Experimental evidence of tetrahedral interstitial and bond-centered Er in Ge

S. Decoster, B. De Vries, and A. Vantomme

Instituut voor Kern- en Stralingsfysica and INPAC, KULeuven, 3001 Leuven, Belgium

U. Wahl and J.G. Correia

Instituto Tecnológico e Nuclear, Estrada Nacional 10, 2686-953 Sacavém, Portugal

and Centro de Física Nuclear da Universidade de Lisboa, Avenida Prof. Gama Pinto 2, 1649-003 Lisboa, Portugal

(Dated: August 28, 2008)

We report on an emission channeling study of the lattice site location of implanted Er in Ge together with its thermal stability. We found direct experimental evidence of Er atoms located on the tetrahedral interstitial site (T) and on the bond-centered (BC) site, with a maximum total occupancy after annealing at 400°C. Whereas Er is expected to occupy the T site in a diamond crystal structure, the observation of bond-centered Er in Ge is more surprising and believed to be related to the Er-vacancy defect in the split-vacancy complex configuration.

During the past decades, the optical properties of Er integrated in semiconductors have been of great interest for photonic applications, especially in Si since the wavelength of the ${}^{4}I_{13/2} \rightarrow {}^{4}I_{15/2}$ transition of the Er³⁺-ion corresponds to the minimum absorption of silica-based optical fibers. The properties and dependencies of the luminescence in Er-doped Si or Si-based materials have been investigated intensively. However, due to the socalled thermal and concentration quenching, its room temperature luminescence yield remains below needs.¹

In the search for alternative semiconducting host materials, the Ge:Er system received little attention so far because bulk Ge has a band gap (0.67 eV) which is smaller than the energy corresponding to the technologically interesting optical emission wavelength of Er, i.e. $1.54 \ \mu m$ (0.80 eV). This results in a large self absorption and consequently a very low efficiency. However, by using Ge nanoparticles, it has been possible to tune the band gap with the size of the nanoparticles, and thus to reduce the interband absorption.² The stronger quantum confining effect and a better controlled oxidation with respect to Si, makes it a promising candidate as an Er-host for future photonic applications.^{3,4}

The growth of these Er-doped Ge-nanoparticles and their luminescence properties have been studied in detail,^{2,5,6} as well as the annealing behavior of highly Er-doped Ge.⁷ Despite the increased interest in this system, a number of fundamental questions remain unsolved. A particular one is the lattice location of the Er atoms, which largely influences the luminescence properties in semiconductors.⁸ In Si, the lattice location of implanted Er has been investigated quite thoroughly. From emission channeling experiments^{9,10} and several first-principles calculations,^{11–13} the tetrahedral (T) interstitial site was found to be the preferred site in oxygenlean Si. However, other studies revealed Er-atoms on the substitutional (S) site and the hexagonal (H) site.^{14,15} In Ge, the only study on the lattice location of rare earths, is the work of Yamamoto et al., who concluded from He channeling experiments that 25% of implanted Tm occupies the T site. 16

In this letter, we present a direct lattice location study

of Er in Ge, measured with the emission channeling technique¹⁷. We make use of the fact that charged particles, emitted from an implanted radioactive isotope will be guided by the potential of atomic rows and planes while traveling through the crystal. The resulting anisotropic electron emission pattern around low-index crystal directions is characteristic for the lattice site occupied by the emitting atom and is measured with a 2-dimensional (2D) energy- and position-sensitive Si detector of 22×22 pixels. The advantages of this technique with respect to ion beam channeling techniques such as RBS/C, are a considerable improvement in accuracy due to the use of 2D-patterns instead of 1D-scans and the use of low implantation fluences which allows us to measure the lattice location of isolated atoms.

