated; however, its interaction with the membrane is likely to affect the structure and properties of any type of aggregate. Dysregulation of ionic homeostasis, particularly of Ca^{2+} , has been a consistent pathology in AD; thus, understanding the molecular mechanics by which $A\beta$ induces ionic flux has become crucial to AD pathology.²⁷ The cellular membrane in live cells is exceedingly complex, involving many variables that are difficult to isolate and control. Consequently, studying structural transitions of $A\beta$ in the presence of membranes (with varying composition) and the ionic flux across a cellular membrane that results due to specific $A\beta$ -membrane interactions has been extremely challenging.

2.1 Ganglioside-containing membranes influence structural changes during $A\beta$ aggregation

Although obtaining structural models of $A\beta$ oligomers is paramount to understanding the pathology of AD, the presence of membranes has been shown to strongly influence the $A\beta$ aggregation pathway.

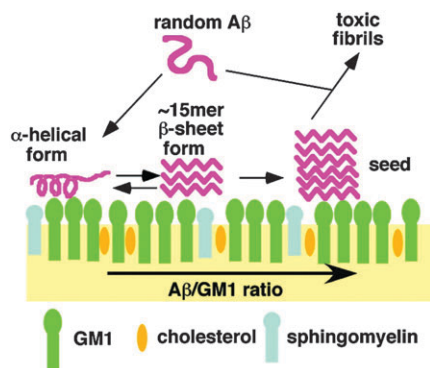
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Fig. 3 Model of GM1 ganglioside-clusters leading to the formation of toxic A β species. Studies have found that low A β :GM1 ratios yield an α -helical A β structure, while an increasing A β :GM1 ratio produces fibrils. Reprinted with permission from Ikeda *et al.* 2011. Copyright 2011 American Chemical Society.³³

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While a number of studies have reported on the A β -membrane interaction, a great body of evidence points to gangliosides playing a significant role in altering the aggregation pathway of A β .^{28,29} Gangliosides are glycosphingolipids that contain a sialic acid headgroup moiety and play a substantial role in cell signaling and memory function. Early work identified A β fibrils tightly bound to monosialoganglioside (GM1), generating a conformationally distinct A β species.³⁰ Since this work, a series of *in vitro* and *in vivo* studies demonstrated that GM1 strongly influences the secondary structure of A β in the early and late stages of aggregation.³¹ Initial biophysical characterization demonstrated the pivotal role the sialic acid played in inducing such structural transitions.³² Work along this direction from the Matsuzaki group showed that structural changes of A β upon interaction with GM1 were dependent upon the A β :GM1 ratio—for a low A β :GM1 ratio A β assumes an α -helix conformation, while at a high A β :GM1 ratio fibril formation ensues (Fig. 3).^{29,33} A great body of evidence from the Matsuzaki group has demonstrated that A β amyloidogenicity can be altered in the presence of GM1-containing membranes, with this effect being amplified by introducing A β into a lipid raft-like membrane mimetic. Cell viability assays showed that co-incubation of A β_{1-40} with GM1 in a lipid raft-like environment generated toxic amyloid species. Furthermore, concurrent measurements of the accumulation of A β and oligomer formation at the cell membrane were performed using TIRF microscopy, revealing that membrane-mediated aggregation affects fibril morphology.³⁴

The enhanced binding of A β to gangliosides is largely dependent upon the sialic acid of the headgroup moiety. Yet, McLaurin and Chakrabarty demonstrated that neither the ceramide nor the sialic acid alone could induce a partial helical structure; rather, the structural transition brought about by gangliosides is dependent upon the association of the sialic acid with the carbohydrate backbone.³² Recently, an NMR study came to a similar conclusion using the paramagnetic relaxation enhancement (PRE) effect to identify A β binding to GM1 at the sugar-ceramide junction.³⁵ In a related study, Williamson *et al.*

