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Subject: Pressure drops in the helium supply lines, for the ACOL cryogenic and cryopumping system.

#### 1. SCOPE

The ACOL cryogenic and cryopumping system is equipped with a helium high pressure supply and return line distribution system (see Fig 1). It consists of one high pressure-line (HP) and one low pressure-line (LP), each with a length of approximately 110 m.

This document presents the calculation for the pressure drops in the system. It also proposes the dimensions of the tubes.

### 2. SPECIFICATION

1. Pressure in HP-line:	20 bar
2. Pressure in LP-line:	5 bar
3. Length of HP-line:	110 m
4. Length of LP-line:	110 m
5. Volume ratio VLP/VHP:	1/4
6. Flow of helium:	400 norm m³/h
7. Temperature:	20-30°C
8. Max allowable pressure	drop per line: <0.5 bar

### 3. CALCULATIONS

The calculated pressure drop as function of the tube internal diameter, at  $t=20^{\circ}C$  and at  $t=30^{\circ}C$ , is shown in Table 1.

iameter	Pressure drop	Pressure drop
n)	∆p(bar),at t=20°C	∆p(bar),at t=30°C
20	0.68	0.70
25	0.23	0.24
30	0.10	0.11
40	0.100	0.104
50	0.034	0.036
60	0.014	0.015
	n) 20 25 30 40 50	Δp(bar), at t=20°C   20 0.68   25 0.23   30 0.10   40 0.100   50 0.034

Table 1:	The	pressure	drop	in	HP-	and	LP-line
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# Conclusion:

In order to satisfy the max allowable pressure drop of less than 0.5 bar, a tube of 25 mm is needed to be used for the HP-line. This gives with the volume ratio 1/4, the diameter of the LP-line; 50 mm.

Remarks:

The influence of one-time losses (valves, turnes, etc) has not been included in this calcultaion.

A rough estimate of these losses indicates that they are very small. Example: A 90° bend gives a drop of 0.0005 bar.

### 4. THEORY USED FOR THE CALCULATION

The pressure drop:

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	$\Delta p = p_1 - p_2 = p_1 (1 - \sqrt{1 - \lambda v_1^2 \rho_1 l / (\rho_1 d)})$
The universal gas law gives	:
	$\rho = \rho/(R \cdot T)$
	R = Rm/M
Continuity equations:	
	$\dot{\mathbf{m}} = \rho \cdot \mathbf{v} \cdot \mathbf{A} = \text{constant}$
	$\dot{\mathbf{v}} = \mathbf{v} \cdot \mathbf{A}$
Flow equations:	
Reynolds number	Re = $v \cdot d/v$
kinematic viscosity	$v = \eta/\rho$
if Re>2320	$\lambda = 0.3164/4\sqrt{\text{Re}}$

Chemical and physical constants for helium

M = 4.00 kg/kmolR = 2079.01 j/(kg K)  $\rho_0 = 0.179 \text{ kg/m}^3$ 

dynamic viscosity  $\eta$  at p=1 atm

t	η
(°C)	(µNs/m²)_
0	18.7
20	19.6
50	21.0

dynamic viscosity  $\eta$  at t=20°C

р	η
(bar)	(µNs/m²)
20	19.6
50	19.6
100	19.6

## REFERENCES

- [1] Dubbel, Taschenbuch fur den Maschinenbau, 15. Auflage.
- [2] G.W.C. Kaye, Tables of Physical and Chemical Constants.