Implantations with radioactive 167 Tm (t_{1/2}=9.25 d) were performed at room temperature at the ISOLDE facility in CERN. This isotope decays into ^{167m}Er $(t_{1/2}=2.27 \text{ s})$, emitting K, L, and M conversion electrons of respectively 150, 199 and 206 keV. Three undoped Gesamples were studied with varying implantation fluence, energy and sample orientation, and will be labeled sample A $(3.6 \times 10^{12} \text{ cm}^{-2}; 30 \text{ keV}; (100))$, B $(1.1 \times 10^{13} \text{ cm}^{-2};$ 60 keV; (100) and C $(7.7 \times 10^{12} \text{ cm}^{-2}; 30 \text{ keV}; (111))$. To obtain accurate and unambiguous results, emission patterns around the four crystal directions ($\langle 100 \rangle$, $\langle 111 \rangle$, $\langle 211 \rangle$ and $\langle 110 \rangle$) were measured, analyzed consistently and fitted to a set of simulated patterns.¹⁸ These simulations, based on the dynamical theory of electron diffraction, were performed for several high-symmetry sites such as the S, T, H, bond-centered (BC) and the so-called AB, DS, SP, Y and C sites,¹⁹ as well as for discrete displacements between these sites along the $\langle 111 \rangle$ -, $\langle 100 \rangle$ - and $\langle 110 \rangle$ -direction. To monitor the thermal stability of the lattice location of the Er atoms, the measurements were performed after implantation as well as after annealing for 10 min in vacuum ($< 10^{-5}$ mbar) at temperatures up to 600° C.

Figures 1(a)-(d) show the normalized experimental electron emission patterns around the four investigated crystal directions for sample C after annealing at 400°C. Along the $\langle 111 \rangle$ - and $\langle 100 \rangle$ -direction, we clearly see an



FIG. 1: (Color online)(a)-(d) Two-dimensional conversion electron patterns emitted from ^{167m}Er in Ge around the $\langle 111 \rangle$, $\langle 100 \rangle$, $\langle 110 \rangle$ and $\langle 211 \rangle$ -axes, following a 400°C annealing in vacuum; (e)-(g) Calculated patterns for Er-atoms occupying the T site and (h)-(j) the BC site around the $\langle 100 \rangle$, $\langle 110 \rangle$ and $\langle 211 \rangle$ -axis respectively; (k)-(n) Best fits to the experimental patterns. The anisotropy is depicted with a color scale between red (low) and blue (high).

enhanced number of electrons emitted along the crystal axis, while the $\langle 110 \rangle$ and $\langle 211 \rangle$ patterns show a blocking effect, which is characteristic for the T site. However, when comparing the experimental patterns for the $\langle 100 \rangle$ -, $\langle 110 \rangle$ - and $\langle 211 \rangle$ -direction (Fig. 1(b)-(d)) to the corresponding calculated patterns for Er on a pure T site (Fig. 1(e)-(g)), a number of major discrepancies are obvious. This leads to the conclusion that the Er-atoms occupy at least a second high-symmetry site. Therefore, a detailed fitting procedure was applied, considering all possible combinations of 2 and even 3 different highsymmetry sites, including eventual displacements. While adding most of the high-symmetry sites (H, S, SP, DS,



FIG. 2: (Color online) The experimental $\langle 211 \rangle$ -pattern from Fig. 1(d) after subtracting the best fit with allowing a fraction on (a) the BC site (b) the T and the BC site and (c) the T and the C site.

AB, Y and C) resulted only in marginal improvements of the fit (χ^2 -improvement < 3%), including a BC fraction resulted in a consistent fit in all four directions, with an average χ^2 -improvement of more than 25%. Superposing the simulations for the T site (e)-(g) and the BC site (h)-(j), solves the discrepancies, as can be seen from Fig. 1. This is further clarified in Fig. 2, showing the residual experimental pattern from Fig. 1 (d) after subtracting the fit with a BC fraction only (Fig. 2 (a)), with a T and a BC fraction (Fig. 2 (b)) and with a fraction on the T site and on the C site (Fig. 2 (c)). Fig. 2 (a) clearly shows the features of T site pattern, while Fig. 2 (c) has the same features as a BC pattern. On the other hand, Fig. 2 (b) results in a featureless pattern, clearly indicating that the only satisfying fit is obtained by including a fraction on the BC and the T site. Adding a third fraction did not significantly improve the fit, indicating that only relatively small fractions (< 2%) of Er-atoms occupy other high-symmetry sites. Similar fit results have been found for the other two samples, resulting in average χ^2 improvements between 15% and 40%.