employed the use of chemical shift perturbations of uniformly ¹⁵N-labeled A β_{1-40} upon titration of GM1 and asialo-GM1 micelles as monitored by HSQC (heteronuclear single-quantum coherence) and revealed: (i) the presence of the sialic acid in the ganglioside headgroup enhanced A β -GM1 binding, (ii) A β binding to GM1 was N-terminally driven, and (iii) A β association is localized to His13 and Leu17.³⁶ In accordance with these data, in a separate NMR study using ¹H-¹⁵N TROSY and TROSY-based saturation transfer experiments, it was found that A β assumes a partial helical structure with an “up and down topological mode” when bound to GM1 micelles.³⁷ The structure was not determined by this study from Utsumi *et al.*; however, it was found through backbone chemical shifts that the regions of A β bound to GM1 micelles assume discontinuous α -helices between residues His14-Val24 and Ile31-Val26 (other regions of the peptide being unstructured).

The studies from Williamson *et al.* and Utsumi *et al.* seem to generate conflicting results of A β binding to gangliosides being N- or C-terminally driven. Yet, one can imagine a multi-step process involving initial electrostatic interactions with the N-terminus and hydrophobic interactions driving binding and/or insertion of the C-terminus into the GM1-containing membrane. The early biophysical characterization of the A β -GM1 interaction demonstrating the subtle dependence on the peptide:lipid ratio was corroborated by the more recent NMR studies. Collectively, these data provide a better understanding of the production of structurally dissimilar A β aggregation states and heterogeneity in the aggregation pathway; the central dogma of the amyloid problem. It is well known that A β membrane binding is amplified by the presence of negatively charged lipid headgroups. Nonetheless, answering questions as to what makes for favorable binding to gangliosides and how gangliosides generate toxic amyloid species remain inadequate in the broader scope of AD pathology. While there is still much to be learned about the A β -ganglioside interaction, recently progress has been made in understanding how gangliosides might play a role in the mechanism of A β toxicity.

2.2 Gangliosides mediate a two-step mechanism of amyloid- β membrane disruption

Studying the interactions between A β and cellular membranes has proven to be of great importance, yet more insight must be gained by understanding the mechanism by which these oligomers can be toxic to cells. A β generates a multifactorial response in neurons, making the study of A β 's cytotoxicity difficult to pinpoint. While the exact mechanism of A β neuronal toxicity remains elusive, one of the long-standing pathologies is believed to arise from increased levels of cytosolic Ca²⁺ resulting from plasma membrane disruption.³⁸ A number of *in vitro* studies over the last two decades have demonstrated that ionic dyshomeostasis can be caused by A β 's ability to form ion channel-like pores: annular oligomer structures that allow ions to permeate the membrane through pores of a hydrophilic interior and hydrophobic exterior.²⁷ The resulting unregulated influx of Ca²⁺ is a result of the general properties of A β pores being cation selective and capable of being blocked by Zn²⁺,

1 suggesting that these pores adopt a specific structure.^{38–40} Yet, there is little information on the structure adopted by A β upon its insertion into the membrane. Studies have suggested that pore structures contain a β -sheet rich conformation which is consistent with the currently available structural data for A β ; particularly the common U-shaped motif found in amyloid structures.⁴¹ The concept of an explicit pore structure is further supported by a study by Capone *et al.*, which demonstrated that A β ion channels could be modulated by non-natural amino acid substitutions and emphasized the necessity of secondary structure for such channels to form.⁴² In a related study from Lal and colleagues, A β with an all D-amino acid sequence retains strikingly similar features known for the all L-amino acids A β , demonstrating that chirality had no bearing on A β channel-like activity.⁴³ The results obtained using the all D-amino acids A β peptide are quite intriguing; however, *in vivo* toxicity measurements would provide a more definitive answer as to whether stereospecific, non-stereospecific, or both mechanisms are involved in membrane perturbations. Nevertheless, both neurons and mitochondria are highly sensitive to perturbations in ionic strength, and a small perturbation in intracellular calcium levels caused by unregulated A β channel activity can trigger an apoptotic cascade.

The amyloid channel hypothesis provides much insight into the membrane disruption mechanism by A β and other amyloid peptides; however, the mechanism by which A β stimulates ionic dyshomeostasis has not been fully elucidated. A multivariate analysis of A β cytotoxicity found a negative correlation with both the particular cross- β sheet structure of amyloid fibers as well as the overall β -sheet content, contrary to the predicted models of A β pore structures.⁴⁴ Moreover, spherical aggregates of A β have repeatedly shown to exhibit cytotoxicity,^{45,46} and it is difficult to fathom how such structural polymorphs could form an A β ion channel. Recently, Sciacca *et al.* demonstrated that membrane disruption by A β involves a two-step mechanism: (i) A β oligomers bind to the membrane to form ion permeable pores and (ii) the process of A β fibrillization causes membrane fragmentation *via* a detergent-like mechanism (Fig. 4).⁴⁷ The first phase of this proposed two-step mechanism (Fig. 4b) shares many of the qualities observed

for the previously described A β channel structures; namely, cation selectivity and the ability to be blocked by Zn²⁺. The second step was found to be correlated with the formation of A β fibrils as revealed by leakage of the dye 6-carboxyfluorescein from large unilamellar vesicles (LUVs) and time course ThT measurements. Further examination of the fiber-dependent form of membrane disruption by ³¹P NMR and a lipid sedimentation assay led to the conclusion that this mechanism acted in a detergent-like manner. Unlike the A β pores, the second phase of membrane disruption displayed neither charge nor size selectivity, lending support to the loss of the integrity of the membrane (Fig. 4c). Interestingly, completion of fibril polymerization halted the progression of further membrane fragmentation, and, perhaps more intriguing, this phase of membrane disruption was entirely dependent upon gangliosides being part of the membrane composition. Given the overwhelming evidence for the effect of ganglioside-containing membranes on A β aggregation, the finding of a relationship between aggregation and membrane disruption is a step forward in bridging ideals for further elucidation of the mechanism. In a similar train of thought, fiber-dependent membrane disruption can be correlated with the results previously discussed by Lashuel that ongoing polymerization was a key factor in cytotoxicity.

It should be noted that both the pore and fiber-dependent mechanisms are not strictly characteristic of A β . For example, IAPP displays a similar biphasic effect in membrane disruption, and our group demonstrated that the fiber-dependent membrane permeabilization could be prevented by insulin⁴⁸ and both mechanisms could be modulated by the presence of PE lipids in the membrane composition.⁴⁹ Furthermore, α -synuclein has been implicated as a membrane disrupting protein by pore formation and membrane fragmentation, in addition to other amyloidogenic proteins.^{39,50} While there is an abundance of existing biophysical and biochemical data to support the amyloid hypothesis for A β and other amyloids, much more characterization needs to be done; namely, in the area of the therapeutic prevention of membrane disruption and further elucidation of the structure of misfolded intermediate states.

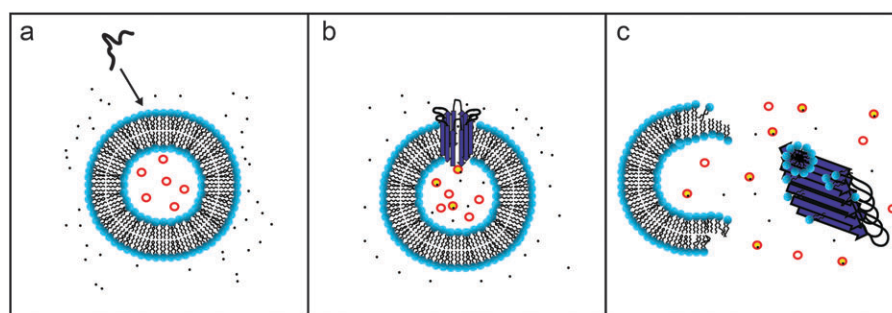


Fig. 4 A two-step mechanism of membrane disruption. Upon addition of A β to a membrane solution (a), it is capable of binding to the membrane and forming ion channel-like pores (b). Appearance of A β pores is increased by the presence of gangliosides in the membrane composition. Furthermore, gangliosides mediate a second step of membrane disruption, a fiber-dependent step, which acts *via* a detergent-like mechanism to fragment the lipid bilayer (c).