Figures 1(k)-(n) show the best fit to the experimental patterns. Averaging the results of the 4 measured directions, 16(2)% of the Er-atoms are found on the BC site and 20(4)% on a slightly displaced T site. The Eratoms on the T site were found to have an average root mean square displacement of about 0.18(3) Å, which is somewhat larger than the Ge host vibration amplitude at room temperature of 0.08 Å, without any preferential direction. The large remaining fraction of Er-atoms (64%) is referred to as the random fraction. This random fraction partly consists of Er-atoms that are distributed randomly within the crystal lattice, but a large contribution will be related to the implantation-induced lattice damage. Even for the low fluences used in this study, heavy ion implantation produces highly damaged regions, especially in materials with small lattice binding energies such as Ge. Firstly, due to the deterioration of the crystal structure, conversion electrons from radioactive isotopes located in damaged regions will be emitted more isotropically. Secondly, a fraction of the electrons emitted from an undamaged region will pass through damaged crystal regions, enhancing the probability for dechanneling. Both effects will result in an isotropic background and consequently in the high random fraction observed in this work. It is important to



FIG. 3: Annealing temperature dependence of the fraction Er-atoms on the T site (circles) and the BC site (squares) for sample A (filled), B (open) and C (half open symbols). The dotted lines are a guide to the eye.

notice that this high random fraction has no influence whatsoever on the anisotropy of the patterns and will not affect the qualitative analysis of the spectra.

In Fig. 3, the temperature dependence of the Erfractions on the T and BC site is plotted for all 3 samples. Despite the slightly different implantation conditions, the same behavior is present in all three samples. Annealing up to 400°C results in an increasing fraction of Er-atoms on high-symmetry sites, which is a direct consequence of the partial lattice recovery of damaged regions. This fraction decreases again after annealing at temperatures higher than 450°C, which is most likely related to diffusion of the Er-atoms. As the depth profile of the implanted Er-atoms is an input parameter for the simulations, it implies that a larger mean projected range of the Er-atoms after diffusion will lead to an underestimation of the measured and fitted fraction on highsymmetry sites. Alternatively, the trapping of Er-atoms by migrating defects, may result in Er-atoms occupying low symmetry sites in defect complexes.

The existence of tetrahedral interstitial Er in Si has been proven both from experiments and calculations. Due to the similar lattice structure of Si and Ge, Er can be expected to occupy the T site in Ge as well, which was found experimentally in this study. So far, there is no clear explanation for the small displacement of the Er-atoms on the T site. However, it could be originating from lattice relaxations as was predicted from density functional computations.¹⁴ Applying stress related arguments, Er is not expected to occupy a BC site in an undamaged Ge crystal. However, we believe this site is related to an Er-vacancy (Er-V) complex. From ab-initio density functional calculations, a range of high-Z impurities (from Cd to Bi) were found to be unstable on the S site with a vacancy on the nearest neighbor S site.²⁰ The impurity prefers to shift towards the BC site, with the vacancy *split* between the two sites. Ion implantation is known to produce many vacancies and interstitials, which are mobile at room temperature in Ge.²¹ This implies that during the implantation, they are free to migrate through the crystal until they either recombine, form complex clusters or get trapped by impurity atoms. Although no calculations have been performed for Er, interpolating the trend that high-Z impurities are stable on the BC site, allows us to conclude that our experimental observation of Er on the BC site in Ge might be indirect evidence of the Er-V complex in the splitvacancy configuration. It can not be excluded that other (more complex) defect configurations may also contribute to the observed bond-centered behavior.

In conclusion, we have found experimental evidence of implanted Er atoms located on the T site and on the BC site in Ge. While the occupation of the T site could be expected from comparison with similar studies in Si, this study offers direct evidence of the occupation of the BC site in Ge. This site has been predicted by ab-initio density functional calculations for heavy impurities in Ge and we believe it is related to the Er-V complex in the split-vacancy configuration. Since the luminescent properties of dopants are significantly influenced by their lattice site, the observation of Er on two different sites can be of great importance in understanding – and even tuning – the optical properties of the Ge:Er system.