2.3 Preventing membrane disruption by A β

Small molecule compounds blocking A β channel formation and other forms of membrane disruption have the potential to alleviate A β -induced cytotoxicity. Current amyloid inhibitors target A β oligomers and fibrils outside the membrane.^{51,52} For A β structures formed on and in the membrane this can be problematic as the small molecule EGCG has been shown to be less effective at the membrane surface.⁵³ Given A β 's reactivity with the membrane, it is greatly important that small inhibitors such as EGCG have equal or superior efficacy at the peptide-membrane interface. Inhibitors of channel formation would therefore seem to be a very attractive target for stopping A β cytotoxicity. Such small molecules designed by Arispe and co-workers protected against the neurotoxic effect of A β and blocked ion conductance activity in model membranes.⁵⁴ Unfortunately, the number of known channel inhibitors of A β is currently very limited. Since the A β channel is formed by the oligomerization of A β , aggregation inhibitors are likely to influence A β channel formation. However, many aggregation inhibitors are specific to the fibrillar form of A β and therefore may not have activity against the specific oligomerization process that creates A β channels. Thus, an attractive approach would be to seek a set(s) of molecules to synergistically prevent all possible modes of membrane disruption by screening known inhibitors of A β aggregation as inhibitors of A β membrane disruption. If pore and oligomer formation are at least partially independent, as our previous data indicate, the elimination of either process alone will be insufficient to stop A β toxicity. However, the addition of both pore blockers and aggregation inhibitors simultaneously may eliminate the cellular dysfunction caused by A β . Given the common pathology associated with many amyloidogenic proteins, this avenue of research seems promising in light of the amyloid hypothesis.

3.0 Obtaining high-resolution structures of membrane-bound A β species

Due to the inability to crystallize stable and homogeneous A β preparations on membranes, the use of computational techniques has become the principal mode to model A β structure in membranes.^{27,41} Yet, the details of A β -membrane interactions and the membrane disruption process are not merely of theoretical importance, but have great practical implications for the prevention of membrane damage by amyloidosis. Membrane damage that is primarily mediated by discrete ion channels can be alleviated by channel blockers designed specifically to plug the channels formed by toxic amyloid species. Such small molecules were designed with the intention of complementary binding to distinct A β channel structures, thereby reducing cytotoxic effects and strengthening the amyloid channel hypothesis.⁵⁴ However, prevention of a non-specific mechanism involving membrane fragmentation requires a different approach aimed at blocking the interaction of protofibrillar

A β with the membrane. To accomplish this it is necessary to have some understanding of the interactions involved in binding of A β to the membrane in fibrillar and prefibrillar conformations and some idea of the conformation of the membrane-bound peptide in a given aggregation state. Unfortunately, while many of the mechanistic details of A β amyloid formation in solution remain to be determined, experimental evidence of structural polymorphisms involved in amyloid formation on or in a membrane beyond the resolution of CD, ThT fluorescence, and other biophysical measurements are almost completely unknown.

Currently, the only membrane bound models of A β that exist are high-resolution structures of A β monomers bound to detergent micelles and computer simulations of channels or oligomers constructed from fragments of the A β fiber.^{27,41,55} The computer simulations of A β channels have been invaluable in identifying molecular features of potential A β pore structures, yet more structural data are necessary to further refine computational models and to collectively understand all forms of membrane disruption. It is apparent that membranes have a two-fold effect on A β : (i) membranes can greatly accelerate the rate of fibrillization and (ii) A β can directly disrupt plasma and possibly organelle membranes. The cell membrane therefore contributes to A β toxicity as both a site for the accumulation/nucleation of toxic oligomers and as a target for their cytotoxic effect. Consequently, it is imperative to better understand how environmental factors within the cell membrane (such as the lipid composition) contribute to structural polymorphisms of A β oligomers. Such data will help us understand both how membrane disruption occurs and possibly if certain cell types are more sensitive to A β toxicity due to an altered lipid composition, generating a more toxic conformation.

4.0 Conclusions

The dependence on gangliosides in the membrane disruption process corroborates previous results in that they clearly influence an alternative pathway for A β aggregation. A β polymerization in the presence of gangliosides generates structurally distinct aggregates; however, specific morphological features of these aggregates have yet to be determined. Yanagisawa *et al.* have demonstrated distinct fibril morphology resulting from GM1-bound A β . Following studies of A β -ganglioside interactions elucidated a specific pathway for the generation of toxic amyloid structures. Whether distinct toxic species are formed as a result of this interaction or ganglioside clusters trigger A β membrane disruption remains to be deciphered. The universal characteristics shared by a variety of amyloids and the great body of biophysical evidence in the literature suggests that such data could be useful for understanding the underlying toxicity of A β and other amyloids. While there is much work to be done in this regard, a directed focus should be geared towards obtaining high-resolution structural information on membrane-bound A β species to advance the understanding of AD pathogenesis, with the ultimate intention of improved and novel therapeutics to alleviate and/or reverse symptoms.

1 References

- 1 Alzheimer's Association: Facts & Figures, *Alzheimer's & Dementia*, 2013, **9**.
- 2 J. Hardy and D. J. Selkoe, *Science*, 2002, **297**, 353–356.
- 3 I. Benilova, E. Karran and B. De Strooper, *Nat. Neurosci.*, 2012, **15**, 349–357.
- 4 R. J. Bateman, C. Xiong, T. L. Benzinger, A. M. Fagan, A. Goate, N. C. Fox, D. S. Marcus, N. J. Cairns, X. Xie, T. M. Blazey, D. M. Holtzman, A. Santacruz, V. Buckles, A. Oliver, K. Moulder, P. S. Aisen, B. Ghetti, W. E. Klunk, E. McDade, R. N. Martins, C. L. Masters, R. Mayeux, J. M. Ringman, M. N. Rossor, P. R. Schofield, R. A. Sperling, S. Salloway and J. C. Morris, *N. Engl. J. Med.*, 2012, **367**, 795–804.
- 5 S. M. Butterfield and H. A. Lashuel, *Angew. Chem., Int. Ed.*, 2010, **49**, 5628–5654.
- 6 R. Tycko, *Q. Rev. Biophys.*, 2006, **39**, 1–55.
- 7 C. A. Ross and M. A. Poirier, *Nat. Med.*, 2004, **10**(Suppl), S10–S17.
- 8 D. B. Teplow, *Alzheimer's Res. Ther.*, 2013, **5**, 39.
- 9 S. Chimon, M. A. Shaibat, C. R. Jones, D. C. Calero, B. Aizezi and Y. Ishii, *Nat. Struct. Mol. Biol.*, 2007, **14**, 1157–1164.
- 10 M. Ahmed, J. Davis, D. Aucoin, T. Sato, S. Ahuja, S. Aimoto, J. I. Elliott, W. E. Van Nostrand and S. O. Smith, *Nat. Struct. Mol. Biol.*, 2010, **17**, 561–567.
- 11 M. P. Lambert, A. K. Barlow, B. A. Chromy, C. Edwards, R. Freed, M. Liosatos, T. E. Morgan, I. Rozovsky, B. Trommer, K. L. Viola, P. Wals, C. Zhang, C. E. Finch, G. A. Krafft and W. L. Klein, *Proc. Natl. Acad. Sci. U. S. A.*, 1998, **95**, 6448–6453.
- 12 J. R. Brender, S. Salamekh and A. Ramamoorthy, *Acc. Chem. Res.*, 2011.
- 13 R. Kaye, E. Head, J. L. Thompson, T. M. McIntire, S. C. Milton, C. W. Cotman and C. G. Glabe, *Science*, 2003, **300**, 486–489.
- 14 D. M. Walsh, I. Klyubin, J. V. Fadeeva, W. K. Cullen, R. Anwyl, M. S. Wolfe, M. J. Rowan and D. J. Selkoe, *Nature*, 2002, **416**, 535–539.
- 15 K. Ono, M. M. Condron and D. B. Teplow, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, **106**, 14745–14750.
- 16 R. Kaye, A. Pensalfini, L. Margol, Y. Sokolov, F. Sarsoza, E. Head, J. Hall and C. Glabe, *J. Biol. Chem.*, 2009, **284**, 4230–4237.