This work was supported by the Fund for Scientific Research, Flanders (FWO), the Concerted Action of the K.U.Leuven (GOA/2004/02), the Inter-University Attraction Pole (IUAP P6/42), the Center of Excellence Programme (INPAC EF/05/2005), the Portuguese Foundation for Science and Technology (POCI-FP-81921-2007) and the ISOLDE collaboration. S. D. acknowledges financial support from FWO, Flanders.

- ¹ A. Kenyon, Semicond. Sci. Technol. **20**, R65 (2005).
- ² S. Sato, S. Nozaki, H. Morisaki, and M. Iwase, Appl. Phys. Lett. **66**, 3176 (1995).
- ³ C. Bostedt, T. van Buuren, T. Willey, N. Franco, L. Terminello, C. Heske, and T. Moller, Appl. Phys. Lett. 84, 4056 (2004).
- ⁴ M. Jing, M. Ni, W. Song, J. Lu, Z. Gao, L. Lai, W. Mei, D. Yu, H. Ye, and L. Wang, J. Chem. Phys. B **110**, 18332 (2006).
- ⁵ C. Heng, T. Finstad, P. Storas, Y. Li, A. Gunnaes, and

O. Nilsen, Appl. Phys. Lett. 85, 4475 (2004).

- ⁶ J. Chen, D. Pang, H. Cheong, P. Wickboldt, and W. Paul, Appl. Phys. Lett. **67**, 2182 (1995).
- ⁷ S. Kucheyev, J. Bradby, S. Ruffell, C. Li, T. Felter, and A. Hamza, Appl. Phys. Lett. **90**, 221901 (2007).
- ⁸ Y. Tang, K. Heasman, W. Gillin, and B. Sealy, Appl. Phys. Lett. 55, 432 (1989).
- ⁹ U. Wahl, A. Vantomme, J. De Wachter, R. Moons, G. Langouche, J. Marques, J. Correia, and the ISOLDE Collaboration, Phys. Rev. Lett. **79**, 2069 (1997).

- ¹⁰ U. Wahl, J. Correia, J. Araújo, A. Vantomme, G. Langouche, and the ISOLDE Collaboration, Physica B 273-**274**, 342 (1999).
- ¹¹ M. Needels, M. Schluter, and M. Lannoo, Phys. Rev. B **47**, 15533 (1993). ¹² Y. Wan, Y. Ling, Q. Sun, and X. Wang, Phys. Rev. B **58**,
- 10415 (1998).
- ¹³ D. Prezzi, T. Eberlein, R. Jones, J. Filhol, J. Coutinho, M. Shaw, and P. Briddon, Phys. Rev. B 71, 245203 (2005).
- $^{14}\,$ A. Raffa and P. Ballone, Phys. Rev. B ${\bf 65},\,121309$ (2002). ¹⁵ A. Kozanecki, R. Wilson, B. Sealy, J. Kaczanowski, and
- L. Nowicki, Appl. Phys. Lett. 67, 1847 (1995).
- ¹⁶ Y. Yamamoto, M. Wakaiki, A. Ikeda, and Y. Kido, Nucl.

Inst. Meth. Phys. Res. B 153, 452 (1999).

- ¹⁷ U. Wahl, J. Correia, A. Czermak, S. Jahn, P. Jalocha, J. Marques, A. Rudge, F. Schopper, J. Soares, A. Vantomme, et al., Nucl. Inst. Meth. Phys. Res. A 524, 245 (2004).
- ¹⁸ H. Hofsäss and G. Lindner, Phys. Rep. **201**, 121 (1991).
- ¹⁹ U. Wahl, Phys. Rep. **280**, 145 (1997).
- ²⁰ H. Höhler, N. Atodiresei, K. Schroeder, R. Zeller, and P. Dederichs, Phys. Rev. B 71, 035212 (2005).
- 21 H. Hässlein, R. Sielemann, and C. Zistl, Phys. Rev. Lett. **80**, 2626 (1998).