- 17 H. A. Lashuel, D. Hartley, B. M. Petre, T. Walz and P. T. Lansbury Jr., *Nature*, 2002, **418**, 291.
- 18 S. Vivekanandan, J. R. Brender, S. Y. Lee and A. Ramamoorthy, *Biochem. Biophys. Res. Commun.*, 2011, **411**, 312–316.
- 19 A. R. Ladiwala, J. Litt, R. S. Kane, D. S. Aucoin, S. O. Smith, S. Ranjan, J. Davis, W. E. Van Nostrand and P. M. Tessier, *J. Biol. Chem.*, 2012, **287**, 24765–24773.
- 20 D. V. Laurents, P. M. Gorman, M. Guo, M. Rico, A. Chakrabarty and M. Bruix, *J. Biol. Chem.*, 2005, **280**, 3675–3685.
- 21 L. Yu, R. Edalji, J. E. Harlan, T. F. Holzman, A. P. Lopez, B. Labkovsky, H. Hillen, S. Barghorn, U. Ebert, P. L. Richardson, L. Miesbauer, L. Solomon, D. Bartley, K. Walter, R. W. Johnson, P. J. Hajduk and E. T. Olejniczak, *Biochemistry*, 2009, **48**, 1870–1877.
- 22 E. Cerf, R. Sarroukh, S. Tamamizu-Kato, L. Breydo, S. Derclaye, Y. F. Dufrene, V. Narayanaswami, E. Goormaghtigh, J. M. Ruyschaert and V. Raussens, *Biochem. J.*, 2009, **421**, 415–423.
- 23 R. Sarroukh, E. Cerf, S. Derclaye, Y. F. Dufrene, E. Goormaghtigh, J. M. Ruyschaert and V. Raussens, *Cell. Mol. Life Sci.*, 2011, **68**, 1429–1438.
- 24 A. Jan, O. Adolfsson, I. Allaman, A. L. Buccarello, P. J. Magistretti, A. Pfeifer, A. Muhs and H. A. Lashuel, *J. Biol. Chem.*, 2011, **286**, 8585–8596.
- 25 S. D. Cady, K. Schmidt-Rohr, J. Wang, C. S. Soto, W. F. Degrado and M. Hong, *Nature*, 2010, **463**, 689–692.
- 26 A. Ramamoorthy and J. Xu, *J. Phys. Chem. B*, 2013, **117**, 6693–6700.
- 27 H. Jang, L. Connelly, F. T. Arce, S. Ramachandran, R. Lal, B. L. Kagan and R. Nussinov, *Phys. Chem. Chem. Phys.*, 2013, **15**, 8868–8877.
- 28 J. McLaurin and A. Chakrabarty, *Eur. J. Biochem.*, 1997, **245**, 355–363.
- 29 A. Kakio, S. Nishimoto, K. Yanagisawa, Y. Kozutsumi and K. Matsuzaki, *Biochemistry*, 2002, **41**, 7385–7390.
- 30 K. Yanagisawa, A. Odaka, N. Suzuki and Y. Ihara, *Nat. Med.*, 1995, **1**, 1062–1066.
- 31 K. Matsuzaki, K. Kato and K. Yanagisawa, *Biochim. Biophys. Acta*, 2010, **1801**, 868–877.
- 32 J. McLaurin, T. Franklin, P. E. Fraser and A. Chakrabarty, *J. Biol. Chem.*, 1998, **273**, 4506–4515.
- 33 K. Ikeda, T. Yamaguchi, S. Fukunaga, M. Hoshino and K. Matsuzaki, *Biochemistry*, 2011, **50**, 6433–6440.
- 34 M. Wakabayashi and K. Matsuzaki, *J. Mol. Biol.*, 2007, **371**, 924–933.
- 35 M. Yagi-Utsumi, T. Kameda, Y. Yamaguchi and K. Kato, *FEBS Lett.*, 2010, **584**, 831–836.
- 36 M. P. Williamson, Y. Suzuki, N. T. Bourne and T. Asakura, *Biochem. J.*, 2006, **397**, 483–490.
- 37 M. Utsumi, Y. Yamaguchi, H. Sasakawa, N. Yamamoto, K. Yanagisawa and K. Kato, *Glycoconjugate J.*, 2009, **26**, 999–1006.
- 38 N. Arispe, E. Rojas and H. B. Pollard, *Proc. Natl. Acad. Sci. U. S. A.*, 1993, **90**, 567–571.
- 39 A. Quist, I. Doudevski, H. Lin, R. Azimova, D. Ng, B. Frangione, B. Kagan, J. Ghiso and R. Lal, *Proc. Natl. Acad. Sci. U. S. A.*, 2005, **102**, 10427–10432.
- 40 S. K. Rhee, A. P. Quist and R. Lal, *J. Biol. Chem.*, 1998, **273**, 13379–13382.
- 41 H. Jang, J. Zheng, R. Lal and R. Nussinov, *Trends Biochem. Sci.*, 2008, **33**, 91–100.
- 42 R. Capone, H. Jang, S. A. Kotler, B. L. Kagan, R. Nussinov and R. Lal, *Biochemistry*, 2012, **51**, 776–785.
- 43 R. Capone, H. Jang, S. A. Kotler, L. Connelly, F. Teran Arce, S. Ramachandran, B. L. Kagan, R. Nussinov and R. Lal, *J. Chem. Theory Comput.*, 2012, **8**, 1143–1152.
- 44 P. Prangko, E. C. Yusko, D. Sept, J. Yang and M. Mayer, *PLoS One*, 2012, **7**, e47261.

- 1 45 I. Solomonov, E. Korkotian, B. Born, Y. Feldman, A. Bitler,
F. Rahimi, H. Li, G. Bitan and I. Sagi, *J. Biol. Chem.*, 2012,
287, 20555–20564.
- 5 46 M. Hoshi, M. Sato, S. Matsumoto, A. Noguchi, K. Yasutake,
N. Yoshida and K. Sato, *Proc. Natl. Acad. Sci. U. S. A.*, 2003,
100, 6370–6375.
- 47 M. F. Sciacca, S. A. Kotler, J. R. Brender, J. Chen,
D. K. Lee and A. Ramamoorthy, *Biophys. J.*, 2012, 103,
702–710.
- 10 48 J. R. Brender, E. L. Lee, K. Hartman, P. T. Wong,
A. Ramamoorthy, D. G. Steel and A. Gafni, *Biophys. J.*,
2011, 100, 685–692.
- 49 M. F. Sciacca, J. R. Brender, D. K. Lee and A. Ramamoorthy,
Biochemistry, 2012, 51, 7676–7684.
- 15
- 20
- 25
- 30
- 35
- 40
- 45
- 50
- 55
- 50 E. Sparr, M. F. Engel, D. V. Sakharov, M. Sprong, J. Jacobs,
B. de Kruijff, J. W. Hoppener and J. A. Killian, *FEBS Lett.*,
2004, 577, 117–120.
- 51 A. S. DeToma, S. Salamekh, A. Ramamoorthy and M. H. Lim,
Chem. Soc. Rev., 2012, 41, 608–621.
- 52 M. Necula, R. Kaye, S. Milton and C. G. Glabe, *J. Biol.*
Chem., 2007, 282, 10311–10324.
- 53 M. F. Engel, C. C. vandenAkker, M. Schleegeer, K. P. Velikov,
G. H. Koenderink and M. Bonn, *J. Am. Chem. Soc.*, 2012, 134,
14781–14788.
- 10 54 J. C. Diaz, O. Simakova, K. A. Jacobson, N. Arispe and
H. B. Pollard, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, 106, 3348–3353.
- 55 J. Jarvet, J. Danielsson, P. Damberg, M. Oleszczuk and
A. Graslund, *J. Biomol. NMR*, 2007, 39, 63–72.
- 15
- 20
- 25
- 30
- 35
- 40
- 45
- 50
- 55